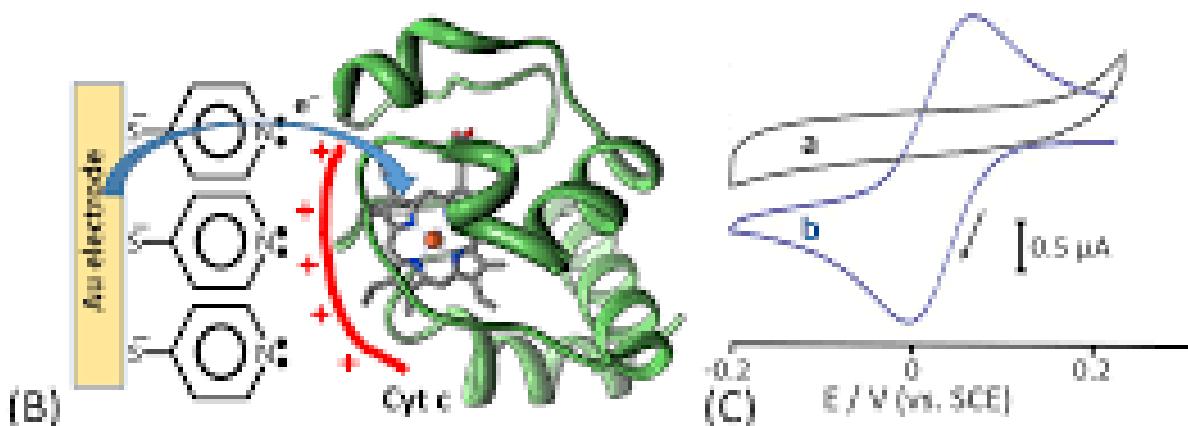
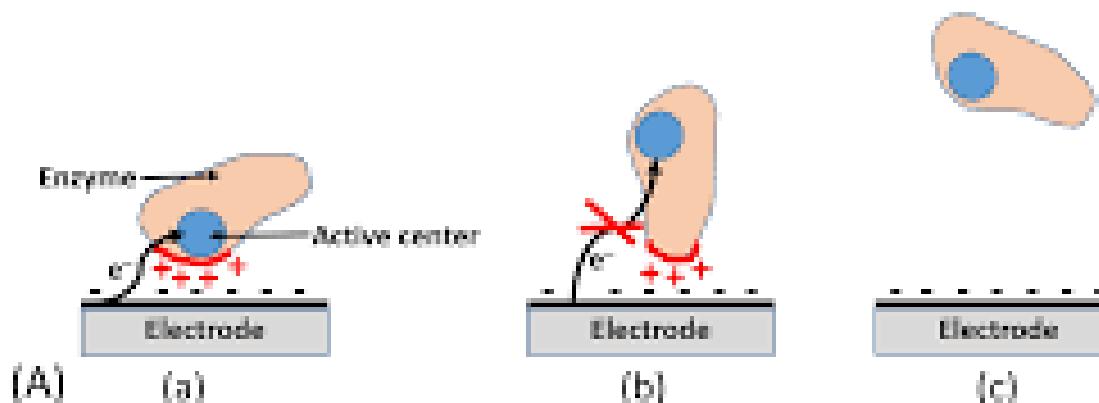
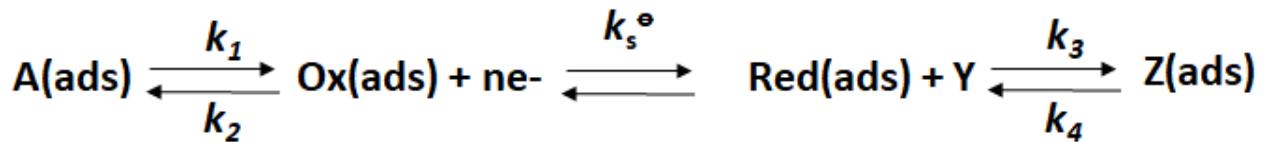


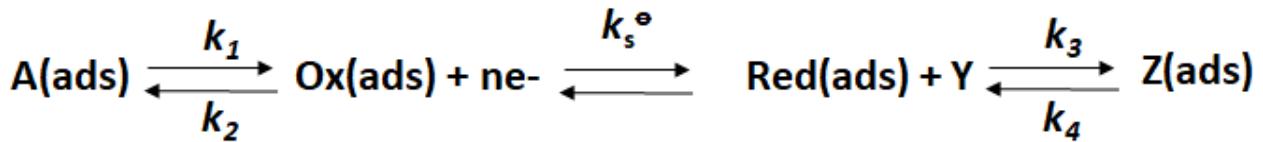
Square-wave voltammetry of Lipophilic Redox Enzymes Featuring Electrode Reaction Associated with Two Reversible Chemical Steps

Pavlinka Kokoskarova, Sonja Risafova, Rubin Gulaboski

Faculty of Medical Sciences, Goce Delcev University Stip
MACEDONIA



The model we consider is a surface electrode mechanism Whose electron transfer step is associated with preceding and follow up reversible chemical reactions



$$t = 0; \Gamma(\text{A}) = \Gamma^*(\text{A}); \Gamma(\text{Ox}) = K_{\text{eq}} \Gamma^*(\text{A}); \Gamma(\text{Red}) = 0 \quad (\text{a})$$

$$t > 0; \Gamma(\text{A}) + \Gamma(\text{Ox}) + \Gamma(\text{Red}) + \Gamma(\text{Z}) = \Gamma^*(\text{A}); \quad (\text{b})$$

$$t > 0; K_{\text{eq, preceding}} = k_1/k_2; K_{\text{eq, follow-up}} = k_3/k_4 \quad (\text{c})$$

For $t > 0$, the following differential equations (d-f) link the surface concentrations with the current and the kinetics and thermodynamics of preceding and follow-up chemical steps:

$$(d) \quad (d\Gamma(\text{A})/dt) = k_1 \Gamma(\text{Ox}) - k_2 \Gamma(\text{A})$$

(d)

$$(e) \quad d\Gamma(\text{Ox})/dt = -I/(nFS) - k_2 \Gamma(\text{Ox}) + k_1 \Gamma(\text{A})$$

(e)

$$(f) \quad d\Gamma(\text{Red})/dt = I/(nFS) + k_3 \Gamma(\text{Red}) - k_4 \Gamma(\text{Z})$$

(f)

The electric current, the applied electrical potential difference, the surface concentrations of the electrochemically active species, and the other parameters relevant to electron transfer step are assumed to be interrelated via the Butler-Volmer equation, having the following form:

$$(g) \quad (I/nFS) = k_s^e \exp(-\alpha\Phi) [\Gamma(\text{Ox}) - \exp(\Phi) \Gamma(\text{Red})]$$

