

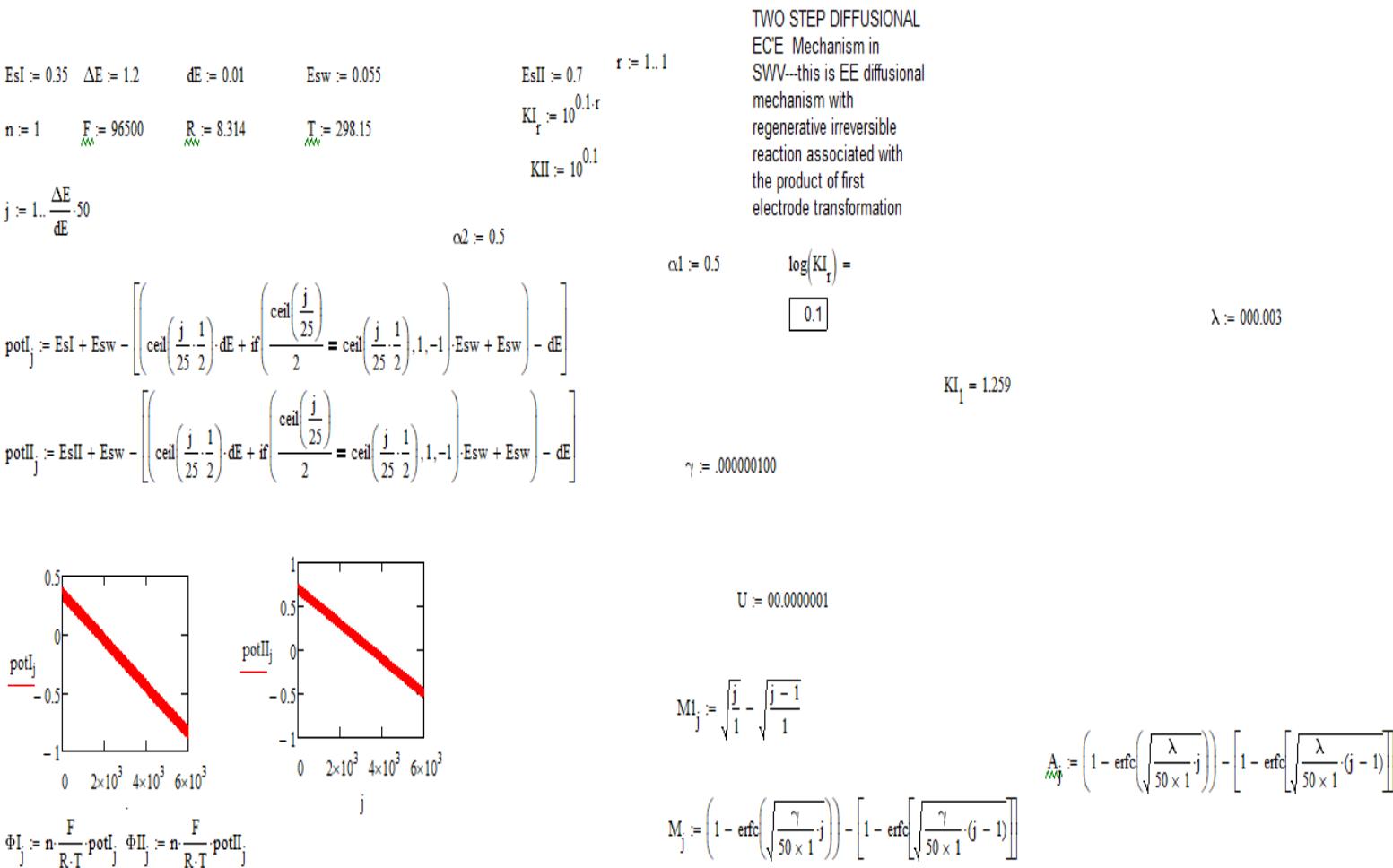
## Two Step Diffusional EE Mechanism Associated with Irreversible Regenerative Reaction to First Electrode Step-MATCAD Simulation Protocol

