

Dissolution and Deposition of Dental Metallic Biomaterials-A Theoretical Model in Cyclic Voltammetry

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Abstract: An on-line MATHCAD protocol is presented, suitable to calculate cyclic voltammograms related to dissolution and deposition of dental metallic biomaterials. The provided protocol provides insights on how the temperature, applied scan rate and the kinetics of electron transfer related to dissolution/deposition process affect the cyclic voltammograms of the system considered. This part of the Model displays plethora of cyclic voltammograms calculated at various kinetics of electron transfer steps of dissolution/deposition process of metallic dental biomaterials.

$E_s = -0.4$ $E_f = 0.4$ $\Delta E = E_f - E_s$ $dE = 0.004$ $\tau = 0.01$ $d = \frac{\tau}{25}$

$n = \frac{\tau}{d} + 1 = \frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d}$ $n = \frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} + 1 = \left(\frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} \right)$

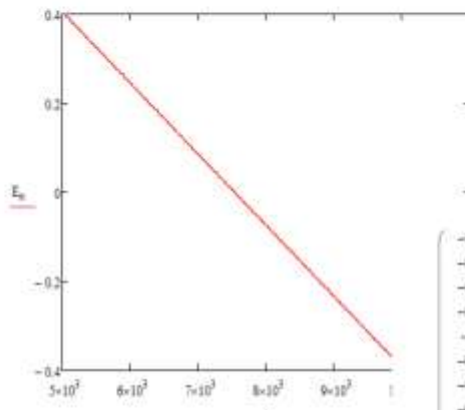
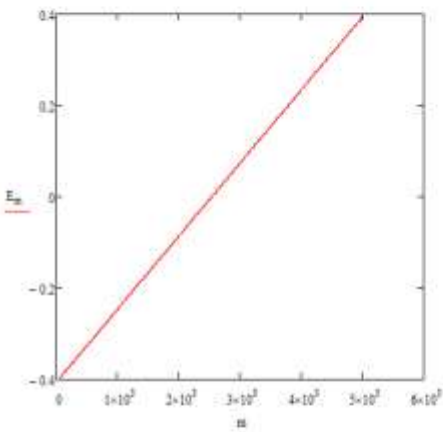
$E_{ca} = E_s + \left[\text{erf} \left(\frac{n - \frac{\tau}{d}}{25} \right) \cdot dE - dE \right]$

$E_{an} = E_f - \left[\text{erf} \left(\frac{n - \left(\frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} \right)}{25} \right) \cdot dE - dE \right]$

$\frac{\Delta E}{dE} = 200$

$\frac{dE}{\tau} = 0.4$

ECrev Mechanism of Metallic Dissolution/Deposition
 Red(ads) \cdot e \cdot = Ox(dissolved)
 in Cyclic Voltammetry
 F: Equilibrium Constant $M > 100$, mechanism converges to simple Red(ads) \cdot e \cdot = Ox(dissolved)
 Mechanism
 K is dimensionless kinetic parameter related to electrode reaction
 γ is dimensionless catalytic parameter related to follow up chemical reaction
 α is electron transfer coefficient
 M is equilibrium constant of follow up chemical reaction
 kf and kb are forward and backward rate constants of follow up chemical reaction
 Es is starting potential
 Ef is final potential
 dE is potential step
 Ψ is symbol for dimensionless current
 Em is cathodic potential ramp in cyclic voltammetry
 En is anodic potential ramp
 Sk is integration factor
 τ is duration of potential steps
 D is diffusion coefficient of Ox



$\frac{\Delta E}{dE} = 0.28$ $\text{erf} = 0.00000008$ $w = 1.000001000002$

$k_1 = 0.2$ $D = 3 \cdot 10^{-6}$ $k_2 = 0.05$ ϵ

$\frac{k_2}{k_1} = \frac{k_2 \cdot \sqrt{\tau}}{\sqrt{D}}$ $\alpha = 0.5$ $k_3 = 0.0002222210$ 0.075

$M = \frac{k_f}{k_b}$ Konstanta na ravnoteza 0.35

$\log(K) = 1.042$ $K = 10^{1.042}$ $\frac{M}{k_1} = 2222.1000$

$E_{ca} = k_f + k_b$ kinetički parametar $k = 1.2 \cdot \left(\frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} \right)$

$r = \epsilon \cdot \tau$

$S_{1,k} = \sqrt{\frac{k}{25}} = \sqrt{\frac{k-1}{25}}$ $z = (k_f + k_b)^{0.5} \cdot \tau$

$\phi_{ac} = d \cdot \frac{F}{R \cdot T} \cdot E_s$ $\gamma = h \cdot l$ $z = 2.108 \cdot 10^{-3}$

$\gamma = 2.108 \cdot 10^{-3}$

$S_k = \left(1 - \text{erfc} \left(\frac{\tau}{25} \cdot k \right) \right) - \left[1 - \text{erfc} \left(\frac{\tau}{25} \right) \right]$