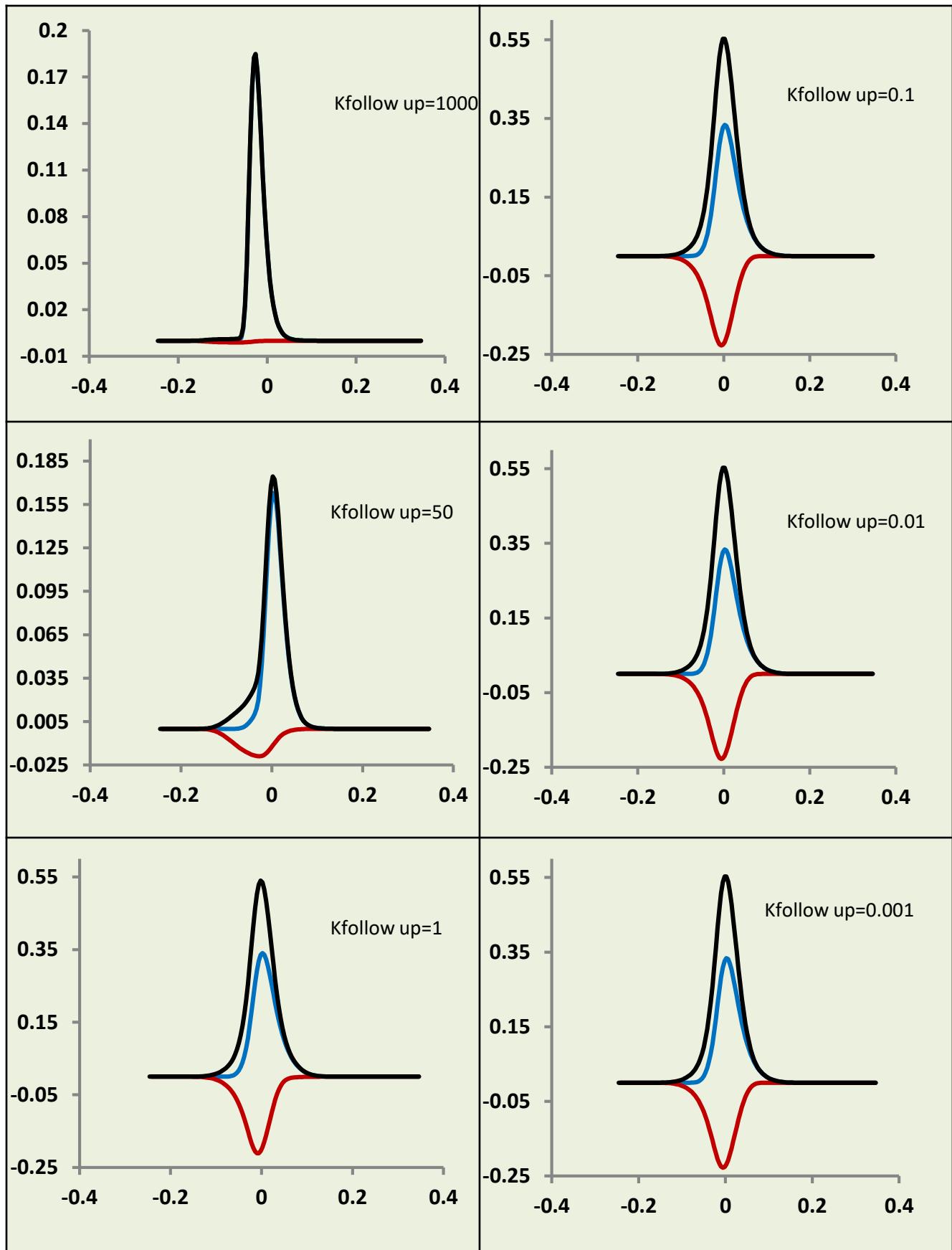
Кај сите овие 6 графици

$$U = K_{eq} = 10$$

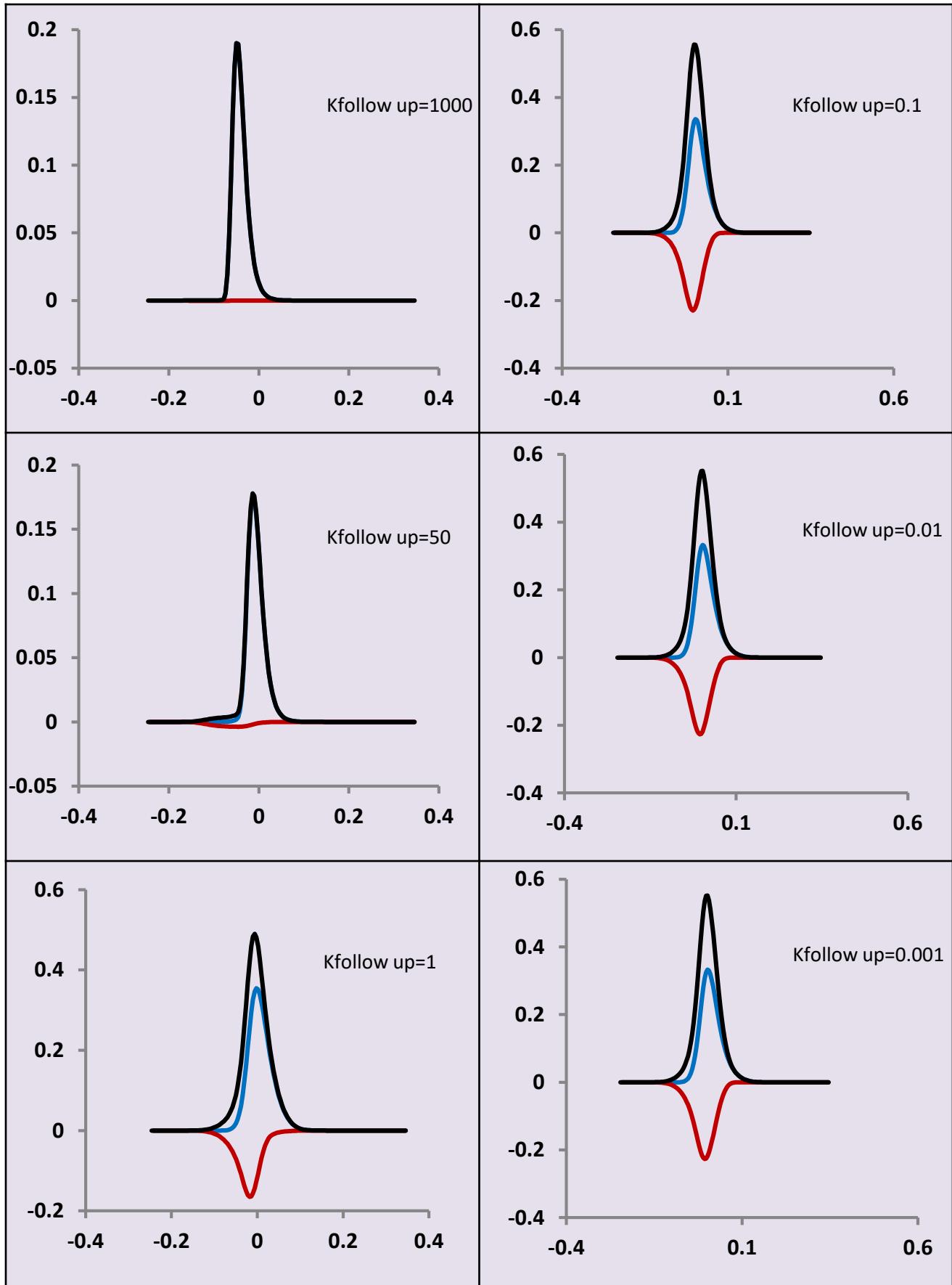
and

$$b = K_{chem} = 10^3, 50, 10; 1:$$

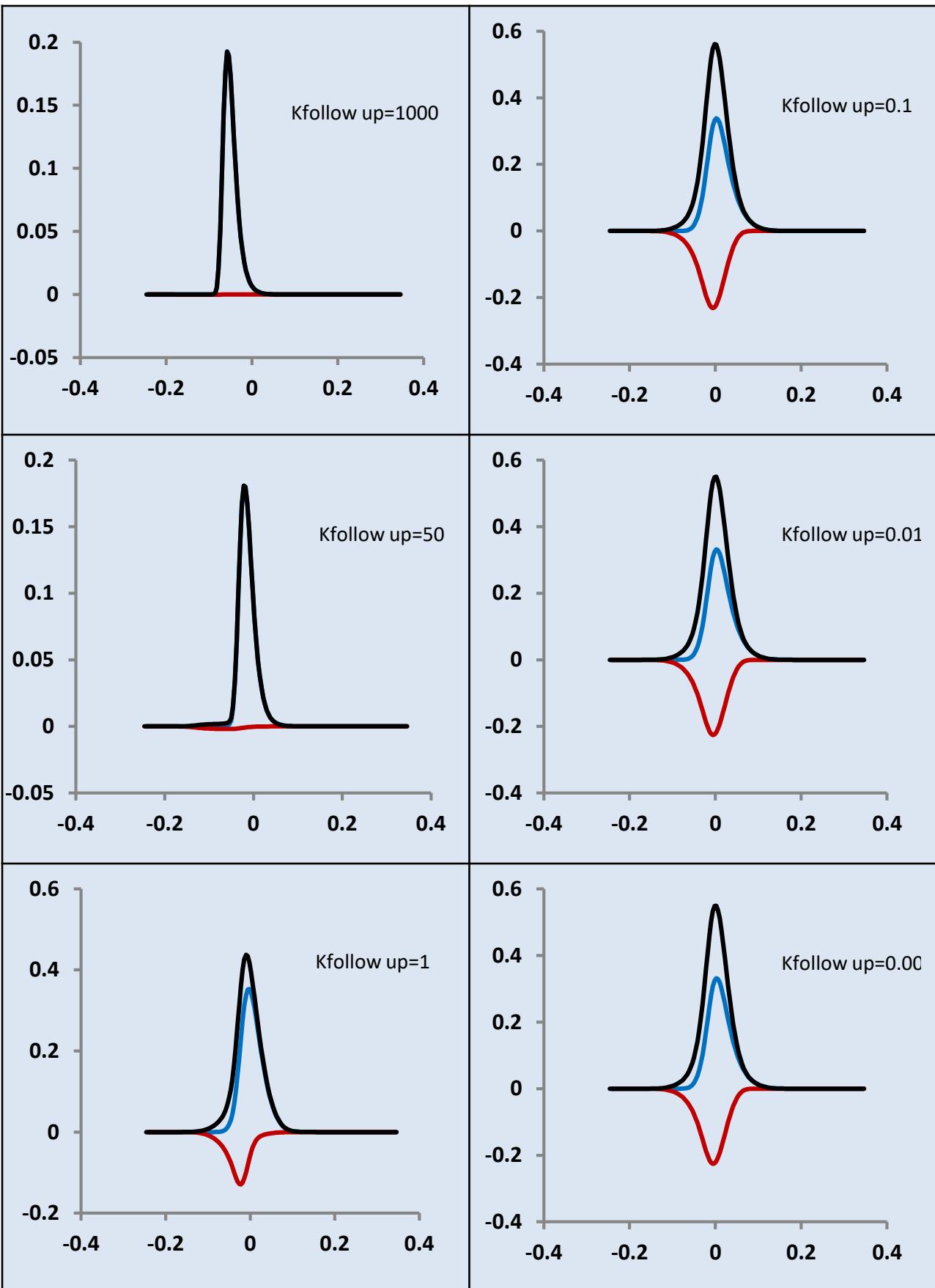
$$0.01, 0.001$$



- Кaj сите овие 6 графици
 $U=1$
 $a, b = 1000, 50, 1, 0.1, 0.01, 0.001$



- Кaj сите овие 6 графици
 $U=0.01$
 $a b=1000, 50, 1, 0.1, 0.01, 0.001$



$K_S = 10^{1,25}$

$U = 1000$

$b = 0.001$

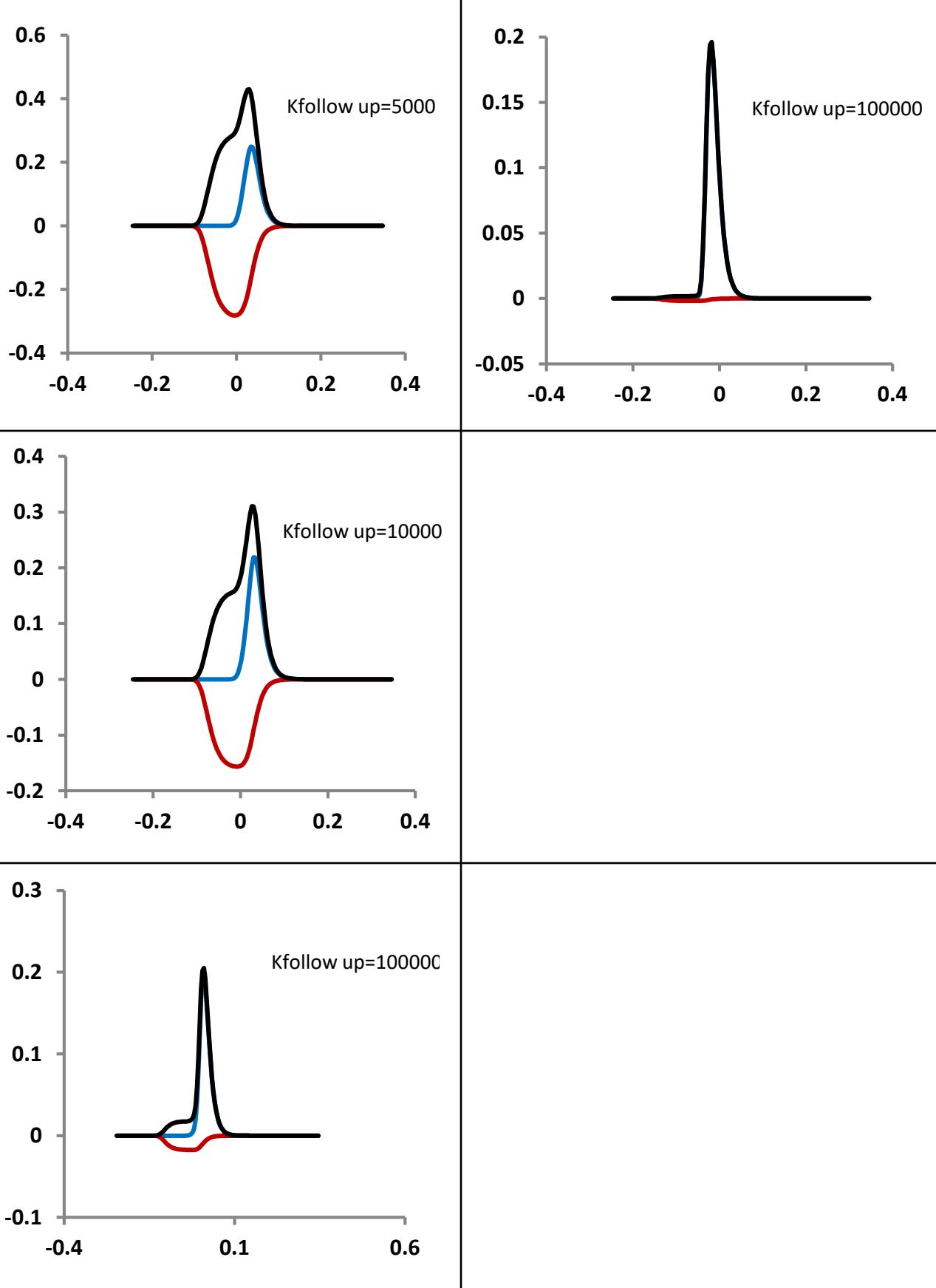
0.01

0.1

1

50

1000



$K_S = 10^{1,25}$

$U = 10$

$b = 0.001$

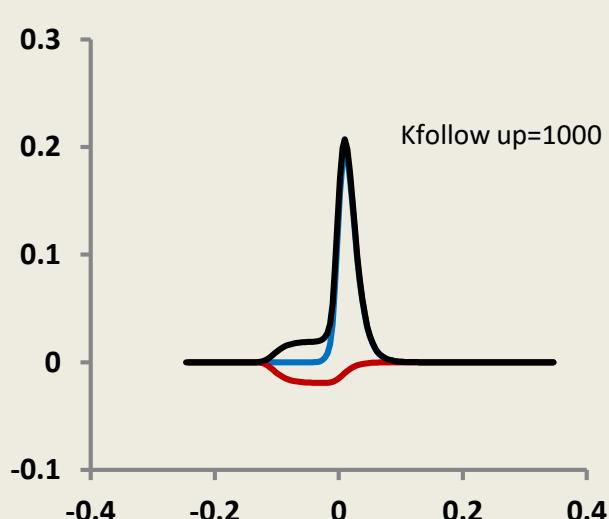
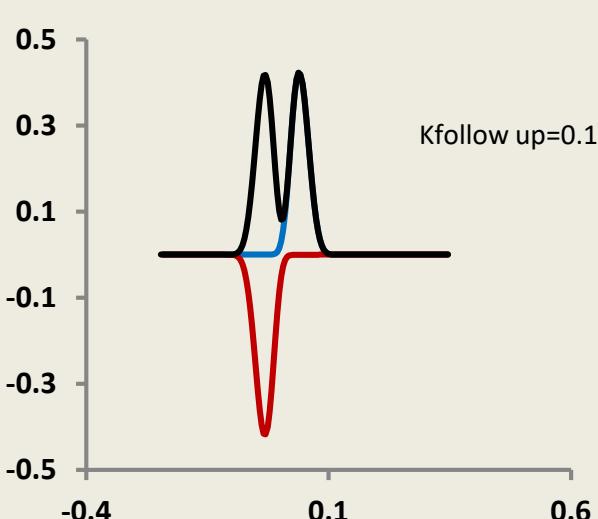
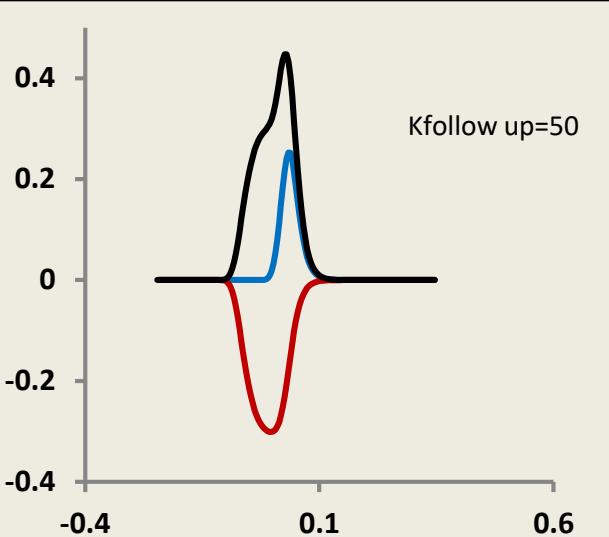
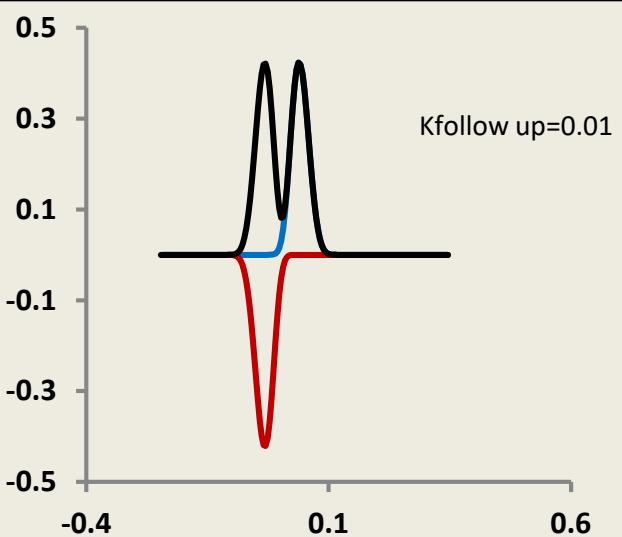
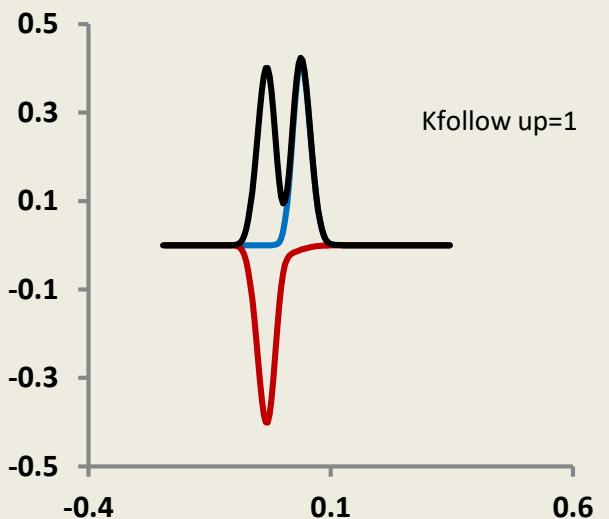
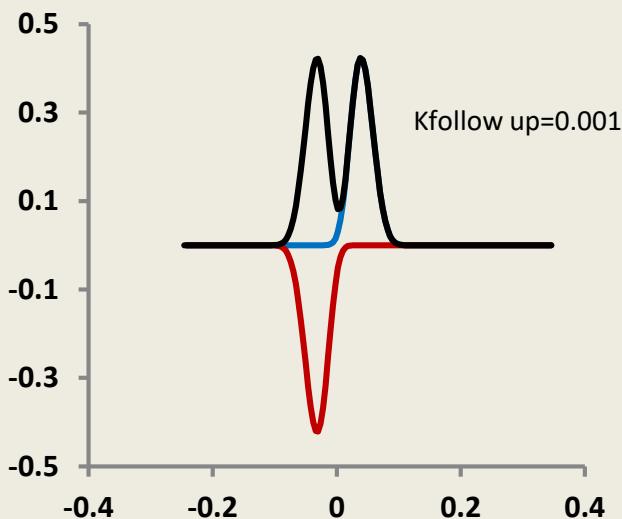
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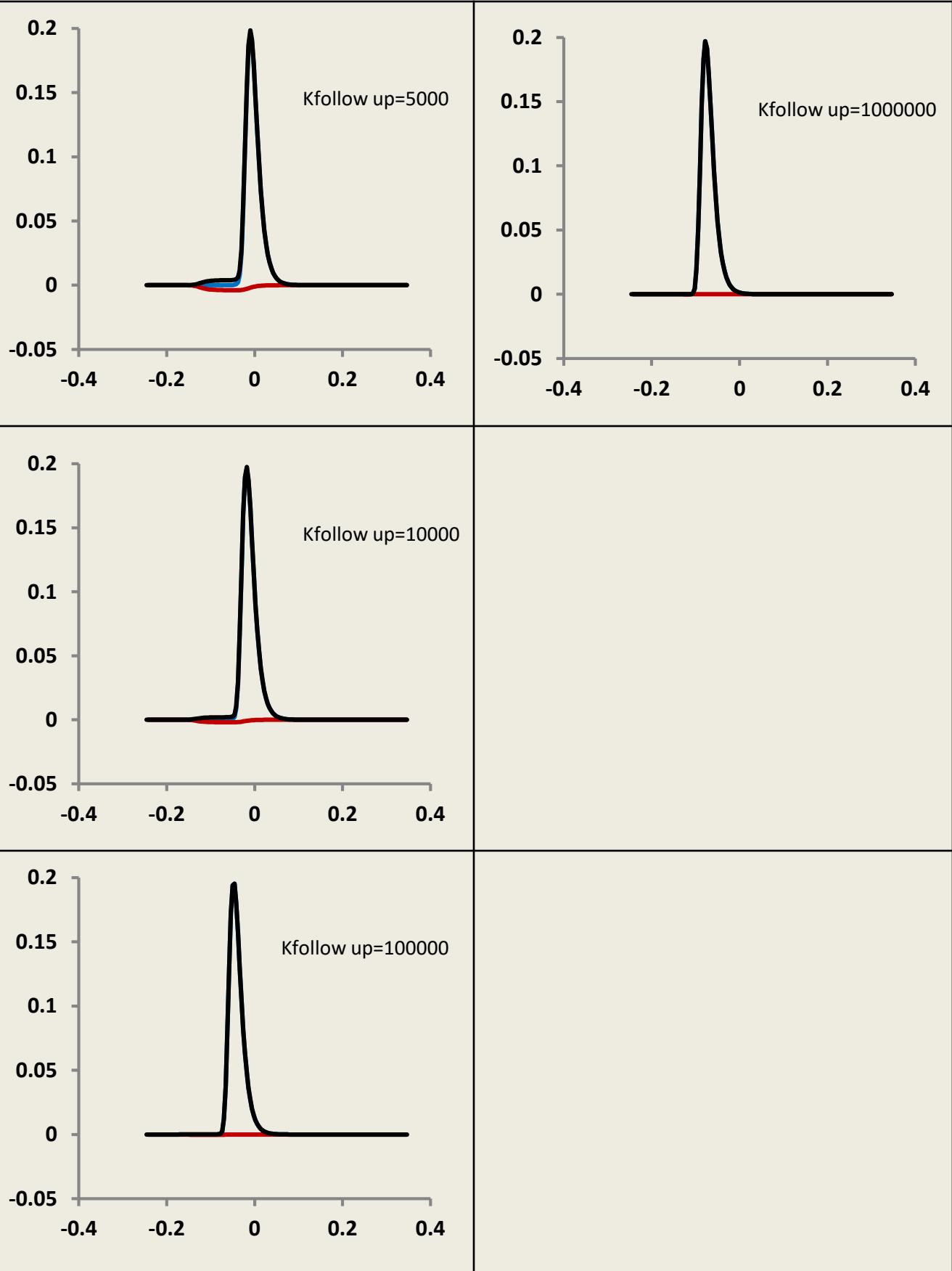
0.1

