

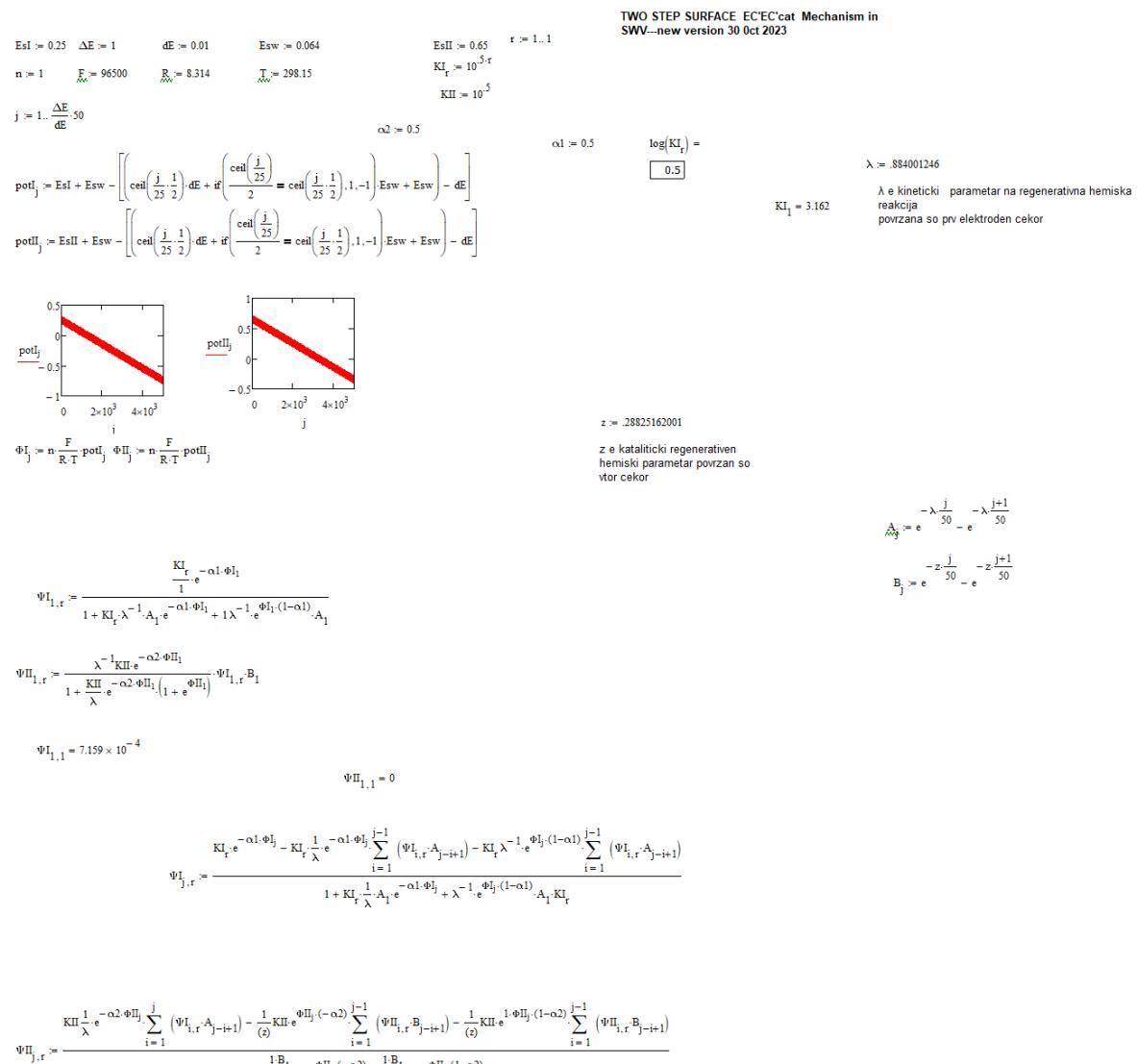
Theory of Two-Step Surface EC'EC' Mechanism in Protein-Film Square-Wave Voltammetry

