Voltametric analysis of the antioxidative potential of medicinal plants traditionally used in North Macedonia

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## Flora of North Macedonia

- Medicinal plants in North Macedonia
- Flora in North Macedonia above 3200 plant species
- Endemic species Pinus peuce
- Around 200-300 used in traditional and folk medicine





#### Popular plants used in traditional and folk medicine in North Macedonia



Mentha piperita L. (Mint tea)



Origanum vulgare L. (Mountain tea)



Melissa officinalis L. Lemon balm



Chamomilla recutita L. (chamomile)



Hypericum perforatum L. (St. John's wort)



Thymus serpyllum L. Wild thyme



Primula veris L. (cowslip)



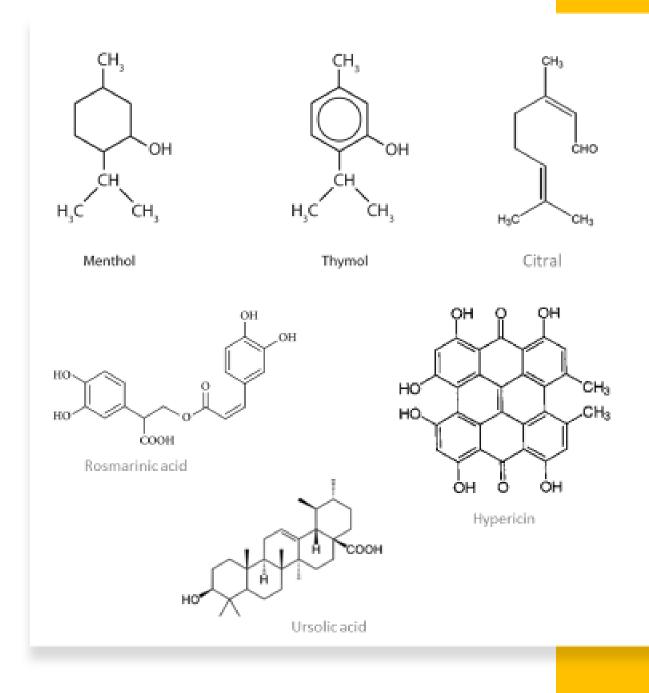
Gentiana lutea L. (cowslip)



Orchis morio L. (orchid)

## Neutralization of free radicals

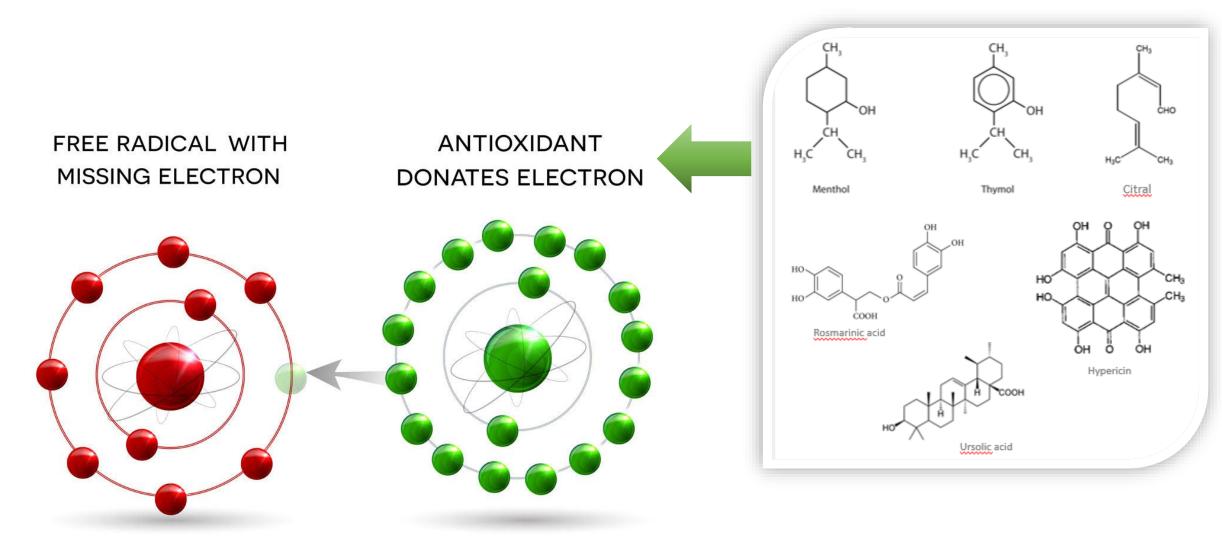
Antioxidant effect as basic property for further pharmacological activities.