1

50

1000





$$\begin{array}{c} K_S=10^{1,25} \\ U=1 \end{array}$$

$$b=0.001$$

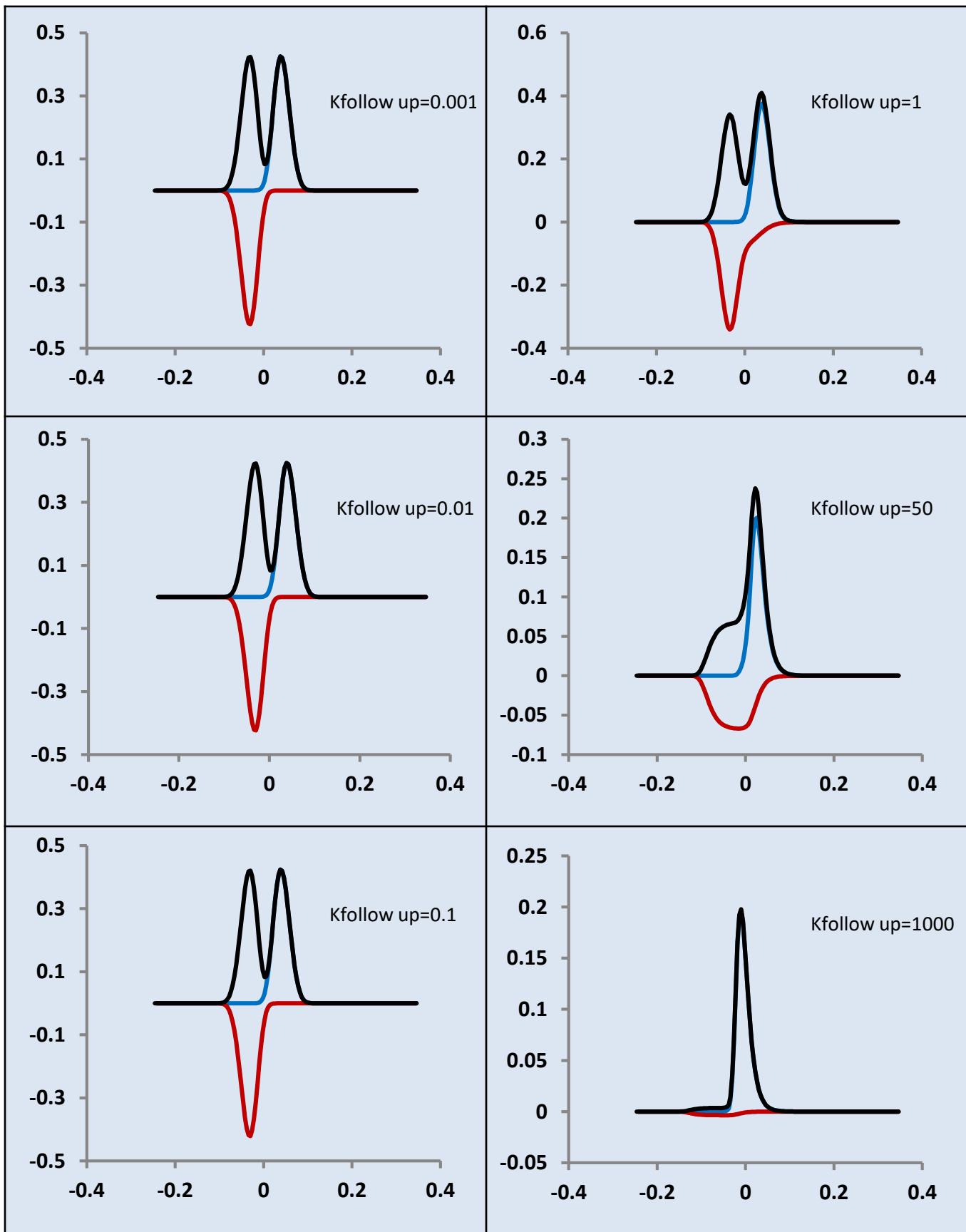
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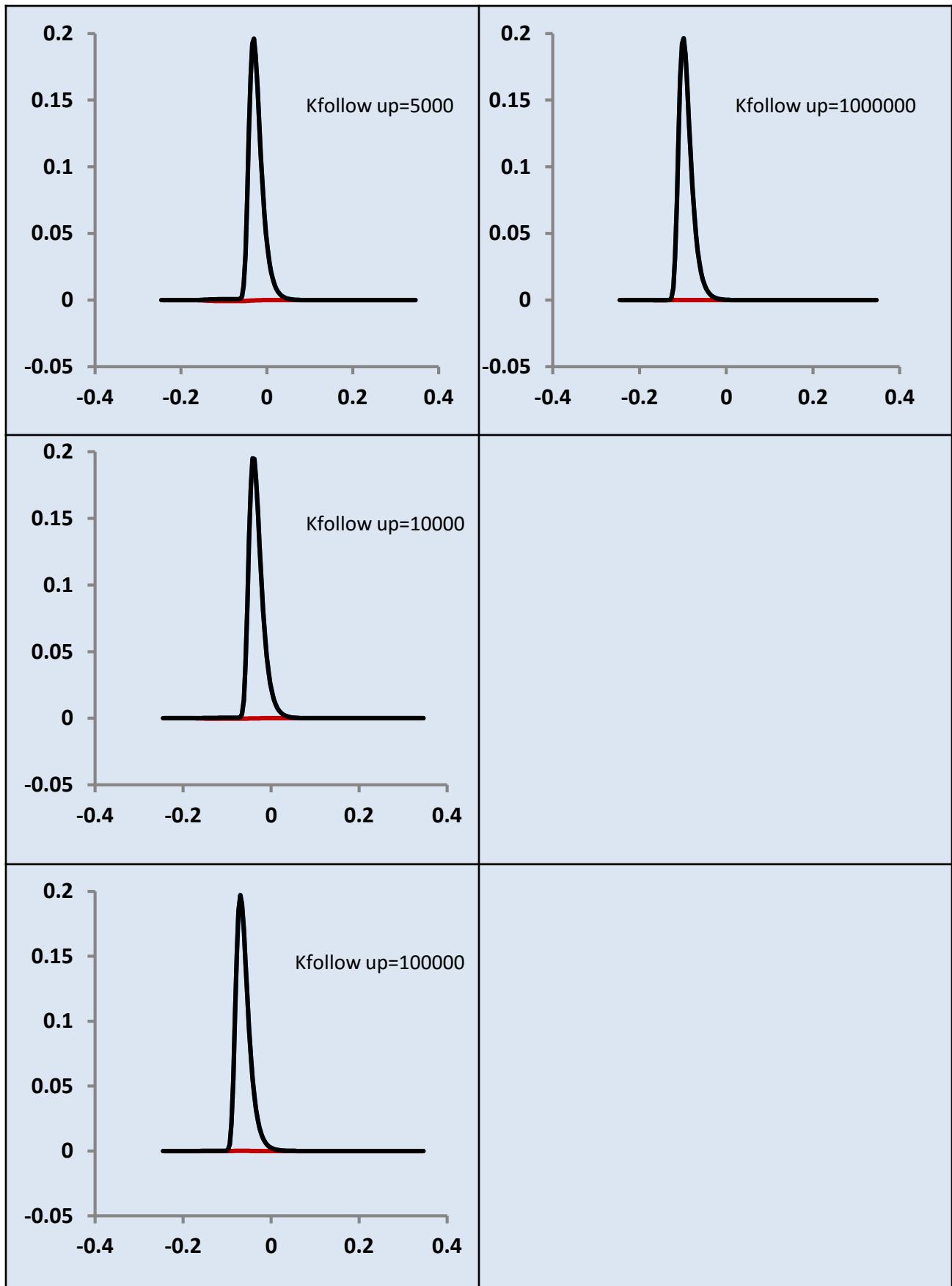
$$0.1\,$$

$$1\,$$

$$50\,$$

$$1000$$





$K_S = 10^{1,25}$

$U = 0.01$

$b = 0.001$

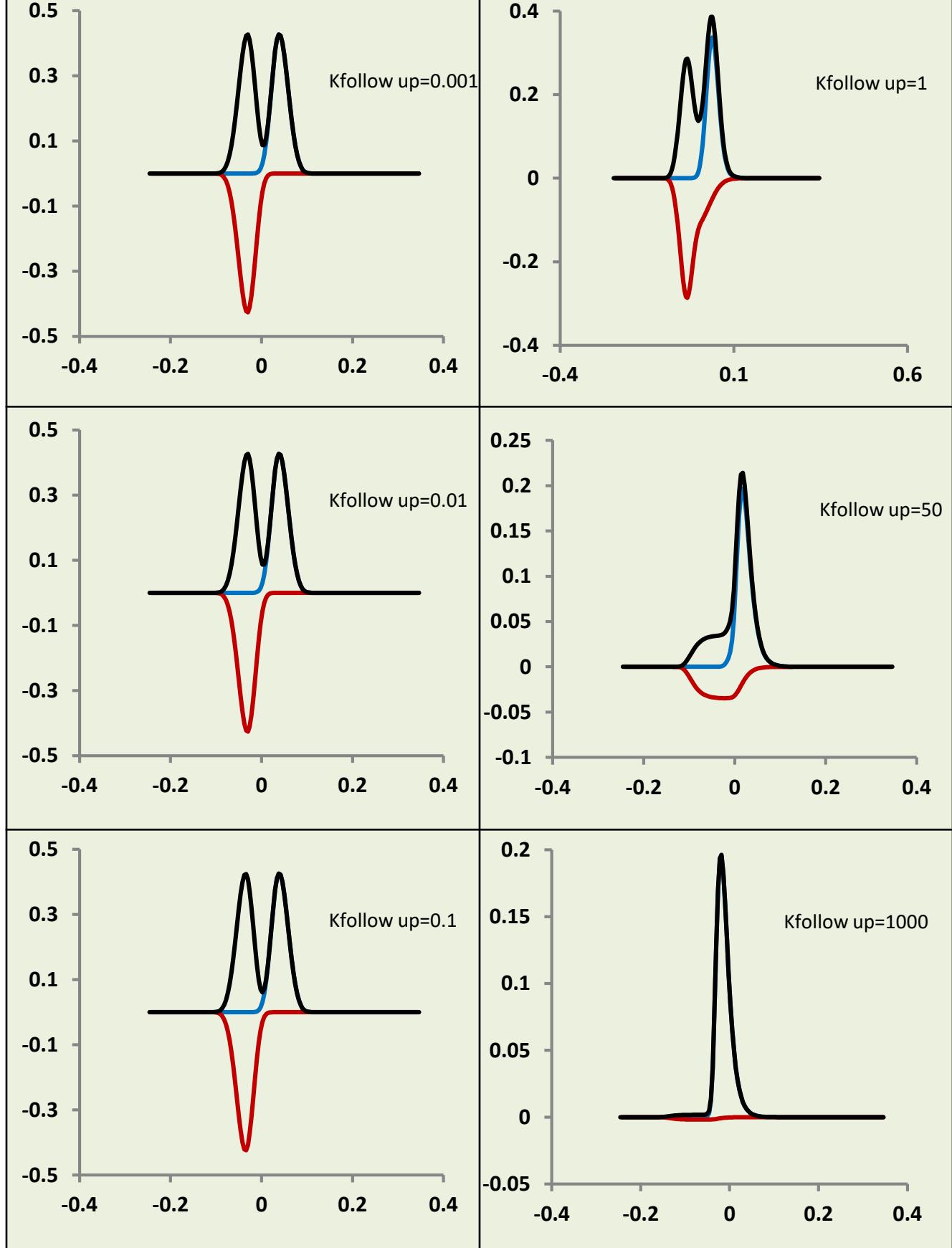
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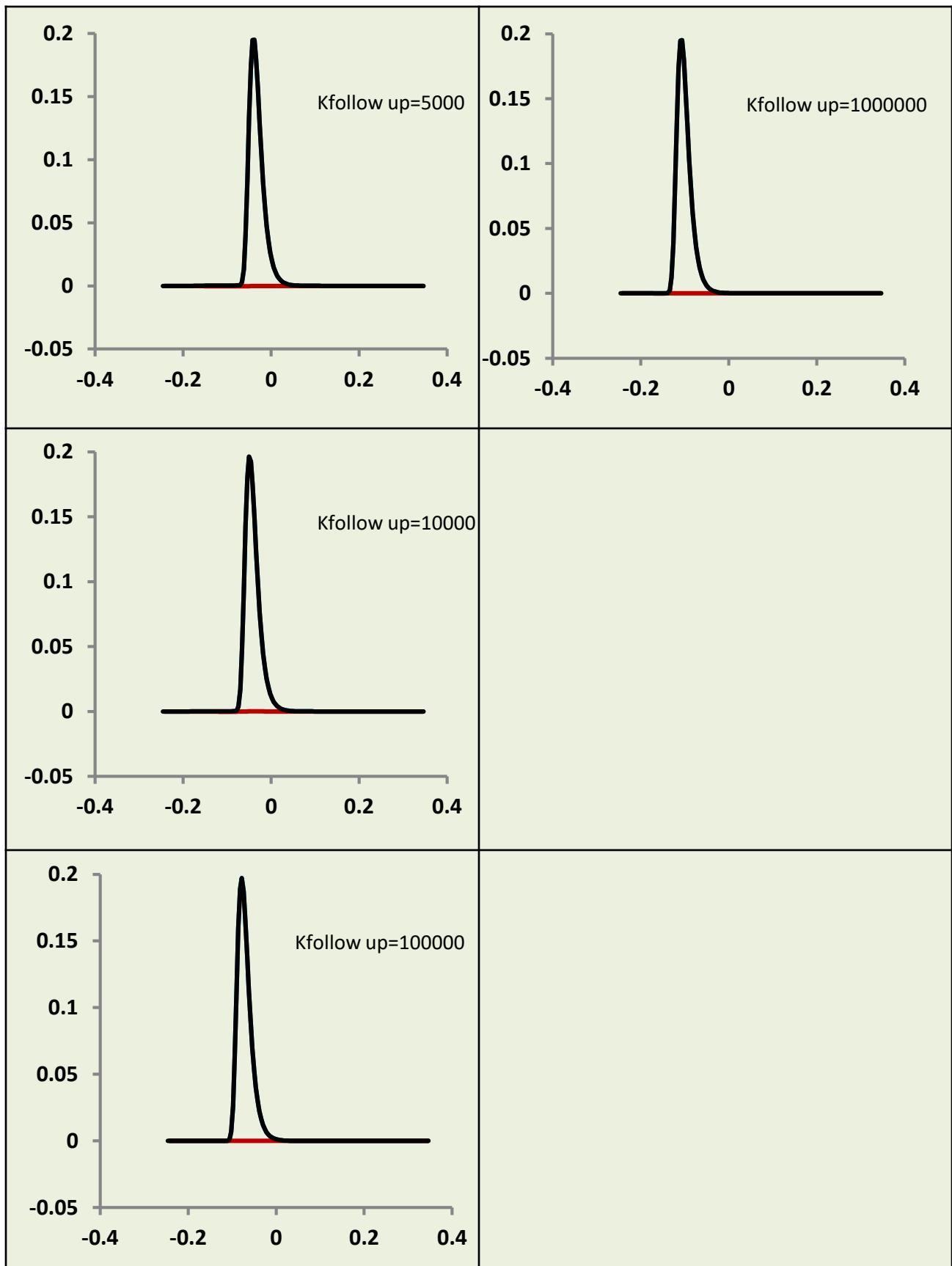
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1