$E_s = 90500$ $d = 2$ $R_s = 8.311$ $T_s = 298.15$

$\phi_{ca} = d \cdot \frac{F}{R \cdot T} \cdot (E_{ca})$ $\phi_{an} = d \cdot \frac{F}{R \cdot T} \cdot (E_{an})$

$$\Phi_{ac} := eI \cdot \frac{F}{R \cdot T} \cdot Es$$

$$\gamma := h \cdot l$$

$$\gamma = 2.108 \times 10^{-3}$$

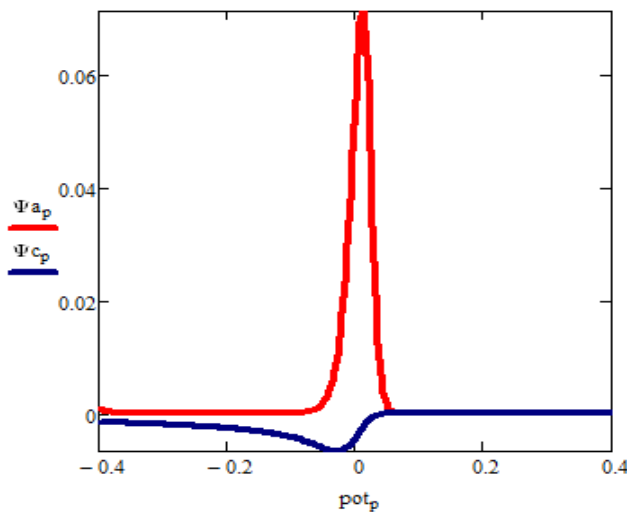
$$\Psi_1 := \frac{K \cdot e^{\alpha \cdot \Phi_1}}{1 + \frac{0.04 \cdot K \cdot e^{(1-\alpha) \cdot \Phi_1} \cdot 1}{\sqrt{1 \cdot 1}} + \frac{1 \cdot K \cdot e^{-(1-\alpha) \cdot \Phi_1}}{\sqrt{\pi \cdot 1}} \cdot \frac{M}{1+M}} + \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot \Phi_1} \cdot S_1$$

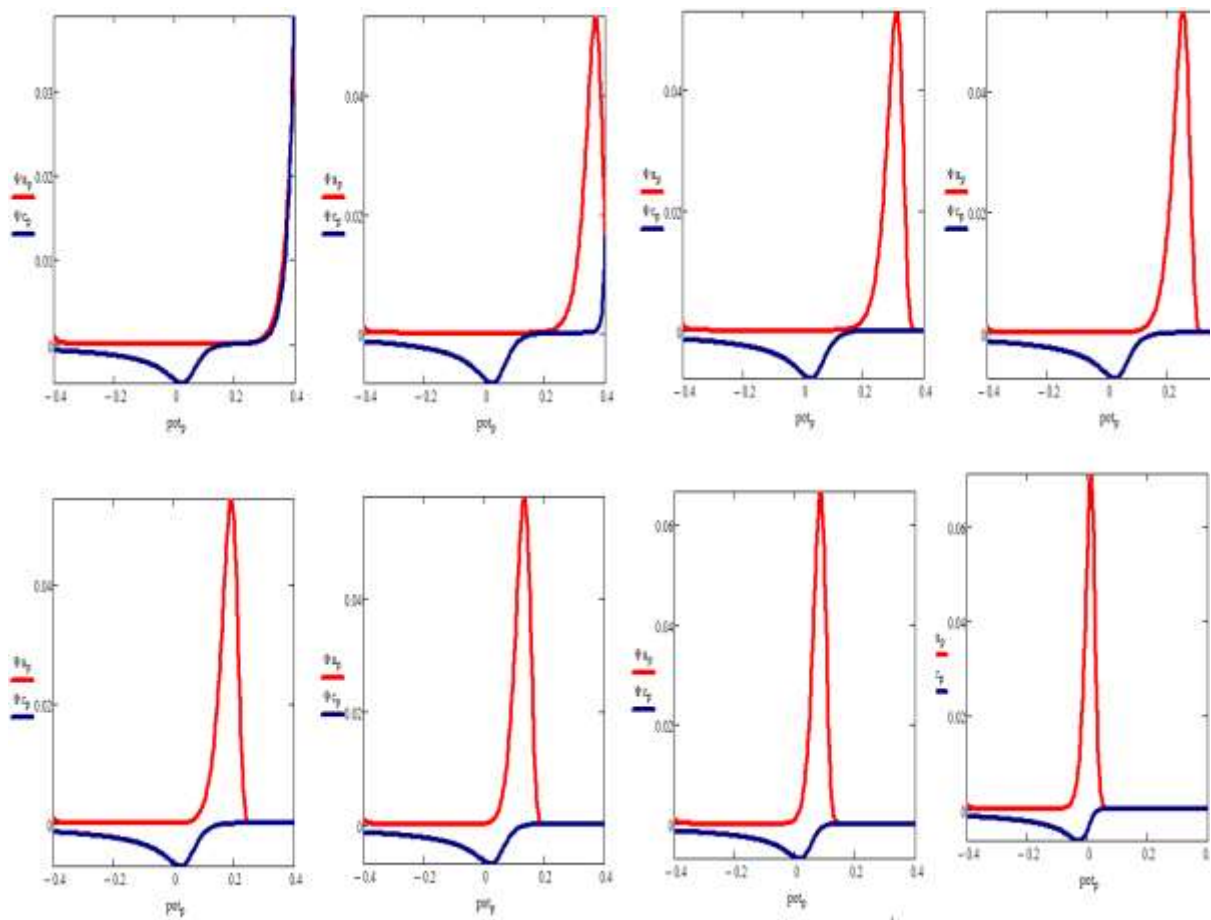
$$\Psi_s := \frac{K \cdot e^{\alpha \cdot \Phi_{ac}} - \frac{0.04 \cdot K \cdot e^{\alpha \cdot \Phi_{ac}}}{\sqrt{1 \cdot 1}} \cdot \sum_{j=1}^{s-1} (\Psi_j \cdot 1) - \frac{2 \cdot K \cdot e^{-(1-\alpha) \cdot \Phi_{ac}}}{\sqrt{\pi \cdot 25}} \cdot \frac{M}{1+M} \cdot \sum_{j=1}^{s-1} (\Psi_j \cdot S_{1_{s-j+1}}) - \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot \Phi_{ac}} \cdot \sum_{j=1}^{s-1} (\Psi_j \cdot S_{s-j+1})}{1 + \frac{0.04 \cdot K \cdot e^{\alpha \cdot \Phi_{ac}} \cdot 1}{\sqrt{1 \cdot 1}} + \frac{2 \cdot K \cdot e^{-(1-\alpha) \cdot \Phi_{ac}}}{\sqrt{\pi \cdot 25}} \cdot \frac{M}{1+M} + \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot \Phi_{ac}} \cdot S_1}$$