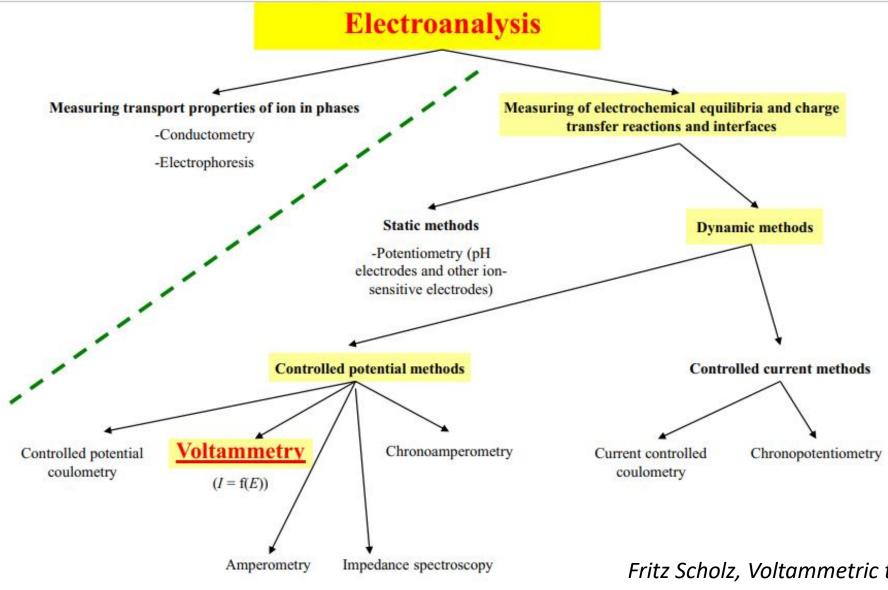
## What's the benefit from exploring plant extracts and their antioxidative potential?

"Bright side" of using plant extracts as antioxidant	"Challenges" of using plant extracts as antioxidants
Natural origin	Bioactive molecules found in very small concentration
Many new still unexplored bioactive compounds	Pharmacokinetics (ADME effects)
Synergistic effects	Different cell factors influencing their antioxidative ability (pH value in cell, physiological pathways activated by free radicals)

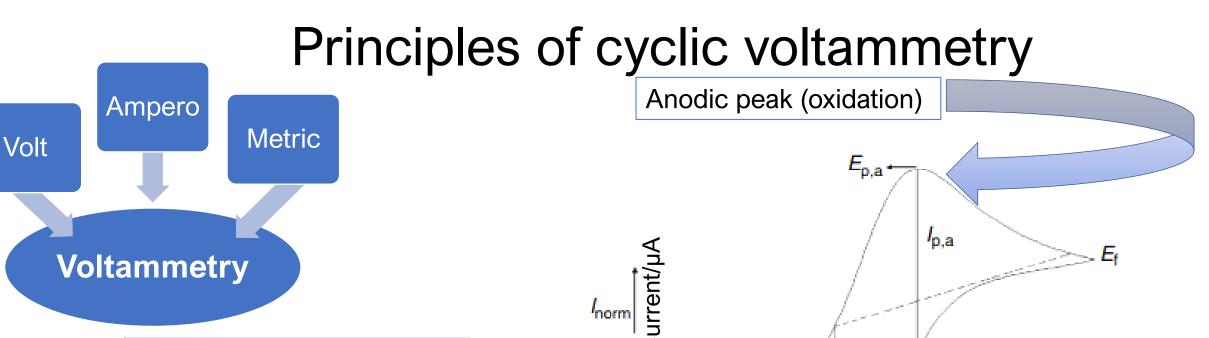
## Mechanism of antioxidative effect



## Wide overview of electrochemical techniques



*Fritz Scholz, Voltammetric techniques of analysis: the essentials. ChemTexts (2015) 1:17* 



E

- Measuring the current as a function of voltage (electrode potential)
- rate of oxidation/ reduction reaction
- anodic/cathodic peak potential