50

1000





$$K_s = 10^0$$
$$U = 1000$$

b=0.001

0.01

0.1

1

50

1000

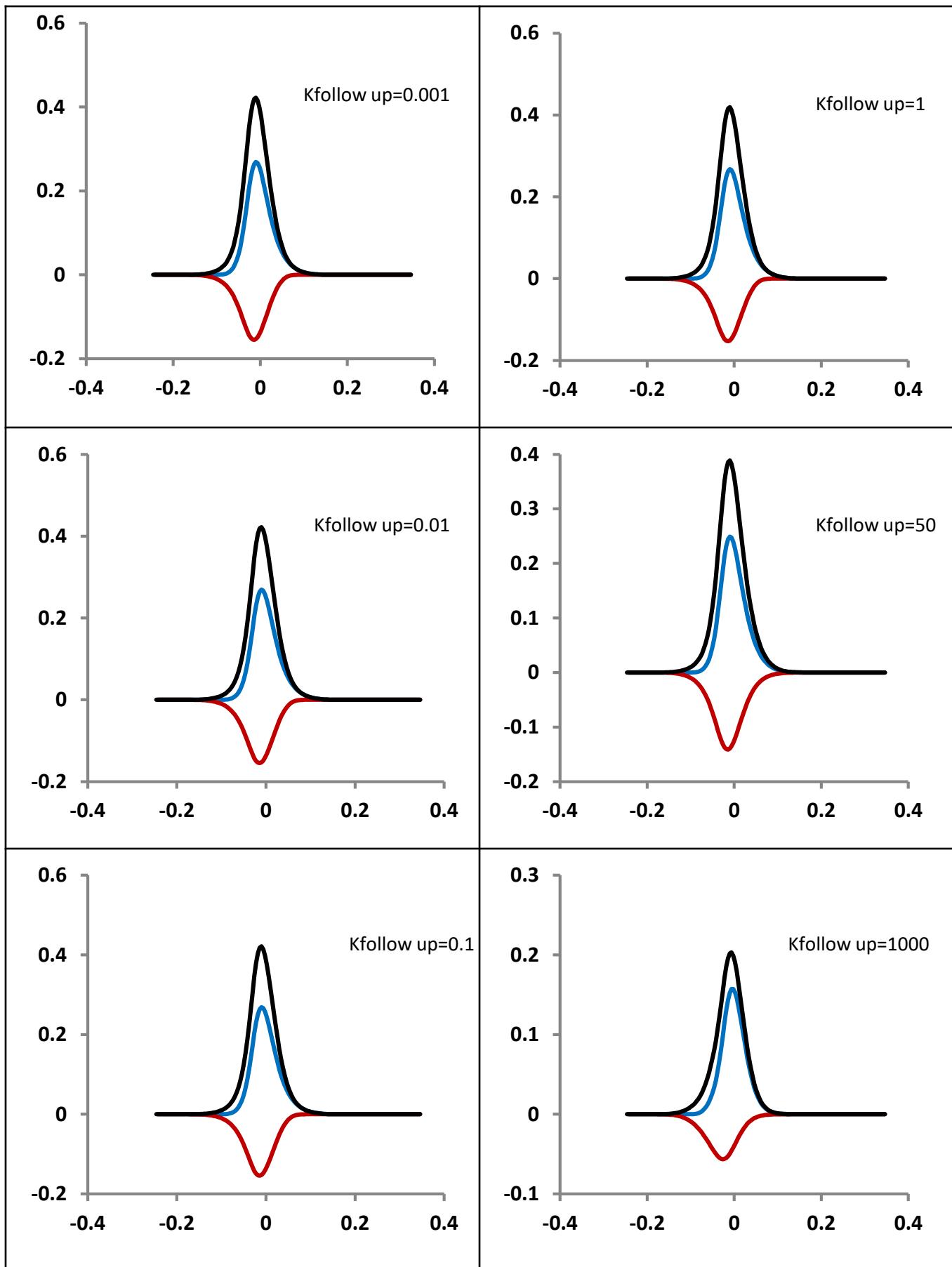
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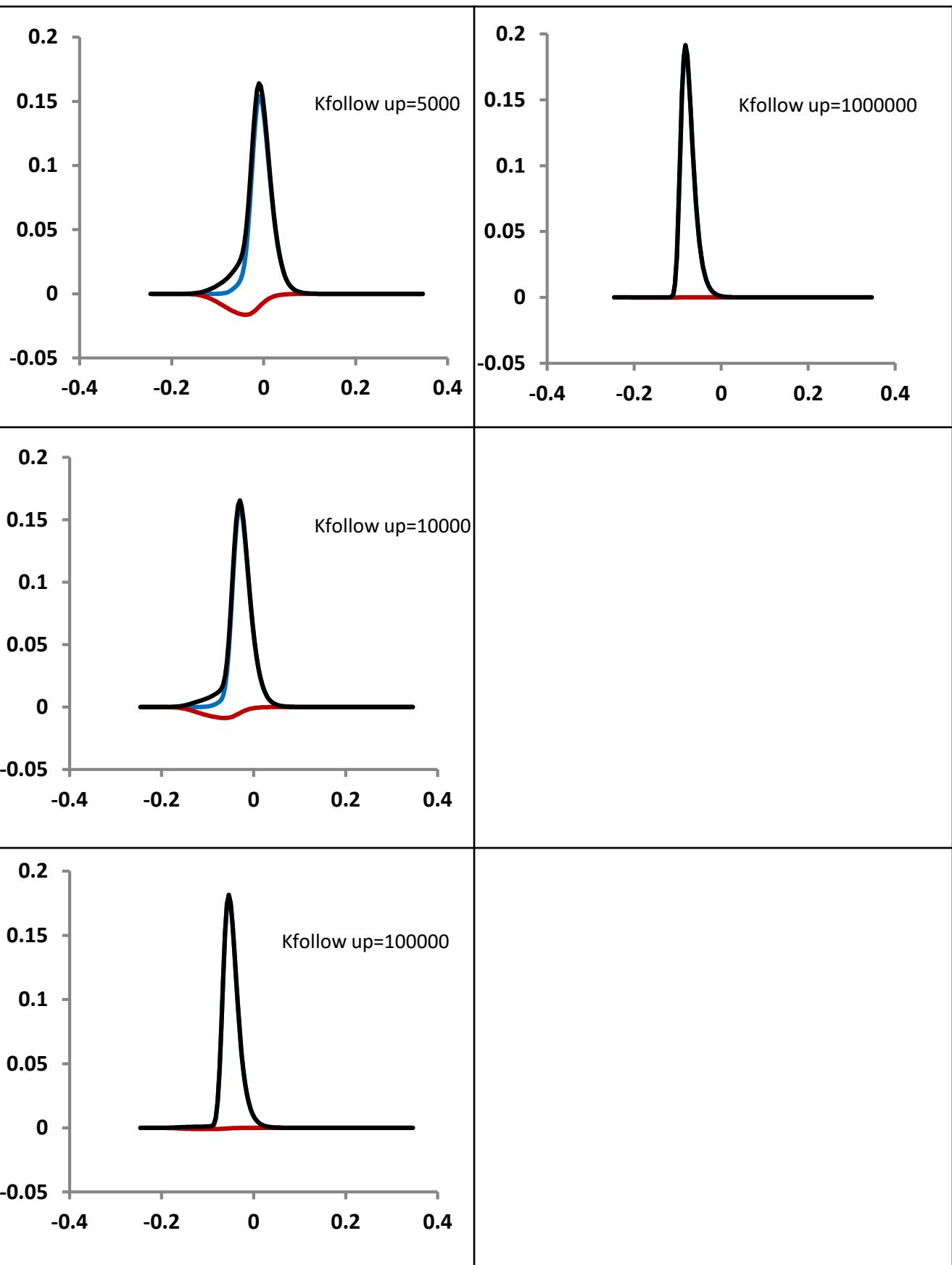
10000

100000

1000000

- Epsilon 10^3





$K_s = 10^0$

$U = 1$

$b = 0.001$

0.01

0.1

1

50

1000

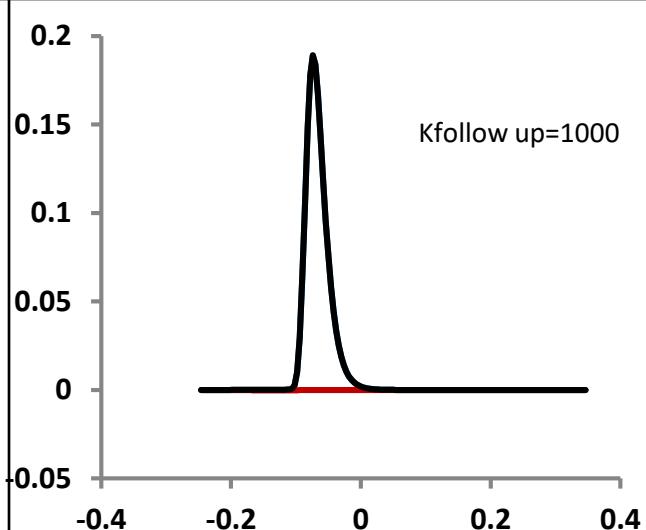
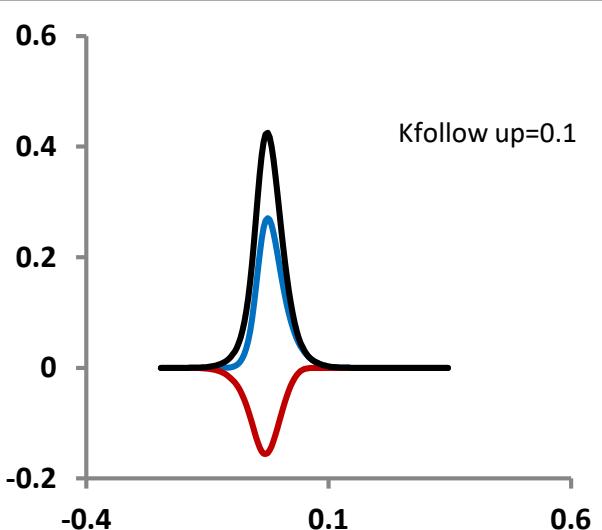
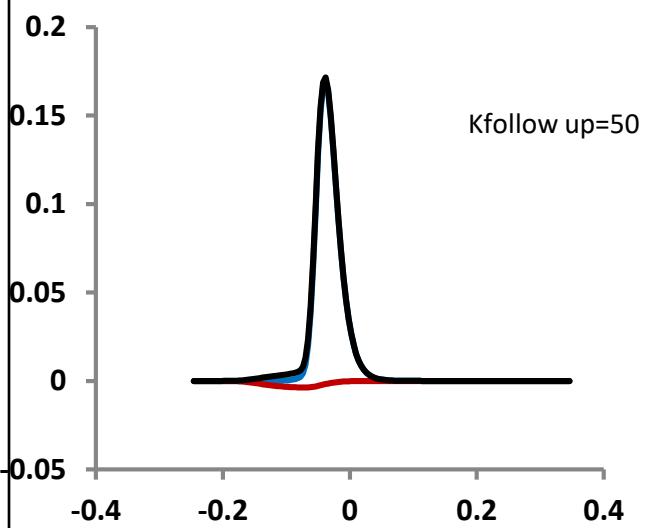
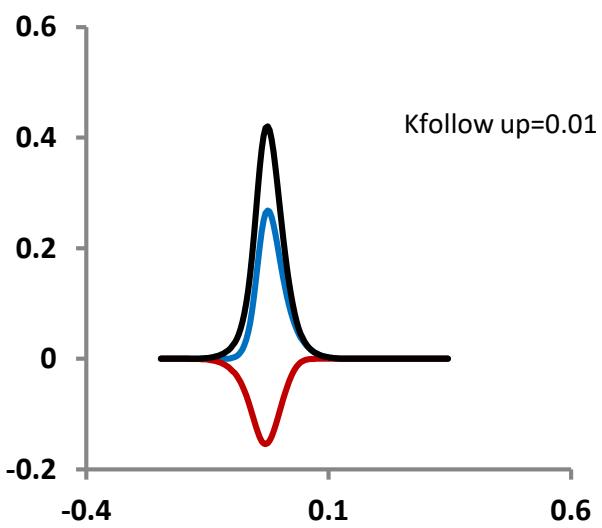
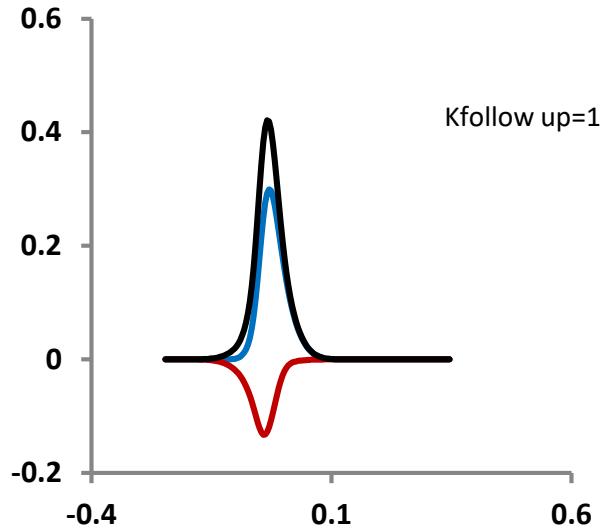
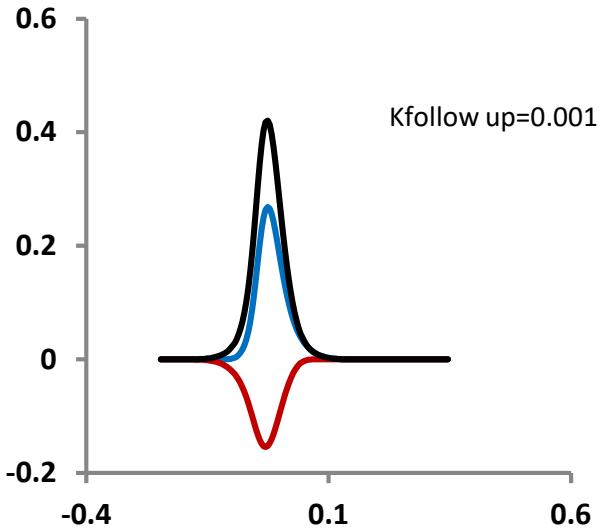
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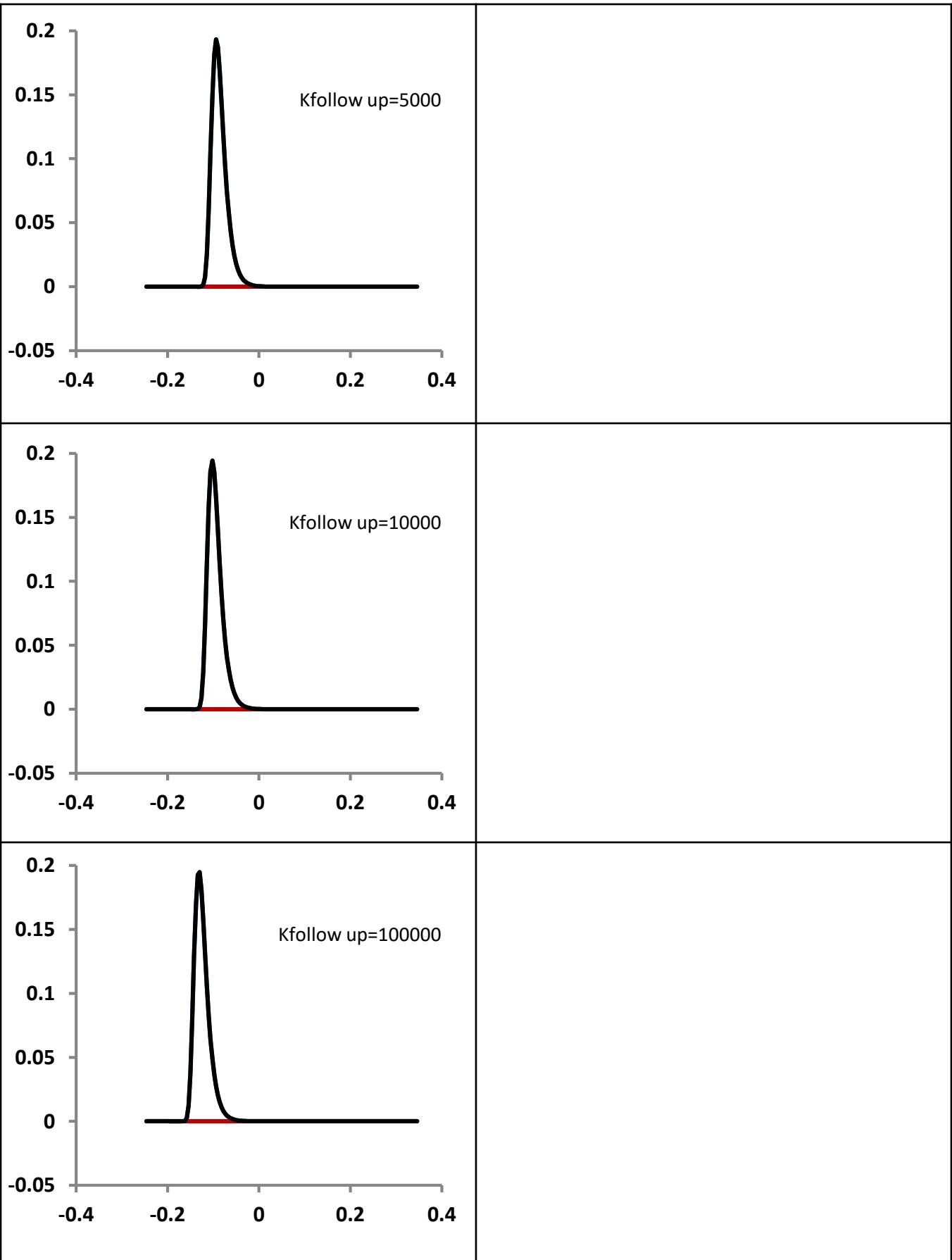
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100000