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Abstract

If two irreversible regenerative chemical reactions get associated to both electrochemically generated products of a two-step successive electrochemical surface mechanism, while causing regeneration of the initial reactant and the intermediate generated electrochemically, then the name of this mechanism in electrochemical terminology is ***EC'EC' mechanism***. In this work, for the first time this complex mechanism is solved under conditions of square-wave voltammetry, and the entire simulation MATHCAD file is provided for free. The voltammetric patterns of this complex mechanism are function of the electron transfer coefficients related to both electron transfer steps, of the kinetics of electron transfer of both electron transfer steps, and of the kinetics of both regenerative chemical steps. Importance of this mechanism is found in the redox behavior of lipophilic redox enzymes whose electrochemical transformation occurs in two successive steps.



$$\Psi_{j,t} = \Psi_{j,t}^I + \Psi_{j,t}^{II}$$

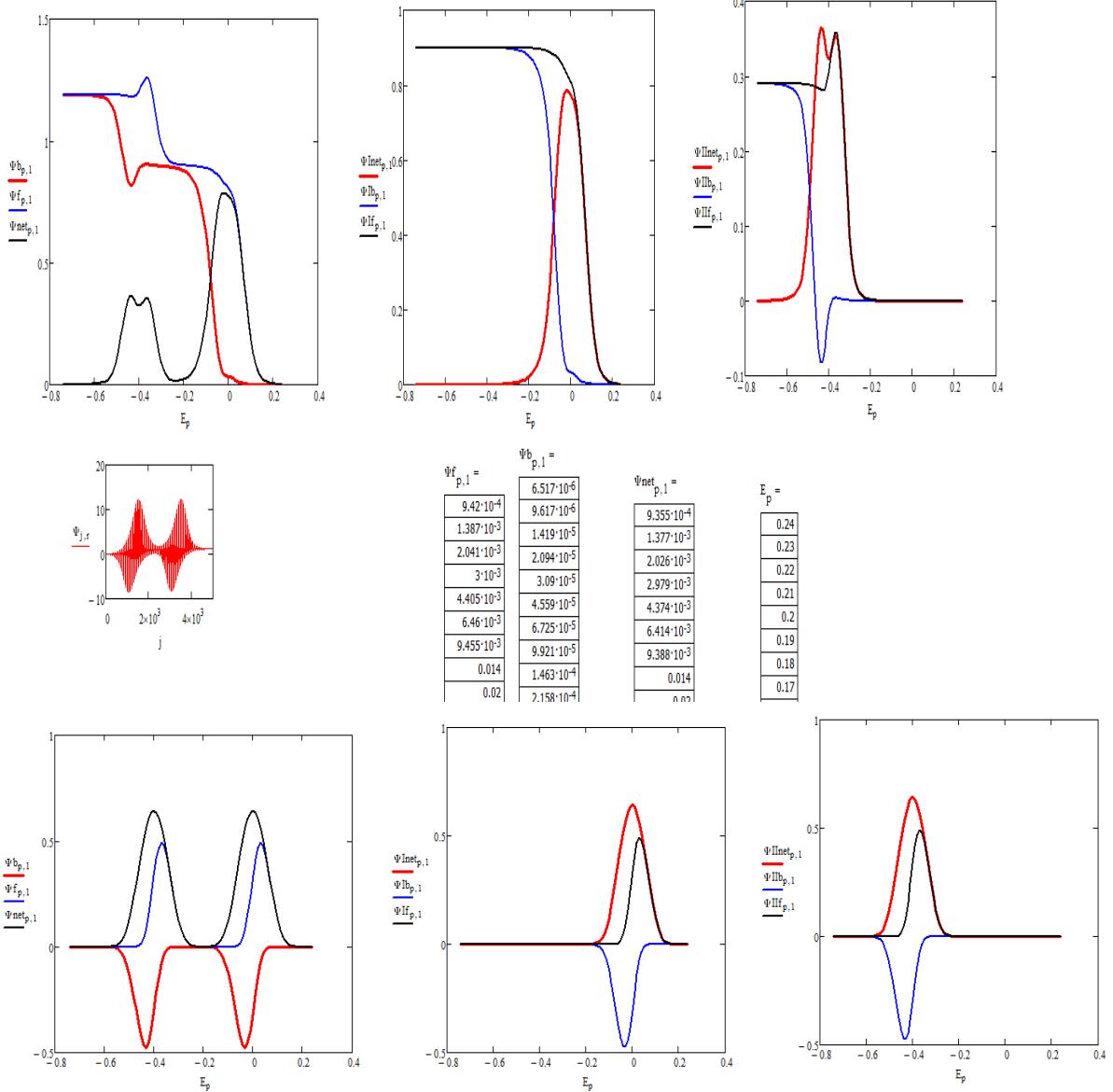
$$p = 1 - \left(\frac{\Delta E}{dE} \right) - 1$$