#### Voltammogram

0.04 0.08 0.12 0.16

Cathodic peak (reduction)

Ip.c

-0.16 - 0.12 - 0.08 - 0.04 0

 $E_{\rm p.c}$ 

E/V

Measuring the potential of oxidation/reduction of a substances in a solution

Electrochemistry as a tool for measuring the oxidation/reduction rate Cyclic voltammetry (electrocatalytic reactions)

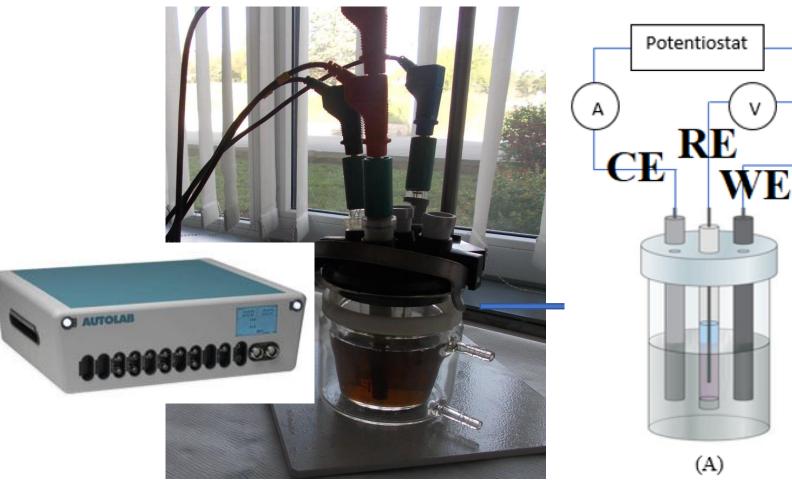
The system is composed of:

- >electrochemical cell
- 3 electrodes are immersed in electrolyte solution:
- Working electrode,
- Reference electrode and
- Helping electrode

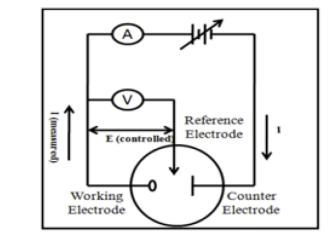




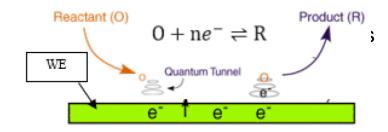
### Electrochemistry as a tool for measuring the oxidation/reduction rate Cyclic voltammetry (electroncatlitic reactions)



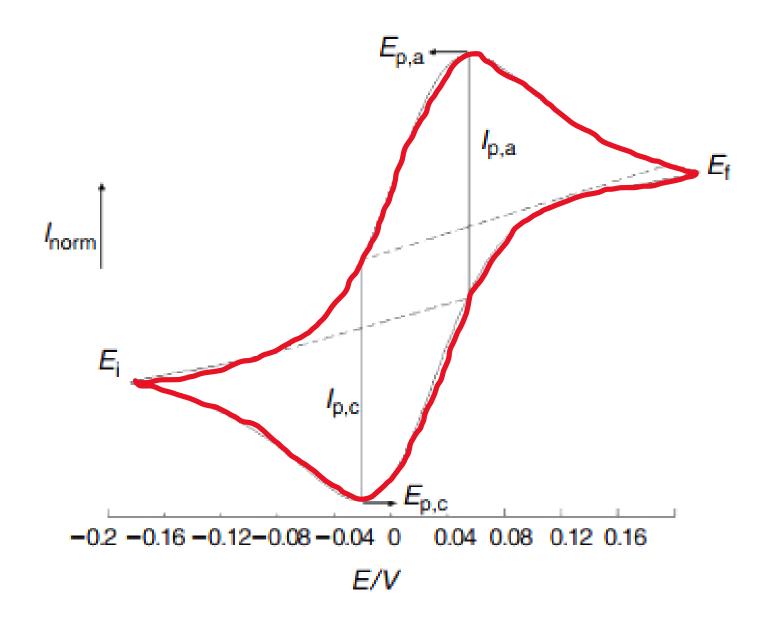
Autolab potentiostat



**(B)** 



(C) Fahmy Taha MH et al., 2020



Addapted from: Gulaboski R., and Pereira C., 2009. Electroanalytical Techniques and Instrumentation in Food Analysis, 379- 402.