1000000

- Epsilon 10^3





$$K_s = 10^0$$
$$U = 0.01$$

b=0.001

0.01

0.1

1

50

1000

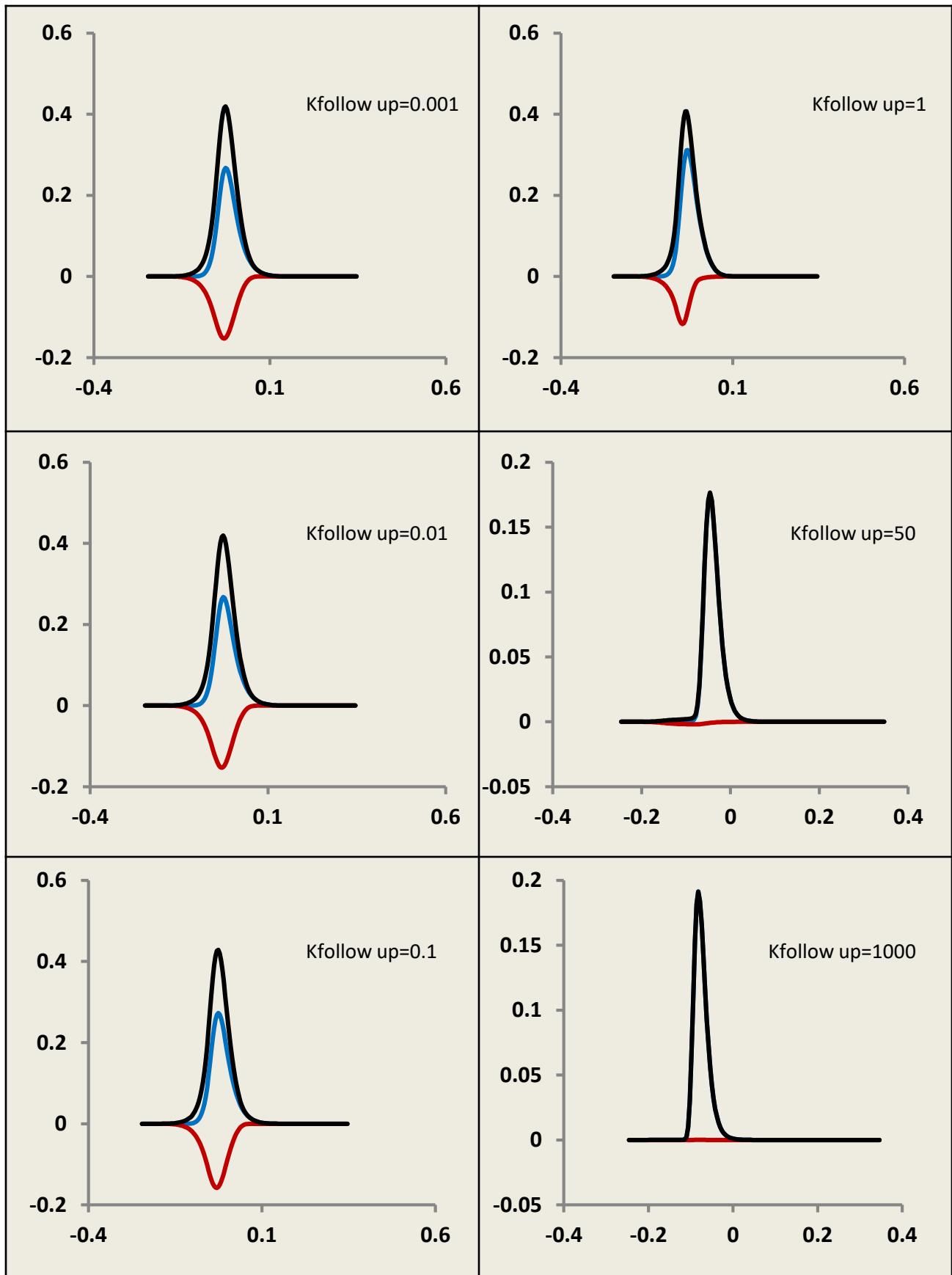
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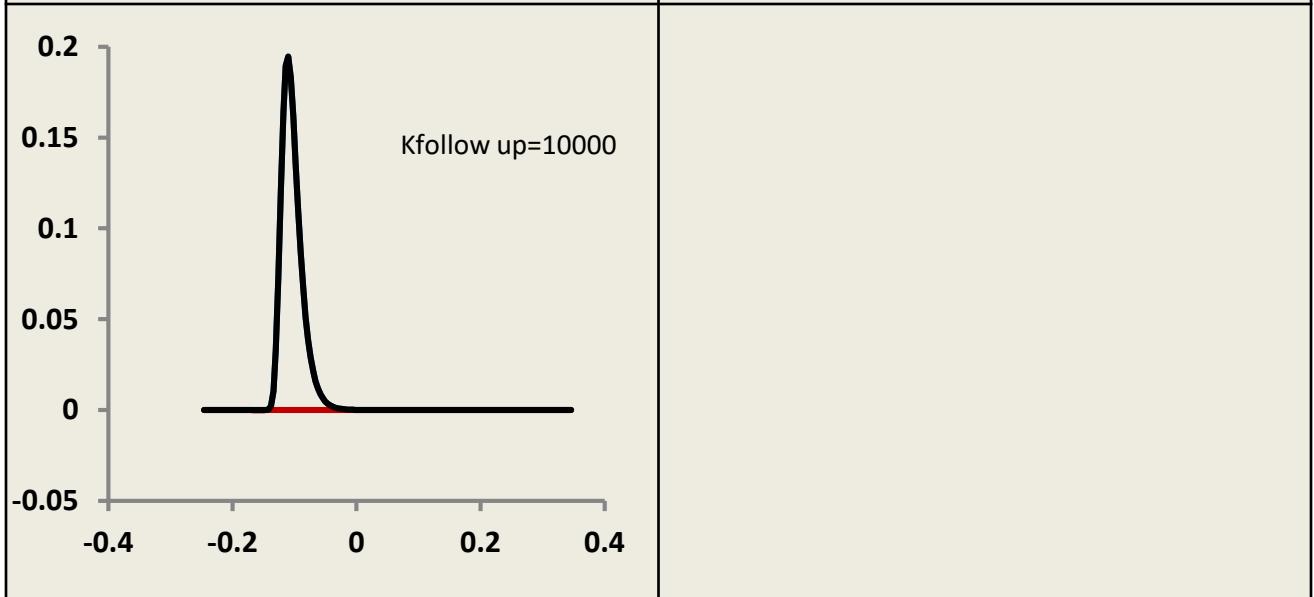
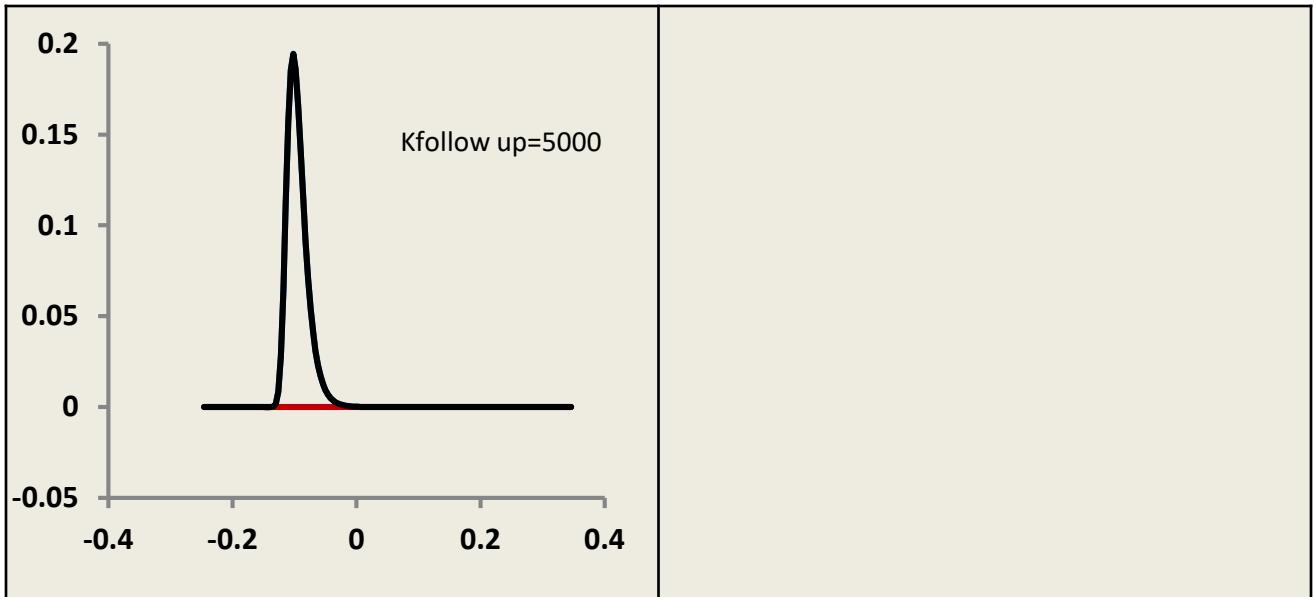
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1000000