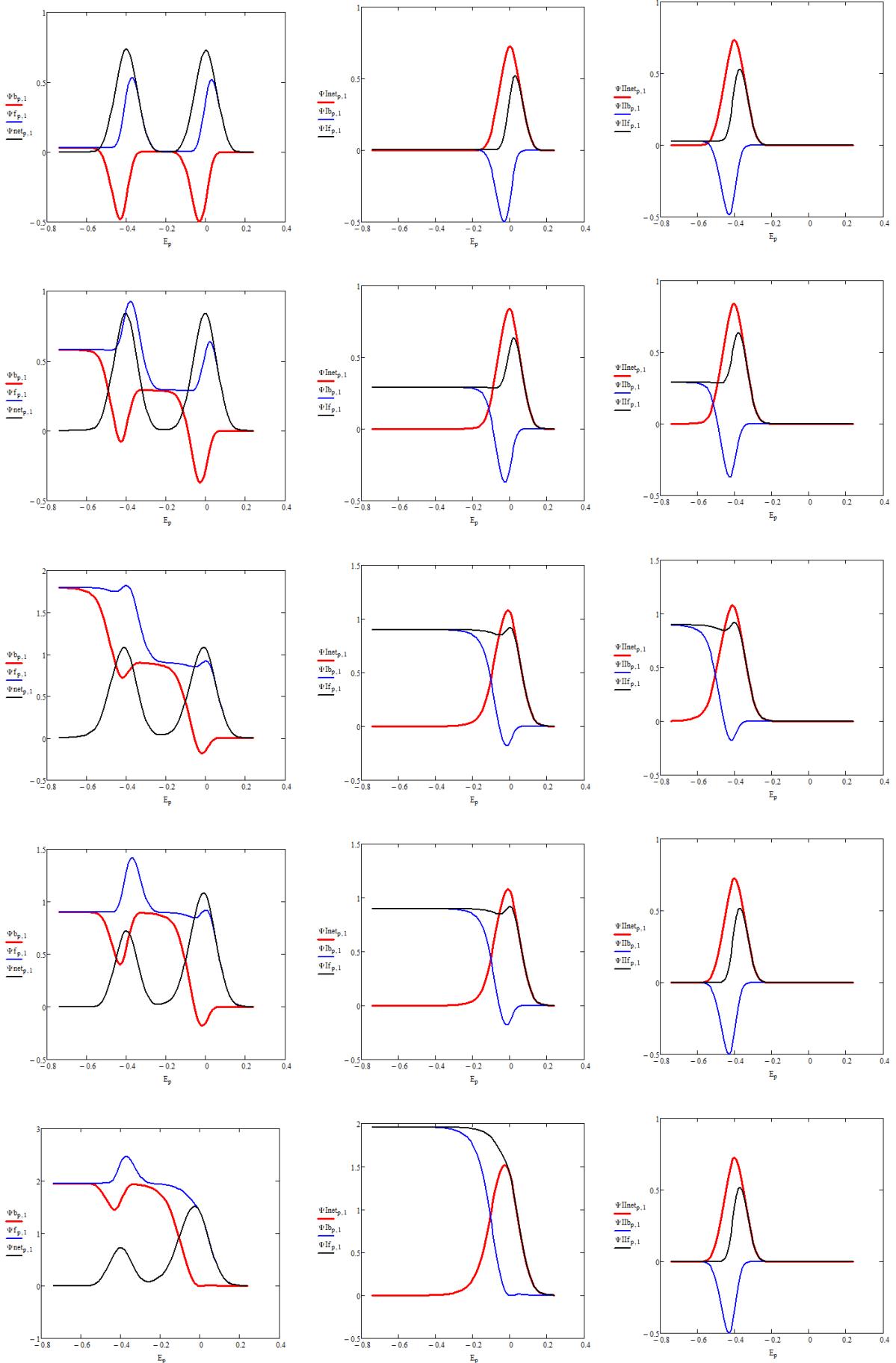
$$\Psi_{f,p,t} = \Psi_{(p+1)50,t}, \Psi_{b,p,t} = \Psi_{50,p+1}, \Psi_{net,p,t} = \Psi_{f,p,t} - \Psi_{b,p,t}$$

