

Influence of beta-cyclodextrin on the phase transition in carbamazepine polymorphs

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INTRODUCTION

Carbamazepin (CBZ), a widespread used antiepileptic drug, branded as Tegretol, firstly was launched of Active in form Pharmaceutical Ingredient (API) as a commercially available polymorphic form III. Additionally, crystallographic studies on crystal packing motifs of carbamazepine molecules in crystal lattices, reveal that, a part of the firstly confirmed structure of polymorphic form III, this drug exists in three other polymorphic forms I, II and IV, as well as pseudopolymorph, dihyrdate form. solid-state Further [1] testing confirmed that in terms Of thermodynamical stability CBZ III and I are related as enantiotropic pair.

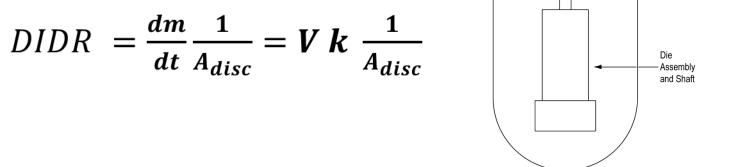
PURPOSE OF THE STUDY

Compromising the differences in crystal packing between CBZ polymorphs III and I that exert differences in density and solubility with the requirements for sufficient plasma concentration available for favorable crossing the blood-brain barrier (BBB) and reaching receptor sites, remain the challenge for crystal engineering CBZ polymorphs with functional excipients which, based on their molecular structures, are appropriate for formation either of inclusion complexes (IC). In terms of processing the CBZ either with solvents or polymorph, by mechanochemical treatments, both technologies impose phase transition; The outlined research objectives address the testing of the influence of cyclodextrin (BCD), native cyclic beta oligosaccharide, on controlling the phase transition of CBZ form III to form I, respectively toward the formation of inclusion complex by non-covalent interactions between nonpolar part of CBZ molecule and hydrophobic CBD cavity that interact each other in stoichiometric ratio [2,3].