$$\Psi_m := \frac{w \cdot e^{\alpha \cdot \Phi_m} - \frac{0.04 \cdot w \cdot e^{\alpha \cdot \Phi_m}}{\sqrt{1 \cdot 1}} \cdot \sum_{j=1}^{m-1} (\Psi_j \cdot 1) - \frac{2 \cdot K \cdot e^{-(1-\alpha) \cdot \Phi_m}}{\sqrt{\pi \cdot 25}} \cdot \frac{M}{1+M} \cdot \sum_{j=1}^{m-1} (\Psi_j \cdot S_{1_{m-j+1}}) - \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot \Phi_m} \cdot \sum_{j=1}^{m-1} (\Psi_j \cdot S_{m-j+1})}{1 + \frac{0.04 \cdot w \cdot e^{\alpha \cdot \Phi_m} \cdot 1}{\sqrt{1 \cdot 1}} + \frac{2 \cdot K \cdot e^{-(1-\alpha) \cdot \Phi_m}}{\sqrt{\pi \cdot 25}} \cdot \frac{M}{1+M} + \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot \Phi_m} \cdot S_1}$$

$$\Psi_n := \frac{w \cdot e^{\alpha \cdot b_n} - \frac{0.04 \cdot w \cdot e^{\alpha \cdot b_n}}{\sqrt{1 \cdot 1}} \cdot \sum_{j=1}^{n-1} (\Psi_j \cdot 1) - \frac{2 \cdot K \cdot e^{-(1-\alpha) \cdot b_n}}{\sqrt{\pi \cdot 25}} \cdot \frac{M}{1+M} \cdot \sum_{j=1}^{n-1} (\Psi_j \cdot S_{1_{n-j+1}}) - \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot b_n} \cdot \sum_{j=1}^{n-1} (\Psi_j \cdot S_{n-j+1})}{1 + \frac{0.04 \cdot w \cdot e^{\alpha \cdot b_n} \cdot 1}{\sqrt{1 \cdot 1}} + \frac{2 \cdot K \cdot e^{-(1-\alpha) \cdot b_n}}{\sqrt{\pi \cdot 25}} \cdot \frac{M}{1+M} + \frac{\gamma}{1+M} \cdot e^{-(1-\alpha) \cdot b_n} \cdot S_1}$$

$$p := 1 \cdot \frac{\Delta E}{dE} \quad \Psi_{a_p} := (\Psi) \left(\frac{\tau}{d \cdot 25} + p \right) \cdot 25 \quad \Psi_{c_p} := (\Psi) \left[\left[\frac{\Delta E}{dE} \cdot 2 + \left(\frac{\tau}{25 \cdot d} \right) \right] \cdot p \right] \cdot 25 \quad \text{pot}_p := Es + p \cdot dE$$





Effect of kinetics of the electron transfer step of dissolution/deposition process of dental metallic biomaterial to the features of calculated cyclic voltammograms at room temperature

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