- Epsilon 10^3





$$K_s = 10^0$$
$$U = 1000$$

b=0.001

0.01

0.1

1

50

1000

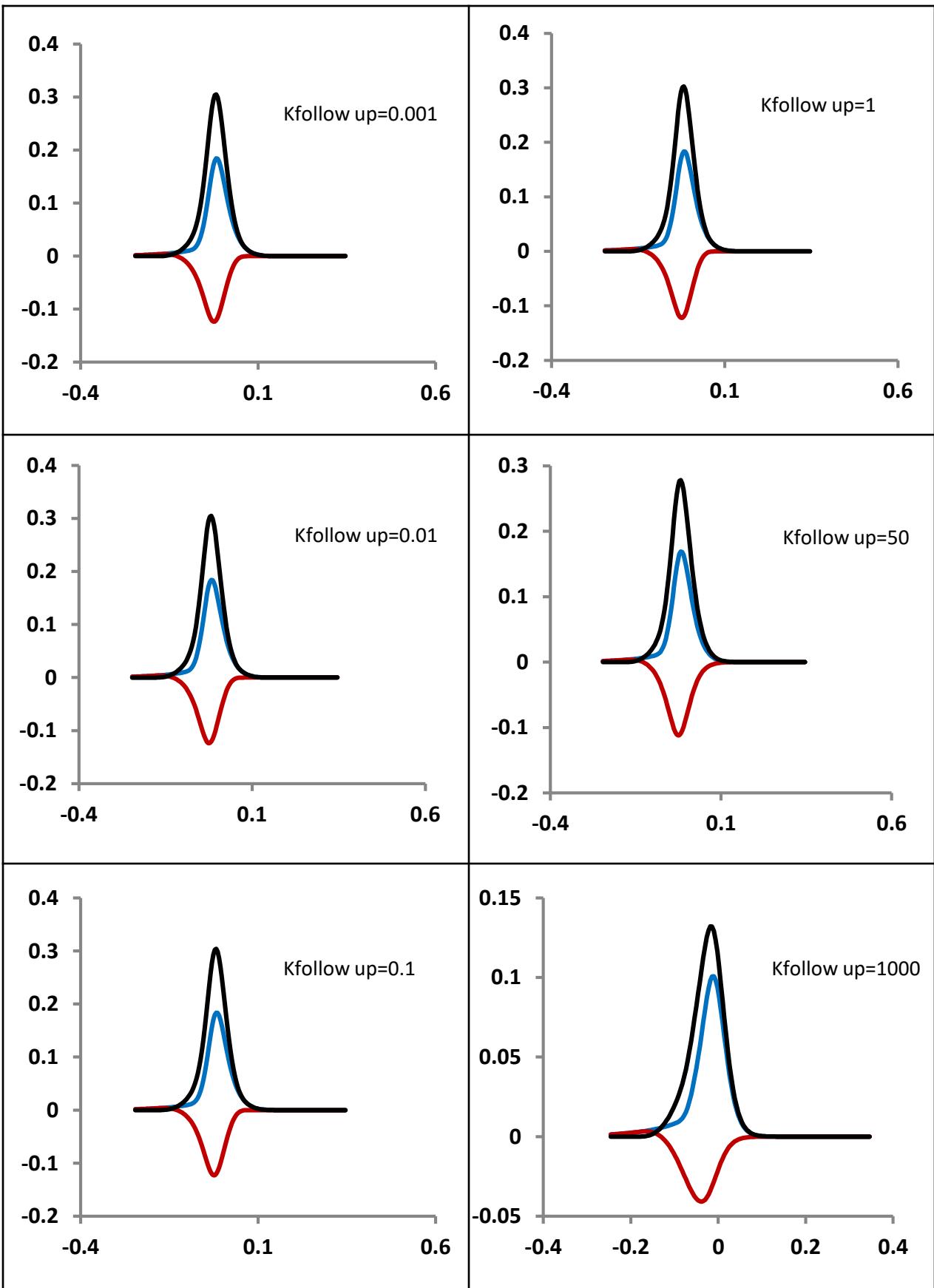
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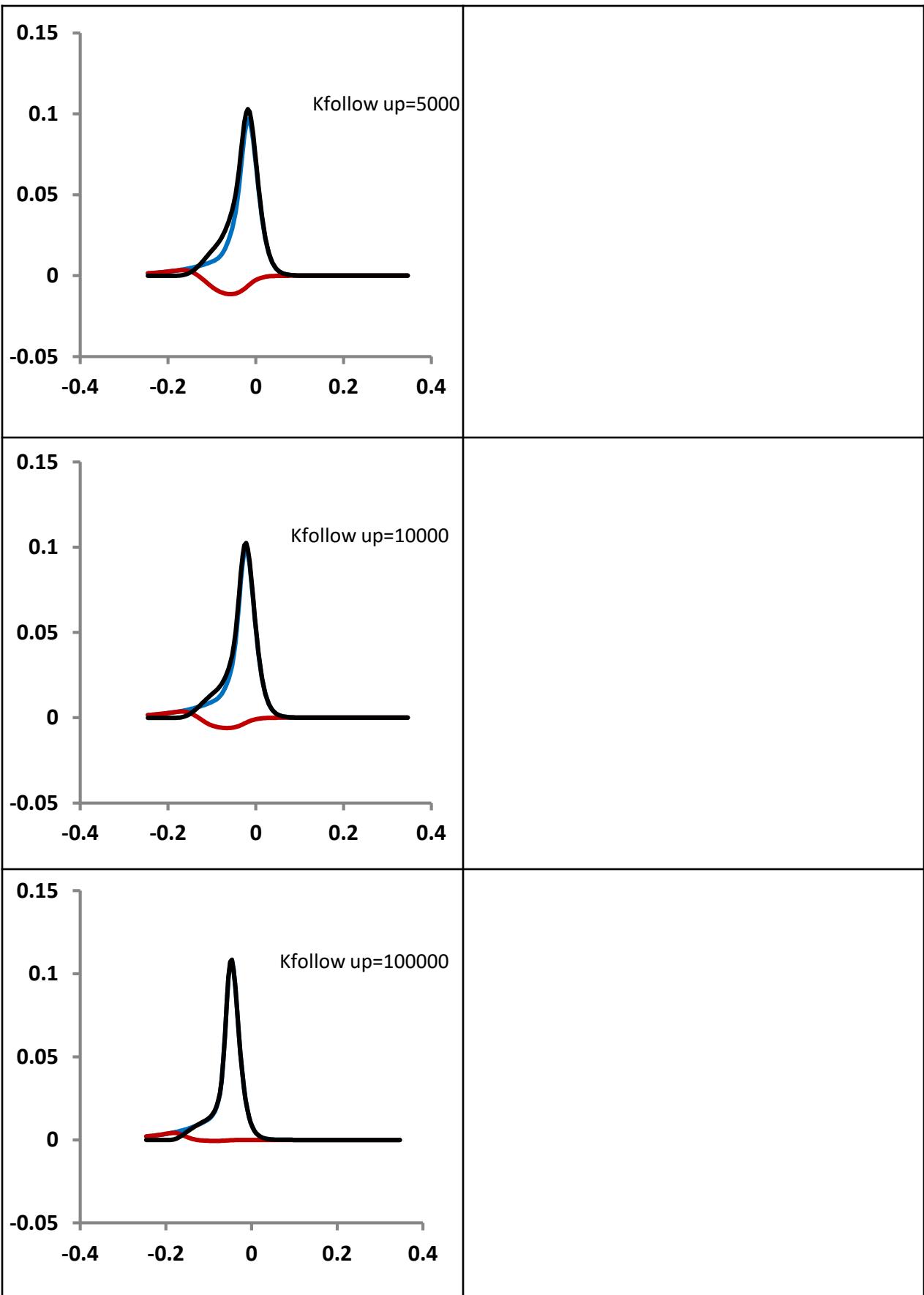
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- Epsilon 10^0





$K_s = 10^0$

$U = 1$

$b = 0.001$

0.01

0.1

1

50

1000

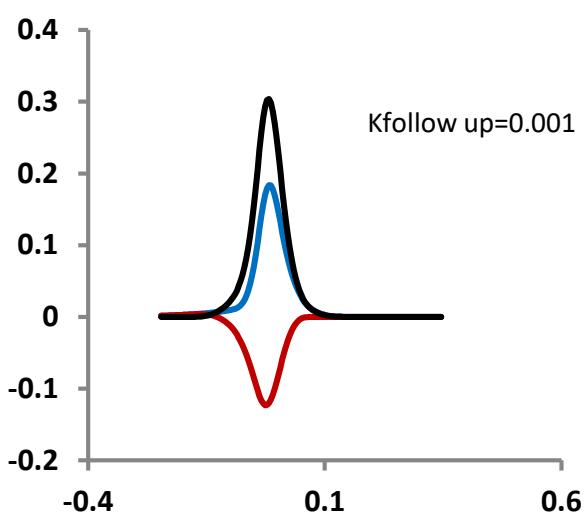
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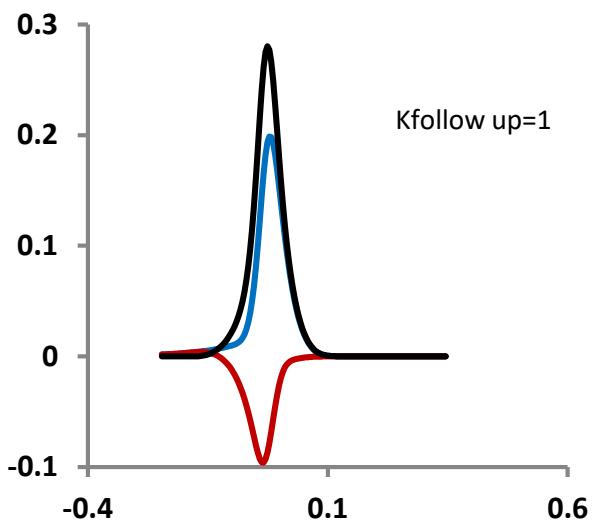
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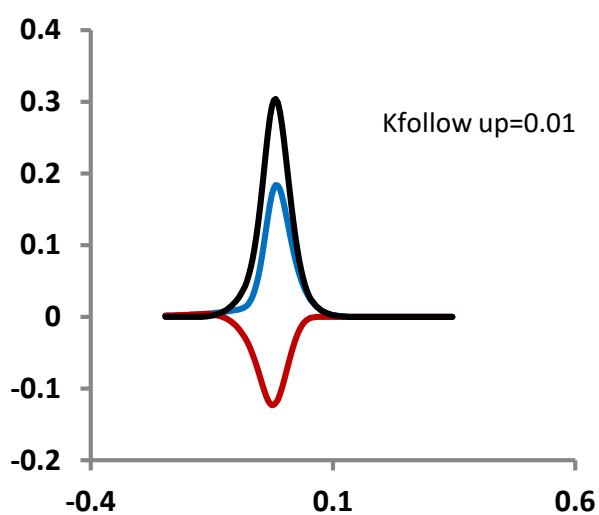
- Epsilon 10^0



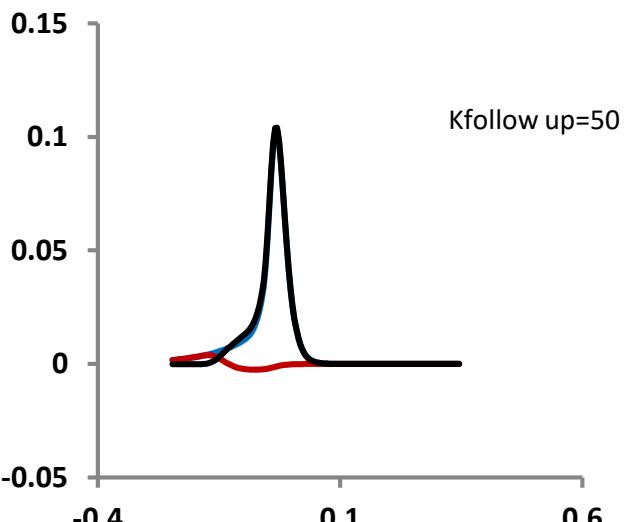
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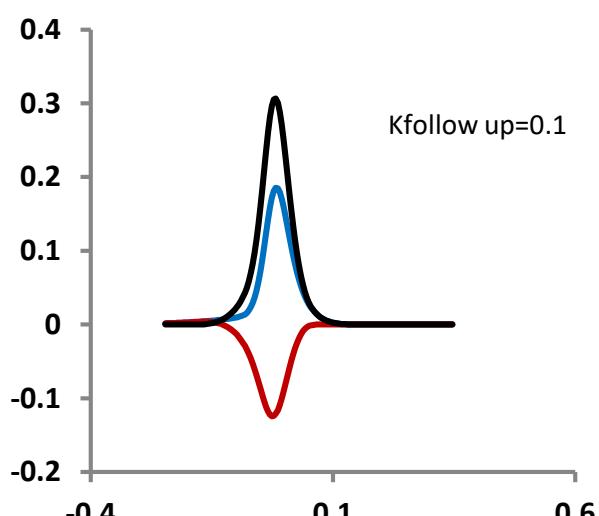
Kfollow up=1