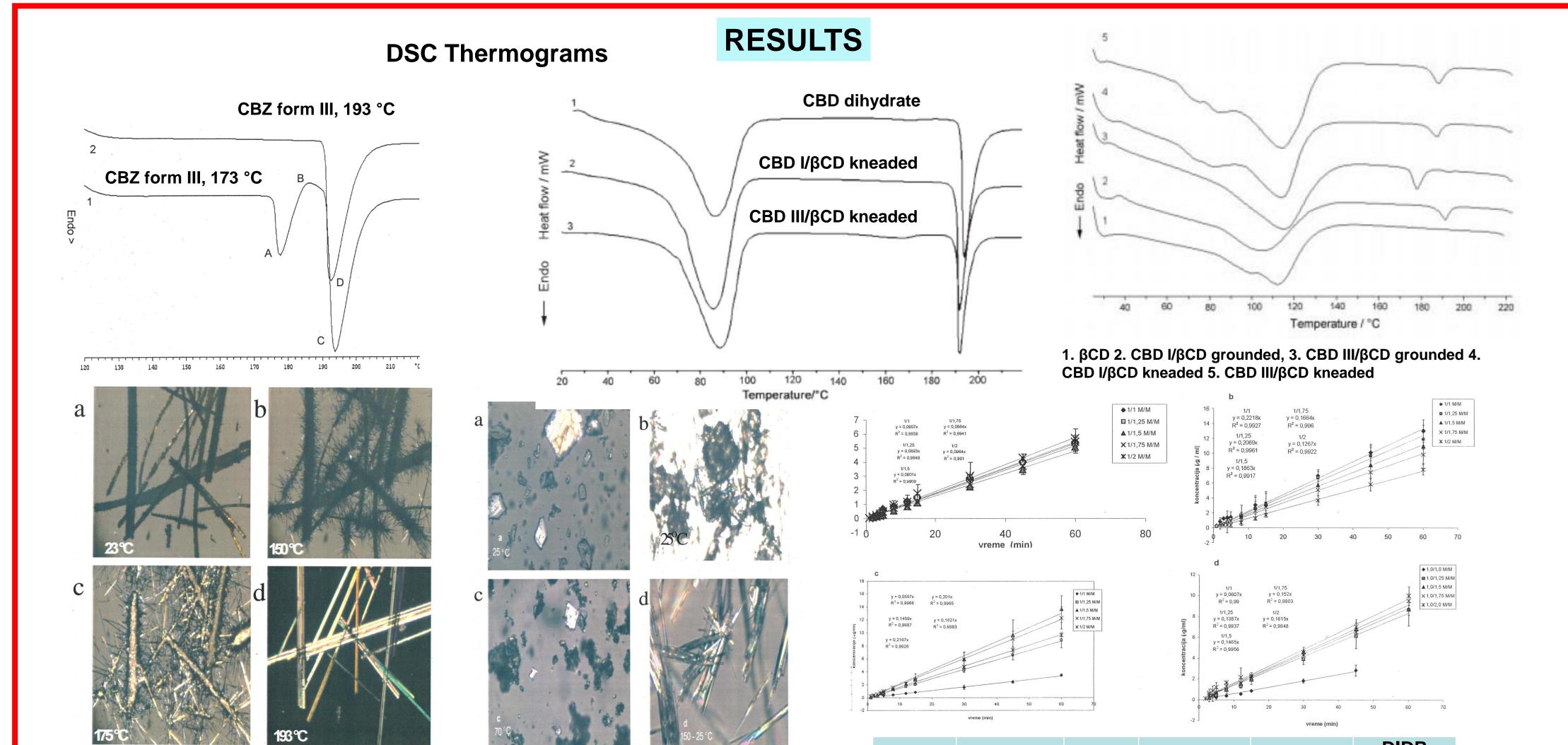
METHODS

Mechanochemical treatment (grinded & kneading) of binary systems: CBZ form I & βCB and CBZ form II & βCB 1/1; 1/1.5 & /2 *M/M*

Rotating disc method



DIDR- disc dissolution rate µg/cm²min⁻¹ *m* - mass (µg), t - time (min), A_{disc} - the disc surface area (cm²), V - the volume of the medium (mL), *k* - the slope of the straight line from the dissolution profile (μ g/(min × mL)



Hot-Stege Microscopy; phase transitions

LITERATURE

1. Grzesiak AL, Lang M, Kim K, Matzger AJ. Comparison of the four anhydrous polymorphs of carbamazepine and the crystal structure of form I. J Pharm Sci. 2003 Nov;92(11):2260-71

2. Li Y, Han J, Zhang GG, Grant DJ, Suryanarayanan R. In situ dehydration of carbamazepine dihydrate: a novel technique to prepare amorphous anhydrous carbamazepine. Pharm Dev Technol. 2000;5(2):257-66

3. Cvetkovski, A., Bettini, R., Tasic, Lj., Stupar, M., Casini, L, Rossi, A, Giordano, F., Thermal properties of binary mixtures of β-cyclodextrin with carbamazepine polymorphs. J. Therm. Anal. Cal., 68, (2002) 669–678

	CBZ form	Treatment	Onset ℃	Enthalpy J/g	Onset ℃	DIDR µg∙cm [−] 2- min [−]
	ш	Grinded	173.6	101.7(2.4)	188.6	77.11
	- 111	Kneaded			181.5	
	1 I.	Grinded			186.3	
	- I	Kneaded			181.9	72.02

Intrinsic Dissolution Rate Measurements

CONCLUSION & FURTHER WORK

These results confirm that phase transition of CBZ polymorphs, anhydrous form III and I leading to less water soluble CBZ pseudopolymoprph, dihydrate form during the kneading and formation of IC with BCD occur simultaneously. Less water soluble and stable at ambient temperature CBZ III, thus less favorable for transition in CBZ dihydrate, in grinded binary sample with BCD exerts higher water solubility due to formation of IC. BCD in high extent retains inhibition of phase transition of CBZ III to CBZ I during the DCS heating cycle. ICs as immediate release drug delivery systems offer further opportunities for design modified release formulations that may include IC in combination a polymeric matrix formation compounds for additional control of both the drug release profile and drug phase transition The further work intends to put in evidence biopharmaceutical profiles CBZ form I/βCB and CBZ form II/βCB.