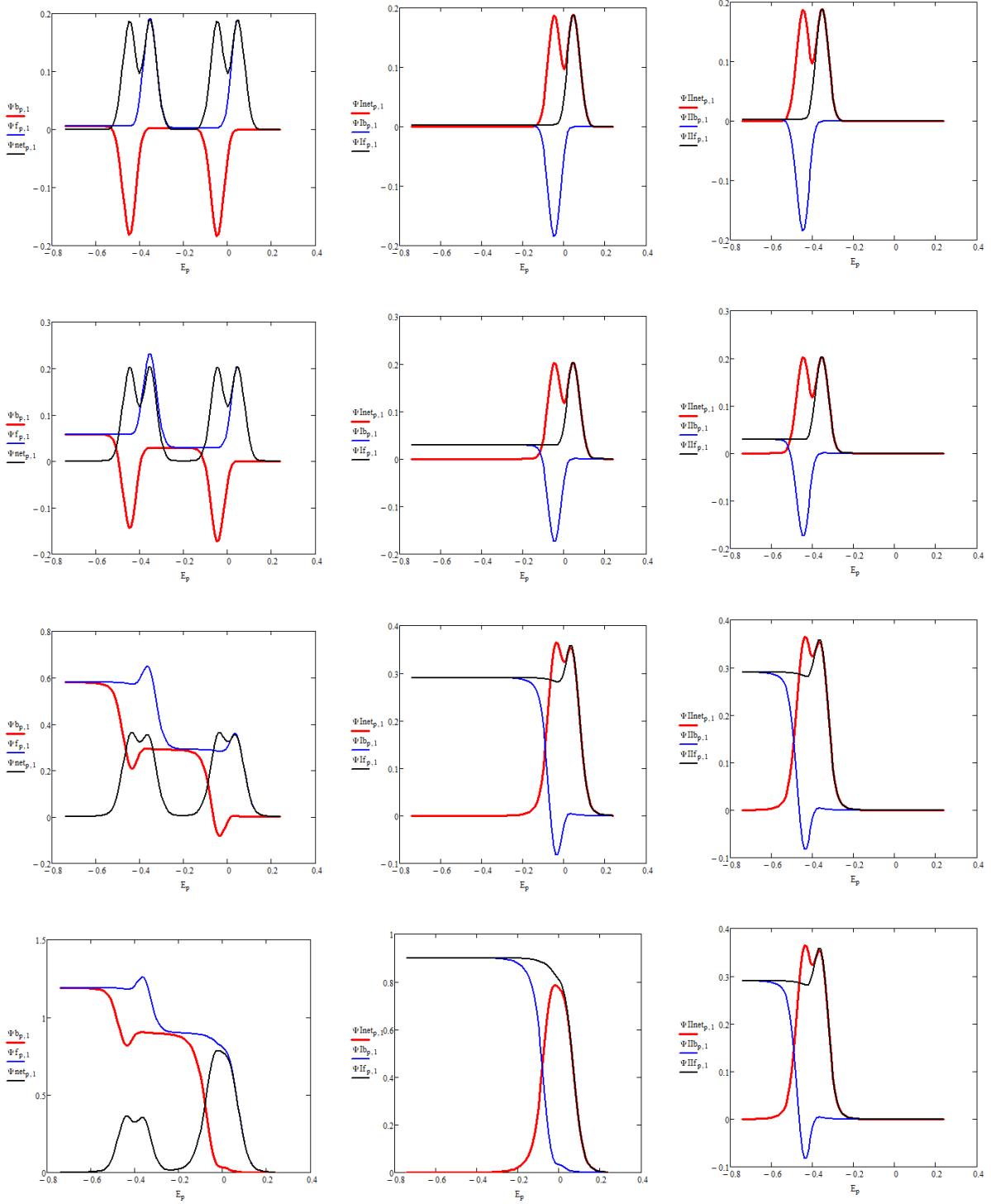
$$\Psi_{IIb,p,t} = \Psi_{50,p+25,t}, \Psi_{III,f,p,t} = \Psi_{(p+1)}, \Psi_{IInet,p,t} = \Psi_{III,f,p,t} - \Psi_{IIb,p,t}$$