Application of voltammetry Identification of different plant bioactive substances

Quantification on different plant bioactive substances Mechanism of interaction with other reductive substances Antioxidative potential of plant bioactive molecules, plant extracts, or antioxidants in plasma

## Experimental work

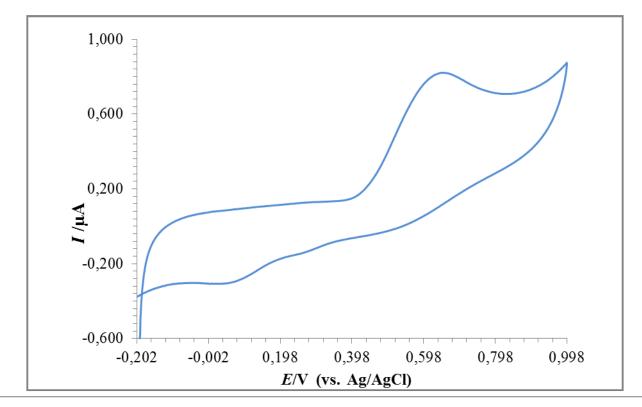
**O**PalmSense Potentiostat connected to a PS Trace system 3.0.

Name	Value	Unit
E vtx1	1	V
E vtx2	-0.2	V
E step	0.001	V
E start	-0.2	V
E cond	0	V
E dep	0	V
scan rate	0.010	V/s
n scans	3	



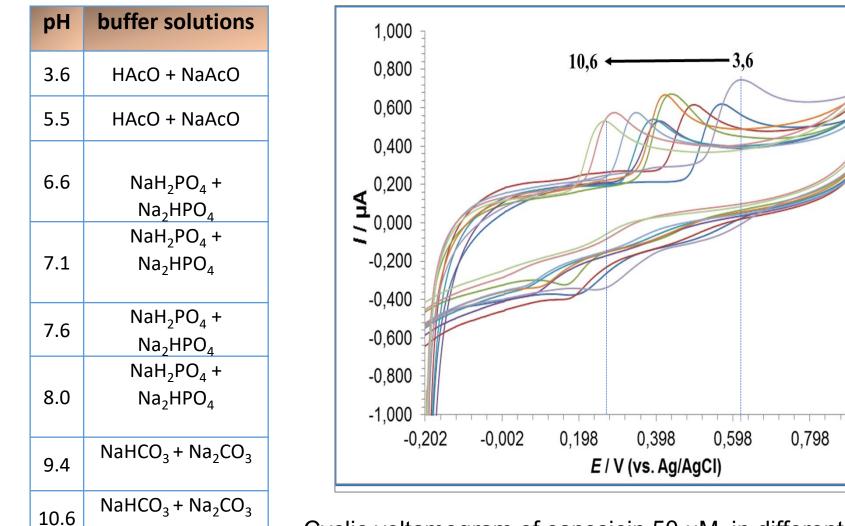
- One compartment 5 ml electrochemical cell in which are immersed three electrodes:
- ≻ Glassy carbon electrode GCE (d=1,5mm) as working electrode,
- > Ag/AgCl (KCl 3mol/L) reference electrode,
- > Pt wire as helping electrode.
- Capsaicin (<99, 8%) Sigma- Aldrich, 96% (V/V) ethanol</li>
- Solution of KCI (0,01mol/L) was used as an electrolyte in the electrochemical cell.

#### Redox behavior of capsaicin



Cyclic voltammogram of capsaicin 100  $\mu$ mol/L, on GCE, scan rate v = 10 mV/s, KCl as electrolyte solution c = 0,010 mol/L.