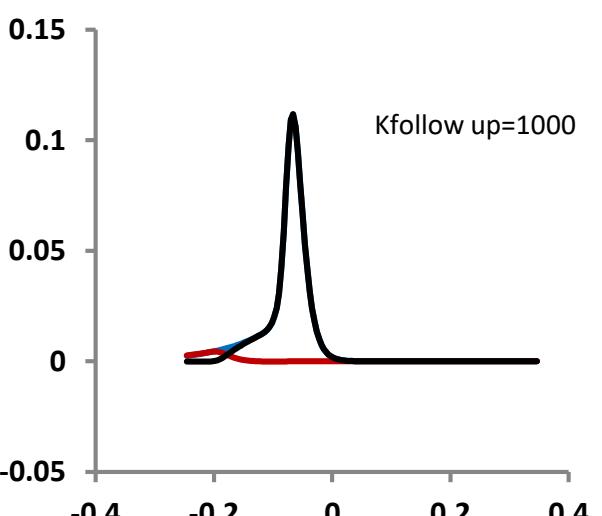
Kfollow up=0.01



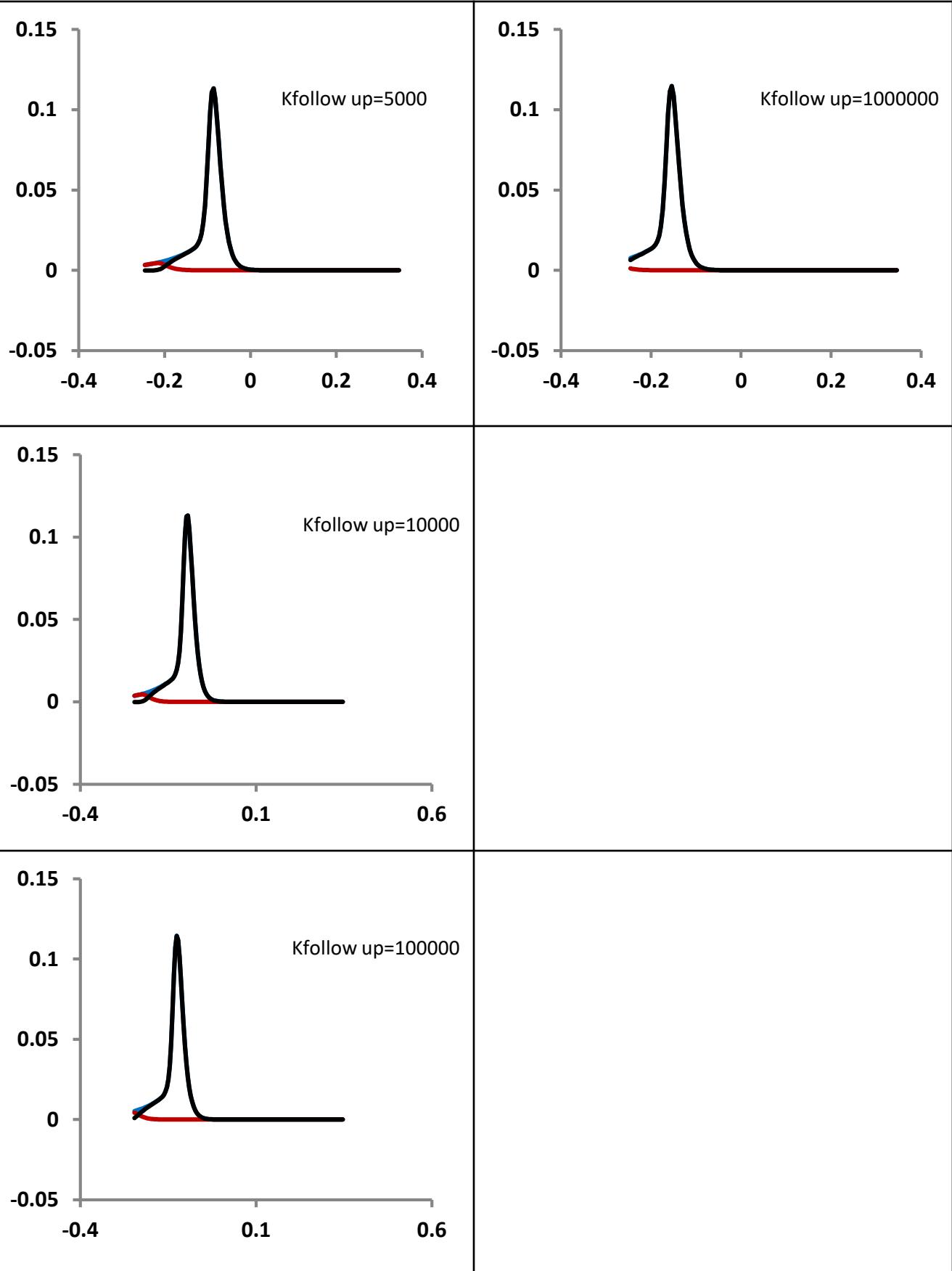
Kfollow up=50



Kfollow up=0.1



Kfollow up=1000



$$K_s = 10^0$$
$$U = 0.01$$

b=0.001

0.01

0.1

1

50

1000

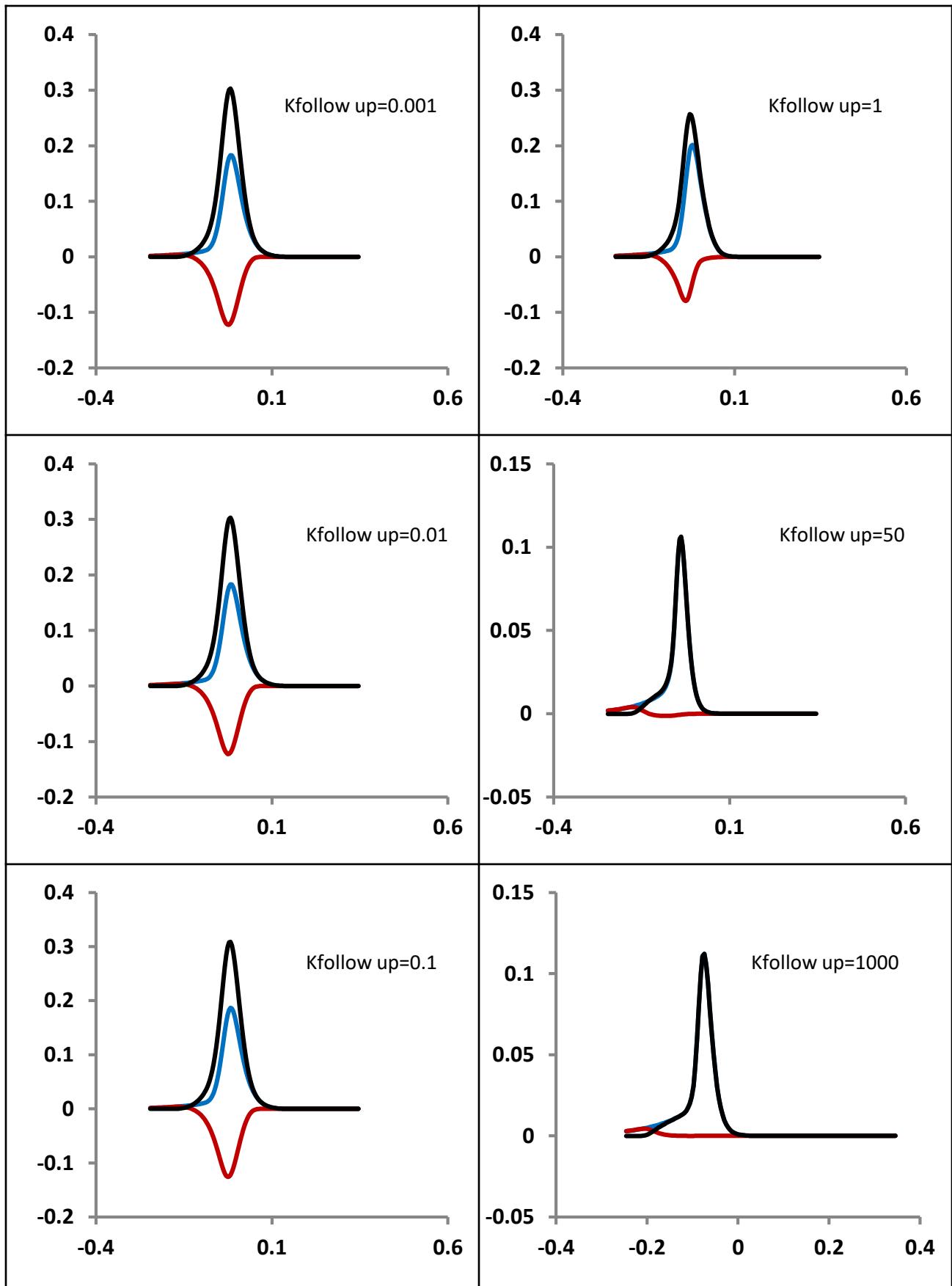
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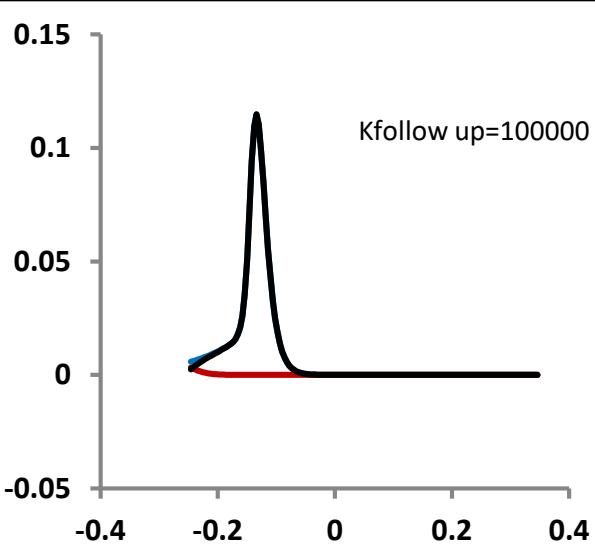
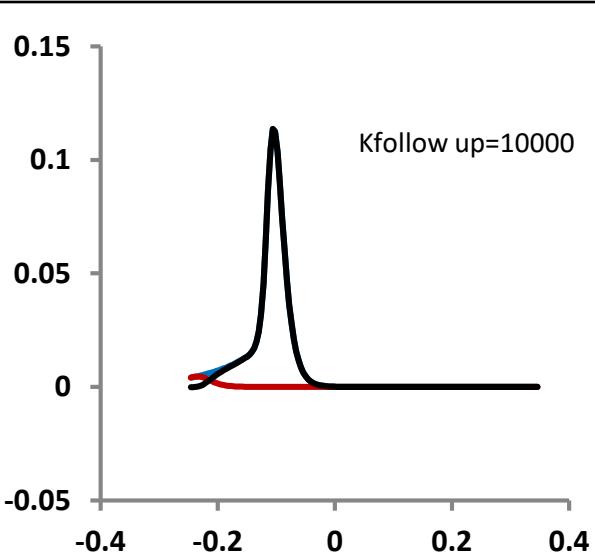
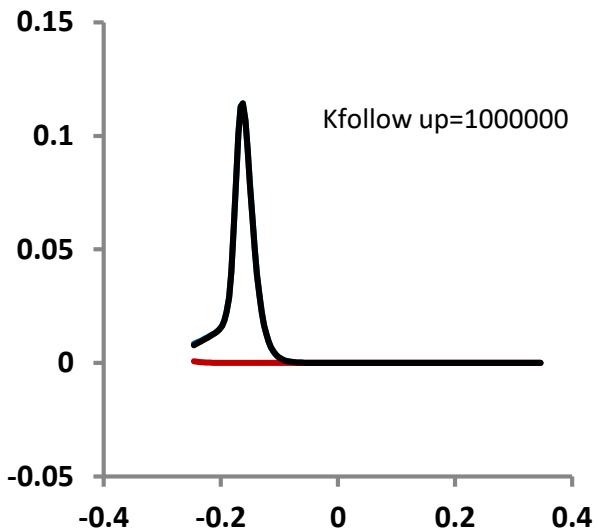
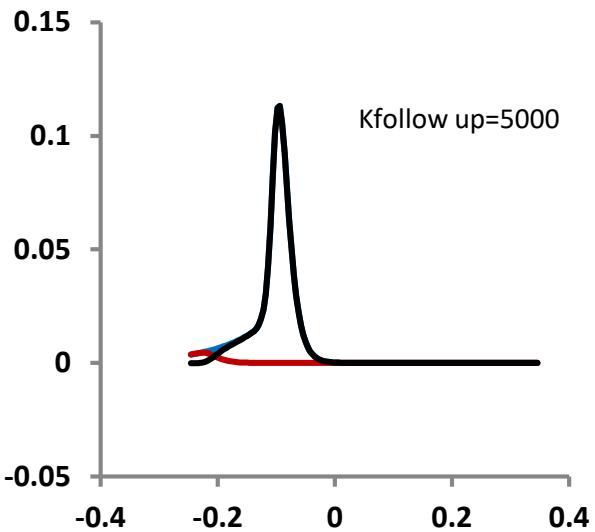
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100000

1000000

- Epsilon 10^0





$$K_s = 10^0$$
$$U = 1000$$

b=0.001

0.01

0.1

1

50

1000

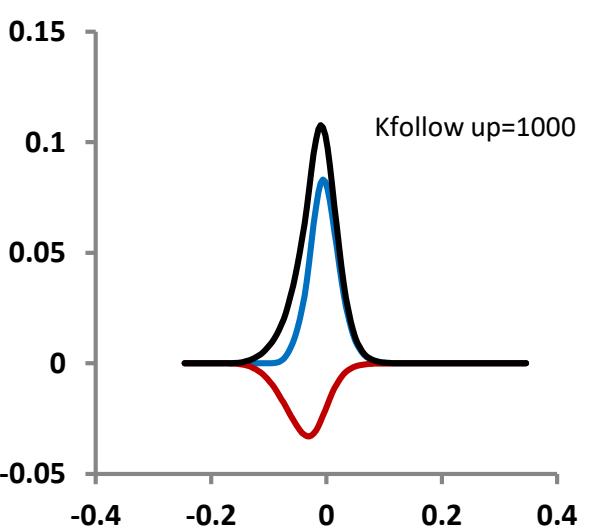
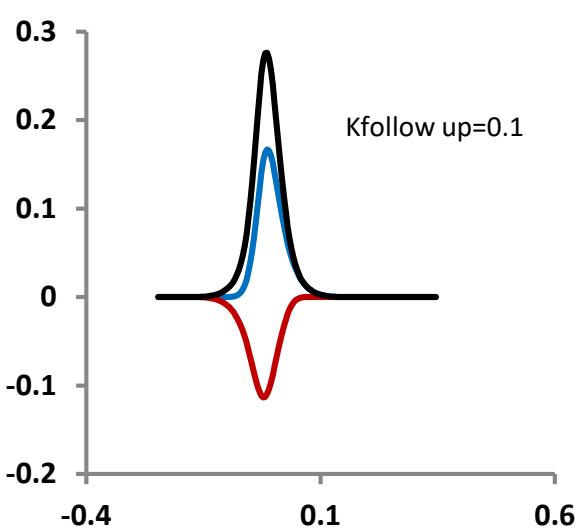
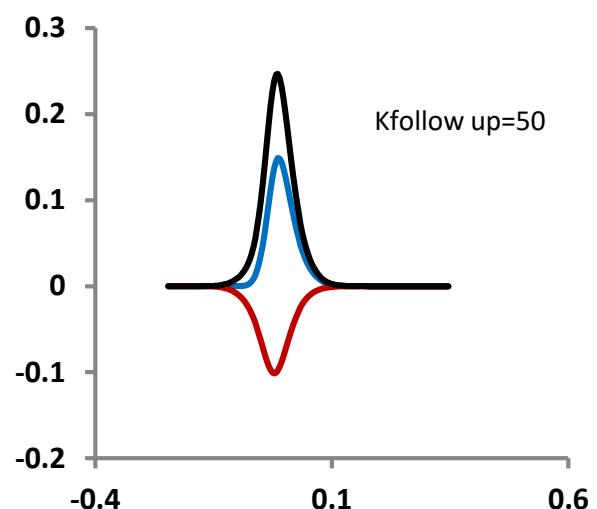
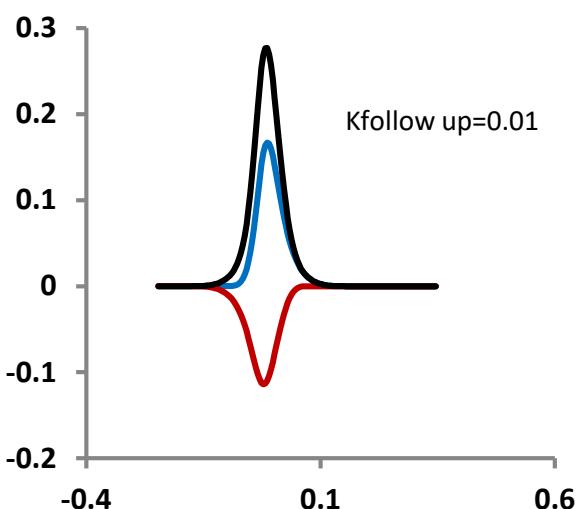
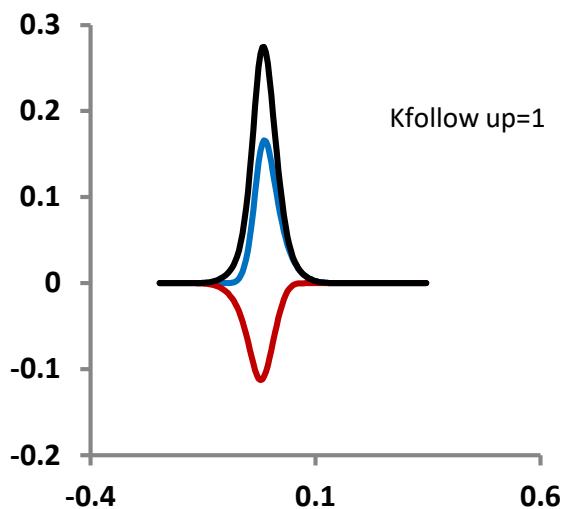
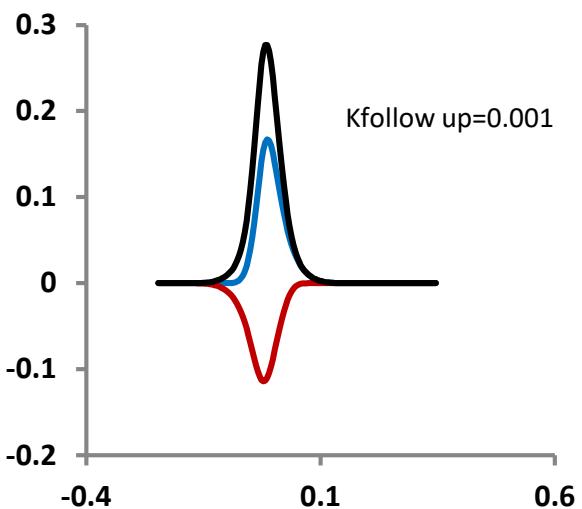
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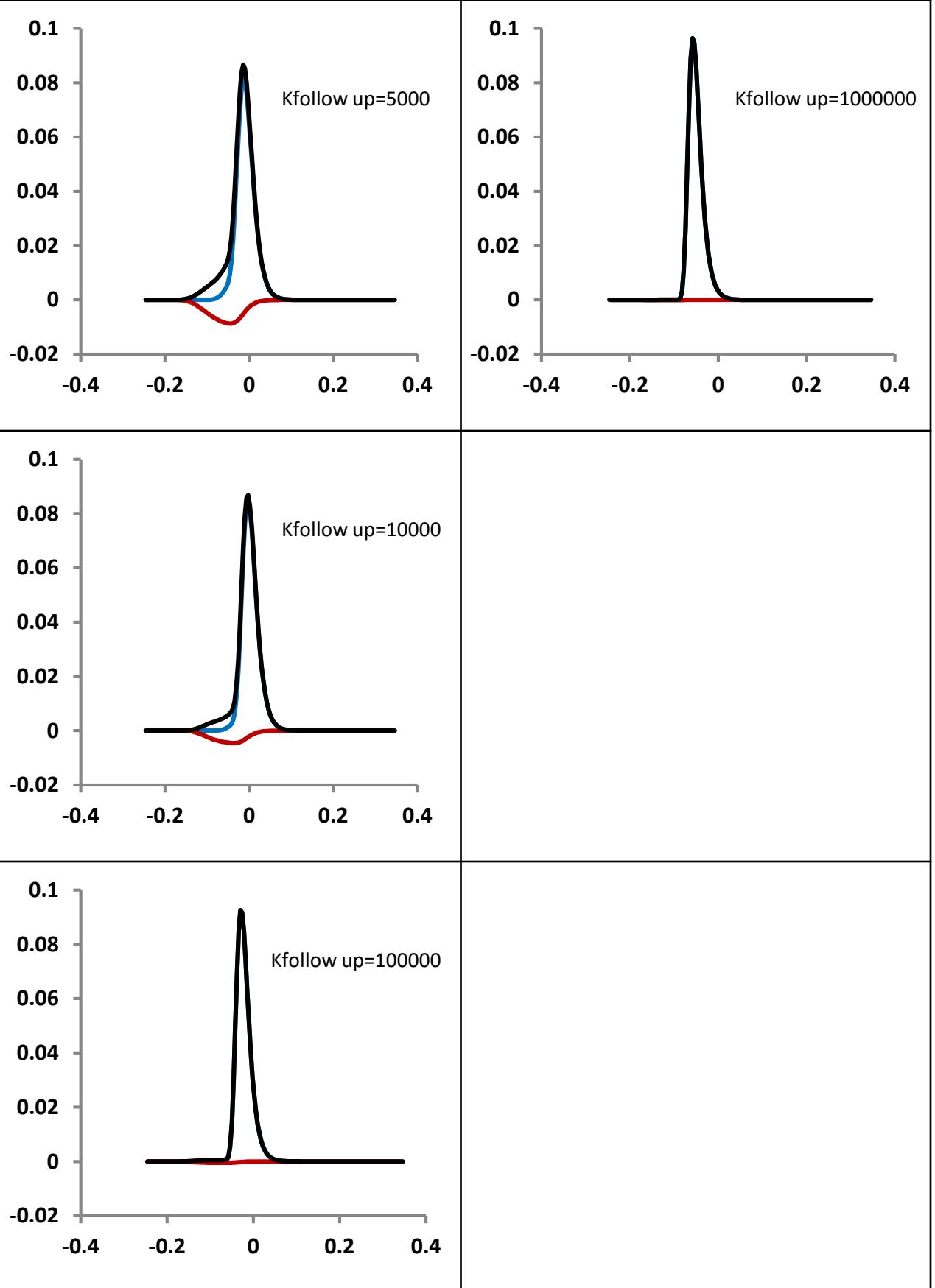
10000