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### **Abstract**

Redox transformation of many water-soluble redox proteins occurs via two step electrochemical process. In such systems, where often an irreversible chemical process of regenerative nature is associated to the product of first redox electrode transformation. In our most recent work, we have developed a theoretical model for simulating this complex electrode mechanism in square-wave voltammetry, and we offer for free the entire simulation protocol. This is the first theoretical model solved mathematically for this complex electrode mechanism that is very important to understand the features of complex electrochemical transformations in enzymatic voltammetry



$$\Psi_{I_1,r} := \frac{\frac{2}{\sqrt{\pi \cdot 50}} KII \cdot e^{-\alpha 2 \cdot \Phi II_1}}{1 + \frac{KII \cdot M1_1 \cdot 2}{\sqrt{\pi \cdot 50}} \cdot e^{-\alpha 2 \cdot \Phi II_1} \cdot (1 + e^{\Phi II_1})} \cdot \Psi_{I_1,r} + \frac{KII \cdot e^{-\alpha 2 \cdot \Phi II_1} - \frac{2 \cdot KII \cdot e^{-\alpha 2 \cdot \Phi II_1}}{\sqrt{\pi \cdot 50}} \cdot 0 - \frac{2 \cdot KII \cdot e^{(1-\alpha 2) \cdot \Phi II_1}}{\sqrt{\pi \cdot 50}} \cdot \frac{U}{(1+U) \cdot 1} \cdot 0 - \frac{\gamma}{1+U} \cdot e^{(1-\alpha 2) \cdot \Phi II_1 \cdot 0}}{1 + \frac{2 \cdot KII \cdot M1_1 \cdot e^{-\alpha 2 \cdot \Phi II_1}}{\sqrt{\pi \cdot 50}} + \frac{2 \cdot KII \cdot e^{(1-\alpha 2) \cdot \Phi II_1}}{\sqrt{\pi \cdot 50}} \cdot \frac{U \cdot M1_1}{(1+U) \cdot 1} + \frac{\gamma}{1+U} \cdot e^{(1-\alpha 2) \cdot \Phi II_1 \cdot M1_1}} \cdot 1$$

$$\Psi_{I_1,1} = 6.141 \times 10^{-8}$$

$$\Psi_{II_1,1} = 4.978 \times 10^{-7}$$

$$x := 0.001$$

$$\Psi_{I_j,r} := \frac{\frac{KI_r}{1} \cdot 1 \cdot e^{-\alpha 1 \cdot \Phi I_j} - \frac{KI_r \cdot \lambda^{-1} \cdot \frac{2}{\sqrt{4}}}{1} \cdot e^{-\alpha 1 \cdot \Phi I_j} \cdot \sum_{i=1}^{j-1} (\Psi_{I_i,r} \cdot A_{j-i+1}) - \frac{KI_r \cdot \lambda^{-1} \cdot \frac{2}{\sqrt{4}} \cdot e^{\Phi I_j \cdot (1-\alpha 1)}}{1} \cdot \sum_{i=1}^{j-1} (\Psi_{I_i,r} \cdot A_{j-i+1})}{1 + \frac{KI_r \cdot \lambda^{-1} \cdot A_1}{\frac{2}{\sqrt{4}}} \cdot e^{-\alpha 1 \cdot \Phi I_j} + 1 \lambda^{-1} \cdot e^{\Phi I_j \cdot (1-\alpha 1)} \cdot A_1 \cdot \frac{1}{\frac{\sqrt{4}}{\sqrt{4}}}}$$

$$\Psi_{II_j,r} := \frac{\frac{2}{\sqrt{\pi \cdot 50}} KII \cdot e^{-\alpha 2 \cdot \Phi II_j}}{1 + 0} \cdot \sum_{i=1}^j (\Psi_{I_i,r} \cdot M1_{j-i+1}) - KII \frac{2}{\sqrt{\pi \cdot 50}} \cdot e^{-\alpha 2 \cdot \Phi II_j} \cdot \sum_{i=1}^{j-1} (\Psi_{II_i,r} \cdot M1_{j-i+1}) - \frac{2}{\sqrt{\pi \cdot 50}} KII \cdot e^{1 \cdot \Phi II_j \cdot (1-\alpha 2)} \cdot (1) \cdot \sum_{i=1}^{j-1} (\Psi_{II_i,r} \cdot M1_{j-i+1})$$