#### Redox behavior of capsaicin



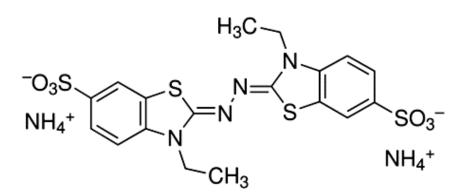
Cyclic voltamogram of capsaicin 50  $\mu$ M, in different buffers (pH =3.6-10.4) and KCl as electrolyte

A

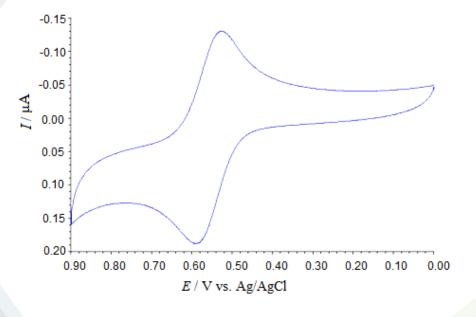
0,998

#### Cyclic voltammetry in measuring antioxidant potential of medicinal plants extracts by using a redox rection of ABTS

 The total antioxidant capacity of these medicinal plants has also been studied by measuring the rate of the homogeneous redox reaction with ABTS (2,2'-azino-bis(3ethylbenzothiazoline-6-sulphonic acid)).



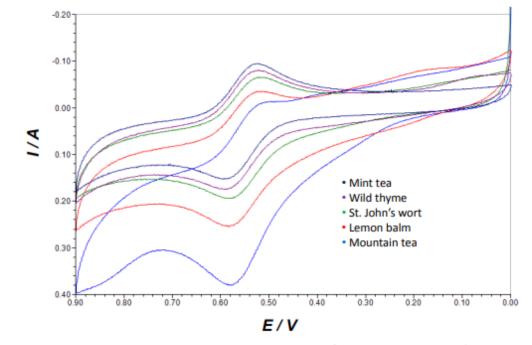
 ABTS++ radical was electrochemically *in situ* generated at the surface of glassy carbon electrode by electrochemical oxidation of ABTS in ethanol electrolyte solution. The method is based on the well-known regenerative catalytic EC' mechanism, where the ABTS++ radical serves as a redox mediator for catalytic oxidation of antioxidants present in the plant infusion.





Naumova G., et al., (2014) *Rapid estimation of antioxidant capacity of some medicinal plants: electrochemical and photometric approaches*. In: 8th CMAPSEEC. Albania.

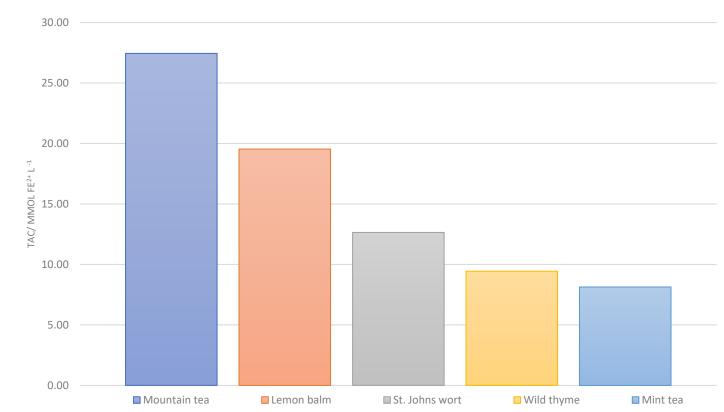
 Cyclic voltammetry performed with plant infusion for investigation of the antioxidant capacity



**Figure 1.** Cyclic voltammograms of Mountain tea (*Origanum vulgare L.*), Lemon balm (*Melissa officinalis L.*), St. John's wort (*Hypericum perforatum L.*), Wild thyme (*Thymus serpyllum L.*) and Mint tea (*Mentha piperita L.*). Extract recorded in 0.1 mmol/L ethanol solution of ABTS and solution of LIClO<sub>4</sub>, at v = 10 mV/s.

Naumova G., et al., (2014) Rapid estimation of antioxidant capacity of some medicinal plants: electrochemical and photometric approaches. In: 8th Conference on medicinal and aromatic plants of southeast european countries, Albania.