100000

1000000

- Epsilon 10^{-3}





$$K_s = 10^0$$
$$U = 1$$

b=0.001

0.01

0.1

1

50

1000

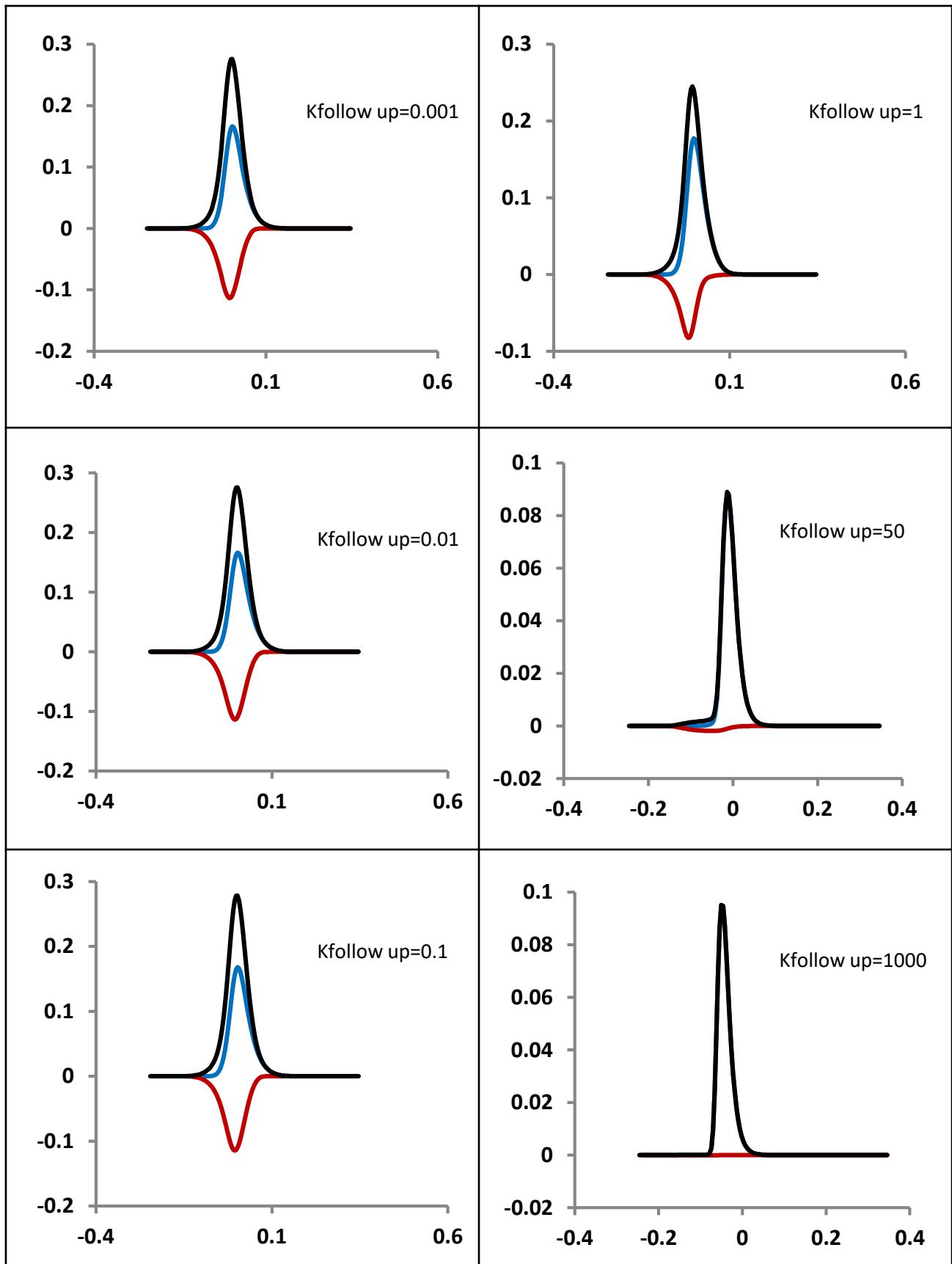
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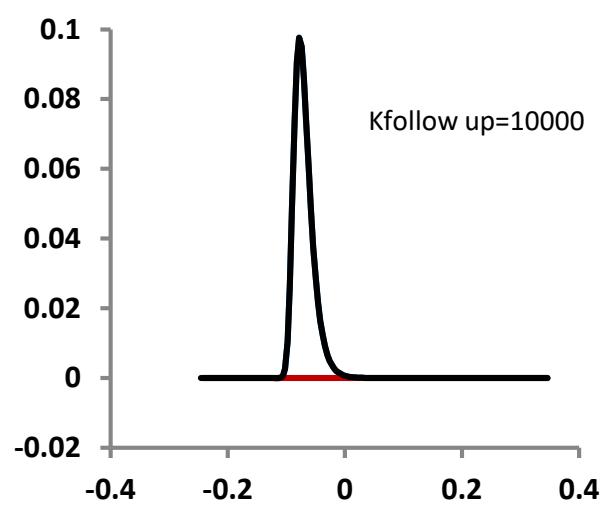
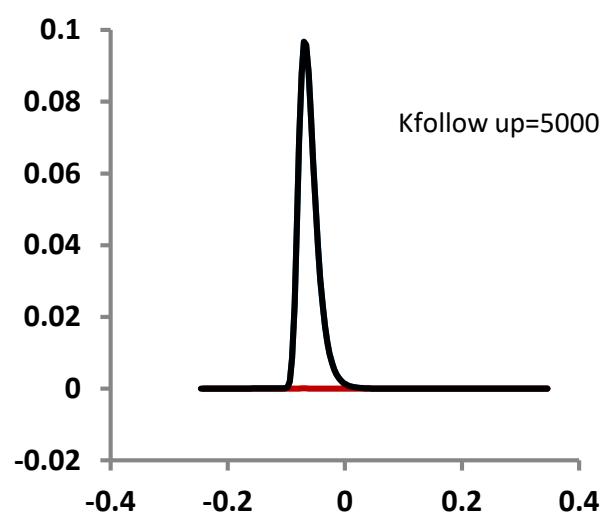
10000

100000

1000000

- Epsilon 10^{-3}





$$K_s = 10^0$$
$$U = 0.01$$

b=0.001

0.01

0.1

1

50

1000

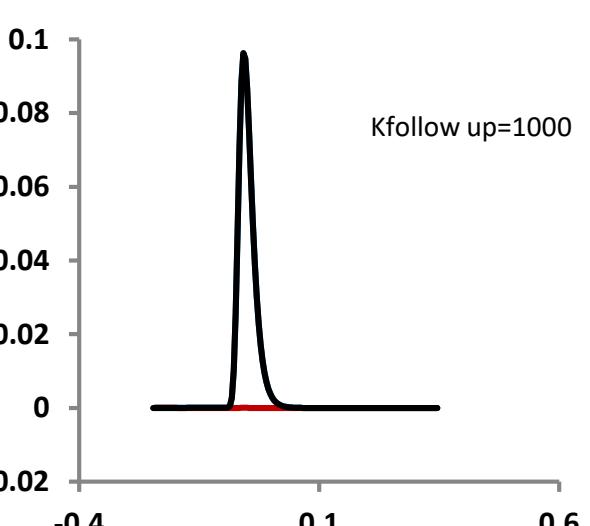
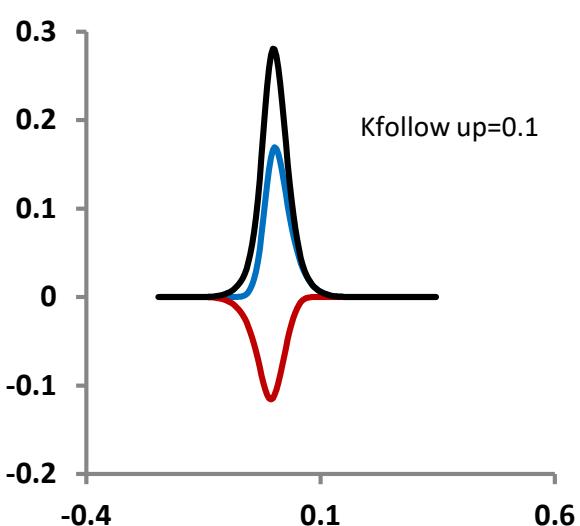
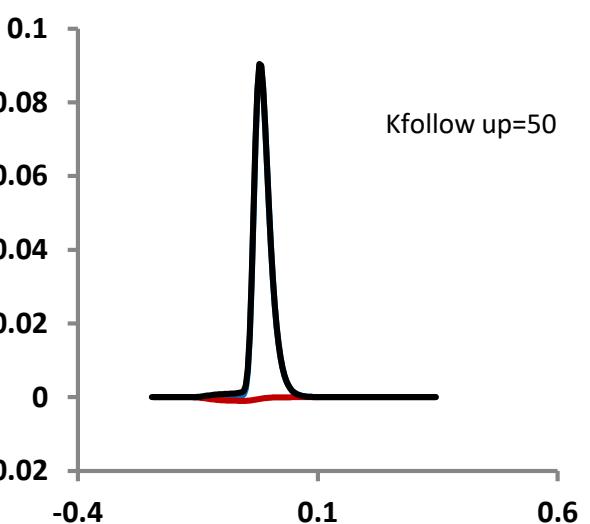
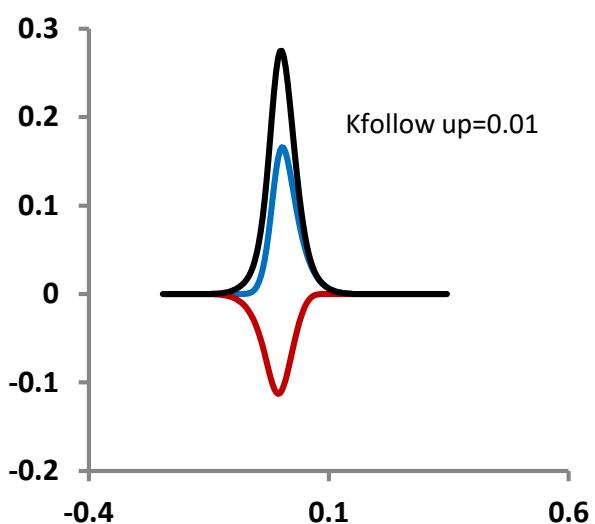
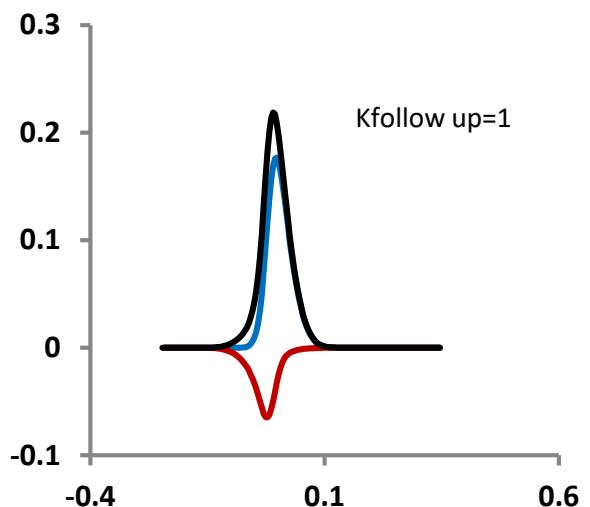
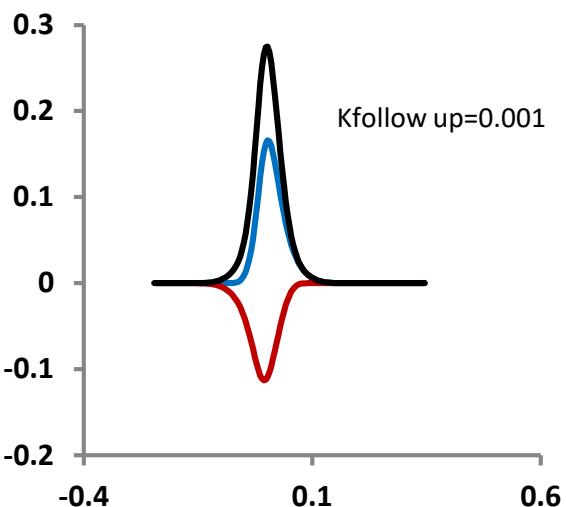
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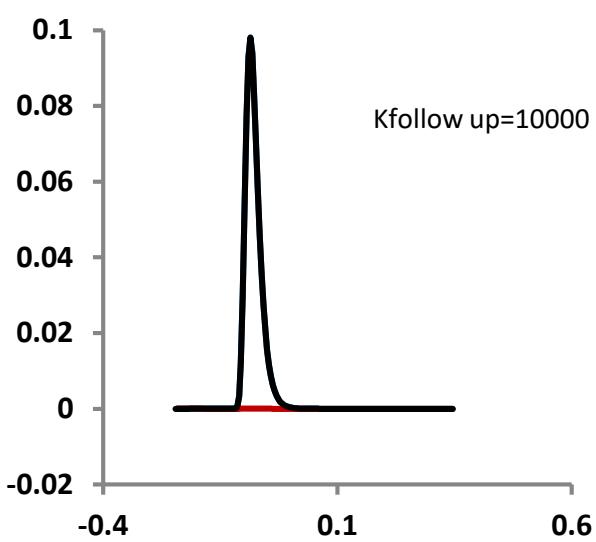
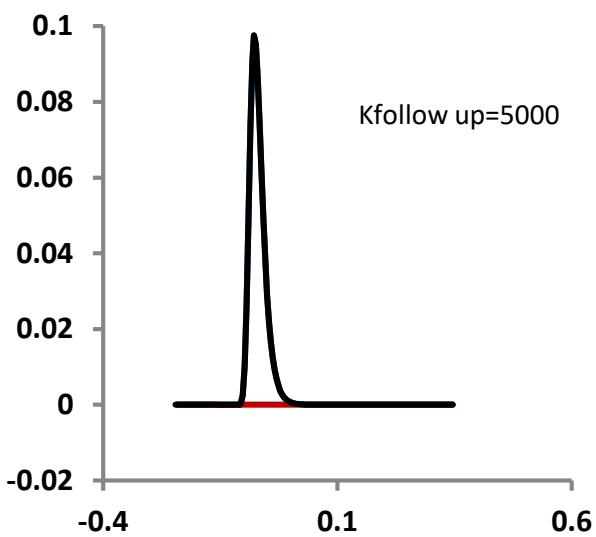
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1000000

- Epsilon 10^{-3}





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13. R. Gulaboski, V. Markovski, and Z. Jihe, Redox chemistry of coenzyme Q—a short overview of the voltammetric features, J. Solid State Electrochem.,20 (2016) 3229-3238.

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15. V. Mirceski, D. Guzijewski and R. Gulaboski, Electrode kinetics from a single square-wave voltammograms, Maced. J. Chem. Chem. Eng. 34 (2015) 1-12. 7. Gulaboski and V. Mirceski, New aspects of the electrochemical-catalytic (EC') mechanism in square-wave voltammetry, Electrochim. Acta, 167 (2015) 219-225.

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