$$1 + \frac{KII \cdot \frac{2 \cdot M1_1}{\sqrt{\pi \cdot 50}}}{1} \cdot e^{-\alpha 2 \cdot \Phi II_j} \cdot (1 + e^{\Phi II_j})$$

$$\Psi_{j,r} := \Psi_{I_j,r} + \Psi_{II_j,r}$$

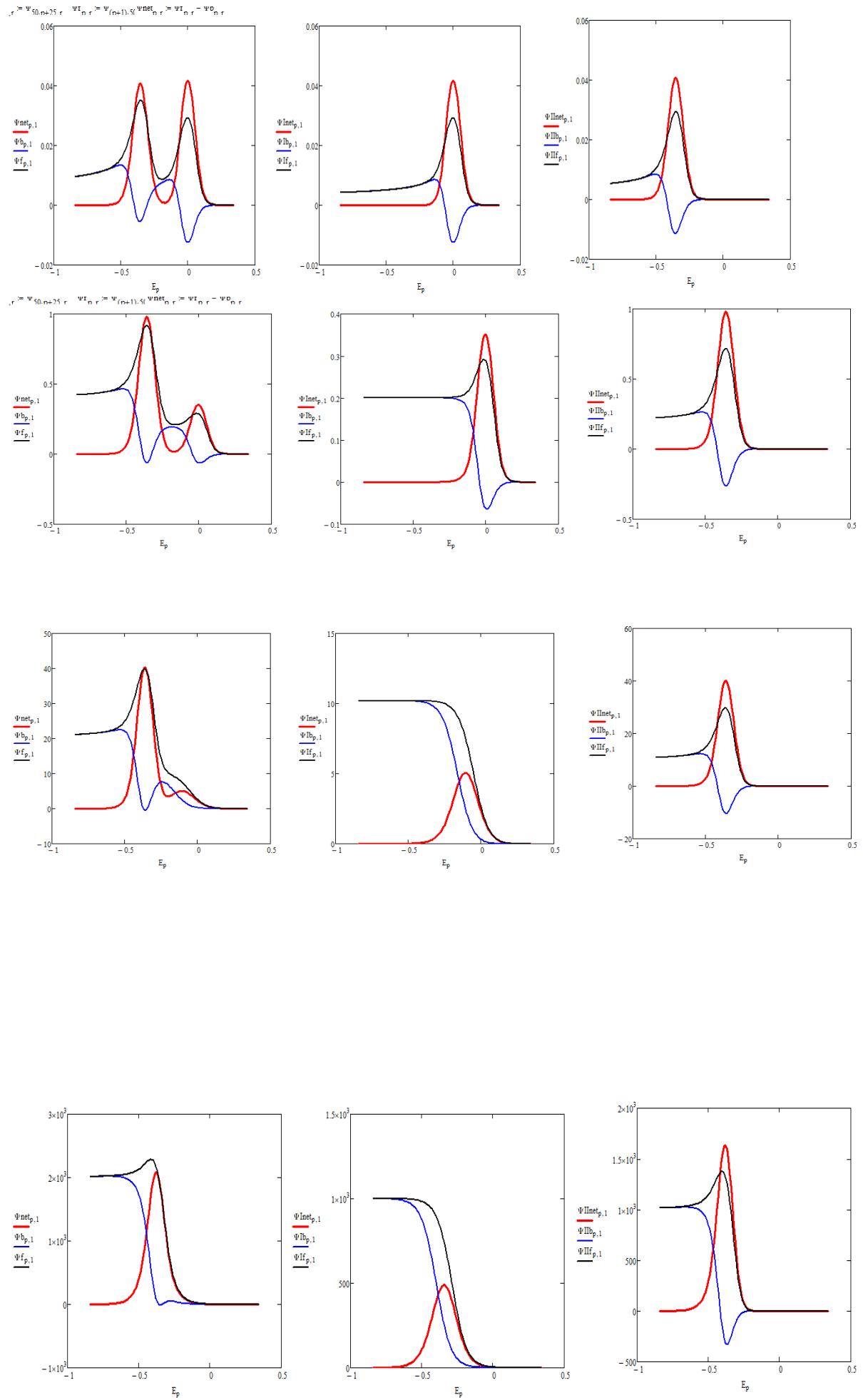
$$p := 1 .. \left( \frac{\Delta E}{dE} \right) - 1$$