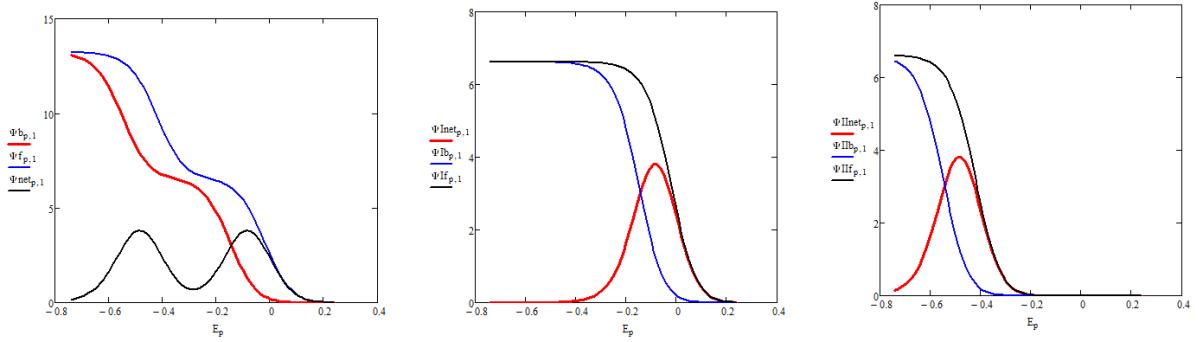
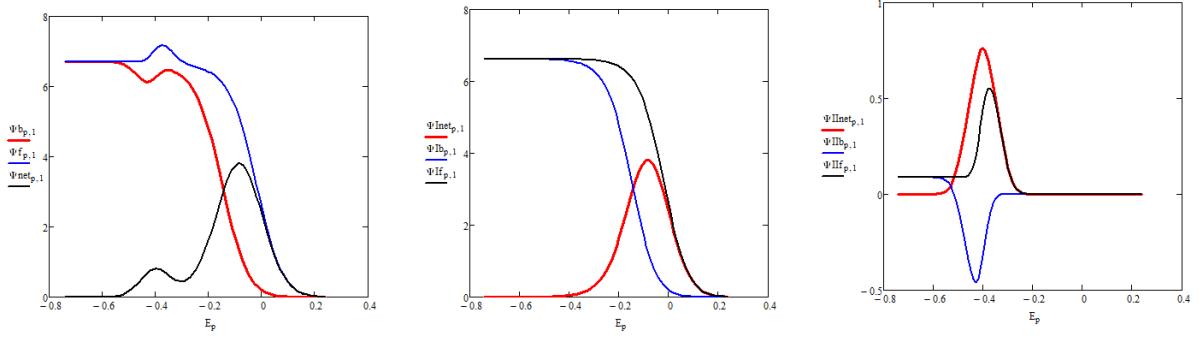
$$E_p = EsI - p \cdot dE$$

$$\Psi_{b,p,t} = \Psi_{50,p+25,t}, \Psi_{f,p,t} = \Psi_{(p+1)50,t}, \Psi_{net,p,t} = \Psi_{f,p,t} - \Psi_{b,p,t}$$









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