In order to confirm cyclic voltammetry results, FRAP assay was also conducted

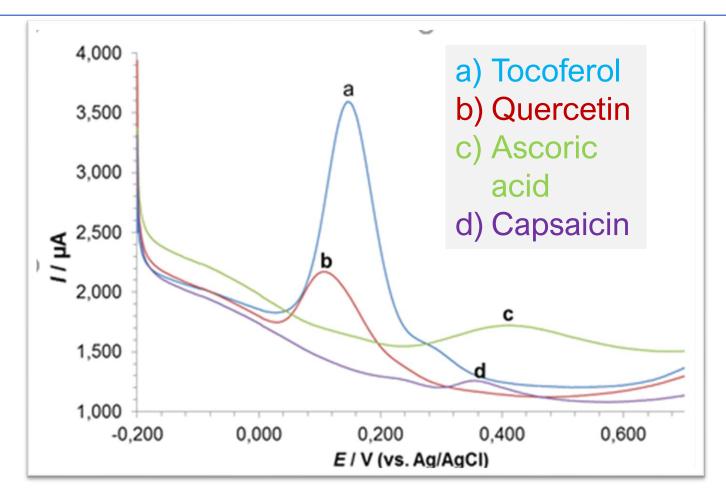


Antioxidant potential of medicinal plants assessed by FRAP method

This method confirmed that mountain tea has the highest total antioxidant capacity

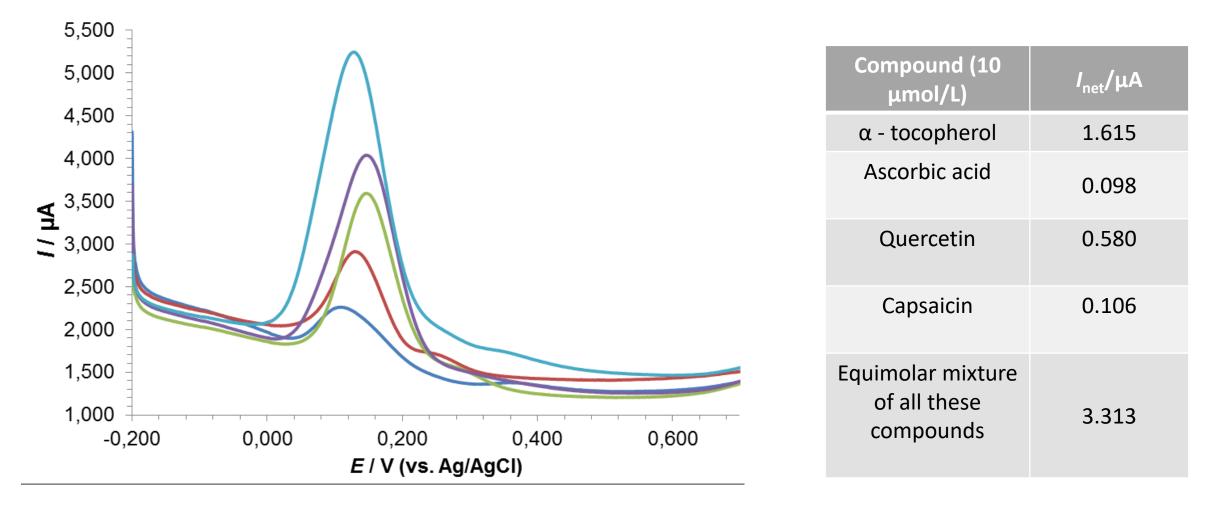
FRAP results have shown no significant differences *v.s* results from CV

### Results obtained from Square Wave Voltammetry (SWV)



SWV of capsaicin, quercetin, ascorbic acid and  $\alpha$ -Tocopherol (10 µmol/L) measured at GCE, pH= 7, potential step d*E* = 0,001 V, amplitude  $E_{sw}$  = 0,05 V and frequency of 10 Hz.

#### Determination of synergistic antioxidant effect of capsaicin



SW voltammograms recorded for the mixture of standard solutions (vitamin E, quercetin, ascorbic acid and capsaicin) in equimolar concentrations (1  $\pm$  0.001/L) at phosphic buffer, pH = 7,1. Instrumental parameters were: d*E* = 0.001V, *E*<sub>sw</sub> = 0.050 V and frequency of 10 Hz.

## Conclusions remarks:

Cyclic voltammetry is a simple and rapid tool for effective estimation of the total antioxidant capacity (TOC) in medicinal plants extracts.

• Infusions prepared from medicinal plants originated from our country exhibit strong antioxidant potential and this fact justify their use as potent natural antioxidant agents.

• There were strong correlations between the results obtained with FRAP method and Cyclic voltammetry and both methods can be used for evaluation of total antioxidant capacity in medicinal plants infusions, but cyclic voltammetry has few adavtages.

# Thank you for your attention!