$$\Psi If_{p,r} := \Psi I_{(p+1) \cdot 50,r} \quad \Psi Ib_{p,r} := \Psi I_{50,p+2} \quad \Psi Inet_{p,r} := \Psi If_{p,r} - \Psi Ib_{p,r}$$

$$\Psi IIb_{p,r} := \Psi II_{50,p+25,i} \quad \Psi If_{p,r} := \Psi II_{(p+1)} \quad \Psi Inet_{p,r} := \Psi If_{p,r} - \Psi IIb_{p,r}$$

$$E_p := EsI - p \cdot dE$$

$$\Psi b_{p,r} := \Psi I_{50,p+25,r} \quad \Psi f_{n,r} := \Psi I_{(n+1),51} \quad \Psi net_{n,r} := \Psi f_{n,r} - \Psi b_{n,r}$$



## REFERENCES

1. R. Gulaboski, M. Chirea, C. M. Pereira, M. N. D. S. Cordeiro, R. B. Costa, A. F. Silva, *J. Phys. Chem. C* 112 (2008) 2428-2435
2. B. Sefer, R. Gulaboski, V. Mirceski, *Journal of Solid State Electrochemistry* 16 (2012) 2373-2381.
3. V. Mirceski, R. Gulaboski, F. Scholz, *Electrochemistry Communications* 4 (2002) 813-818
4. R. Gulaboski, V. Mirceski, S. Mitrev, *Food Chemistry* 138 (2013) 116-121
5. R. Gulaboski, M. Chirea, C. M. Pereira, M. N. D. S. Cordeiro, R. B. Costa, A. F. Silva, *J. Phys. Chem. C* 112 (2008) 2428-2435
6. R. Gulaboski, V. Mirceski, S. Komorsky-Lovric, M. Lovric, *Electroanalysis* 16 (2004) 832-842
7. R. Gulaboski, C. M. Pereira, M. N. D. S. Cordeiro, A. F. Silva, M. Hoth, I. Bogeski, *Cell Calcium* 43 (2008) 615-621
8. V. Mirceski, R. Gulaboski, *Bulletin of the Chemists and Technologists of Macedonia* 18 (1999) 57-64.
9. R. Gulaboski, C. M. Pereira, *Electroanalytical Techniques and Instrumentation in Food Analysis*; in Handbook of Food Analysis Instruments (2008) 379-402.
10. M. Jorge, R. Gulaboski, C. M. Pereira, M. N. D. S. Cordeiro, *Journal of Physical Chemistry B* 110 (2006) 12530-12538.
11. V. Mirceski, D. Guziejewski, L. Stojanov, R. Gulaboski, *Analytical Chemistry* 91 (2019) 14904-14910.
12. V. Mirceski, R. Gulaboski, F. Scholz, *Journal of Electroanalytical Chemistry* 566 (2004) 351-360.
13. R. Gulaboski, V. Mirceski, S. Mitrev, *Food Chemistry* 138 (2013) 116-121
14. R. Gulaboski, V. Mirceski, S. Komorsky-Lovric, M. Lovric, *Electroanalysis* 16 (2004) 832-842
15. R. Gulaboski, C. M. Pereira, M. N. D. S. Cordeiro, A. F. Silva, M. Hoth, I. Bogeski, *Cell Calcium* 43 (2008) 615-621
16. R. Gulaboski, V. Mirceski, F. Scholz, *Amino Acids* 24 (2003) 149-154

17. V. Mirceski, R. Gulaboski, *Croatica Chemica Acta* 76 (2003) 37-48.
18. F. Scholz, R. Gulaboski, *Faraday Discussions* 129 (2005) 169-177.
19. V. Mirceski, R. Gulaboski, F. Scholz, *Electrochemistry Communications* 4 (2002) 814-819.
20. R. Gulaboski, K. Caban, Z. Stojek, F. Scholz, *Electrochemistry Communications* 6 (2004) 215-218.
21. M. Janeva, P. Kokoskarova, V. Maksimova, R. Gulaboski, *Electroanalysis* 31 (2019) 2488-2506
22. V. Mirceski, R. Gulaboski, F. Scholz, *Electrochemistry Communications* 4 (2002) 814-819