



# EVALUATION OF MULTI-ELEMENT CONTENT AND BIOAVAILABILITY RECORDS FOR VARIOUS PLANT FOOD DUE TO THE HISTORICAL AND MODERN METAL POLLUTION EXPOSURE

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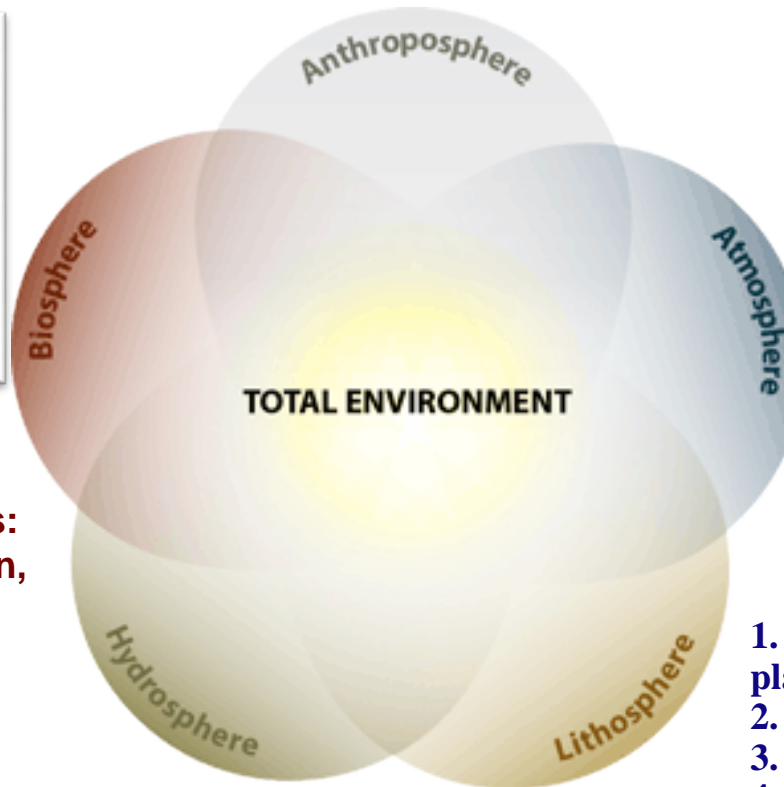
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# CHEMICAL ELEMENTS IN ENVIRONMENT



**Biologically effective elements:**  
Na, Mg, Si, P, S, Cl, Ca, Ti, V, Mn,  
Fe, Co, Mo

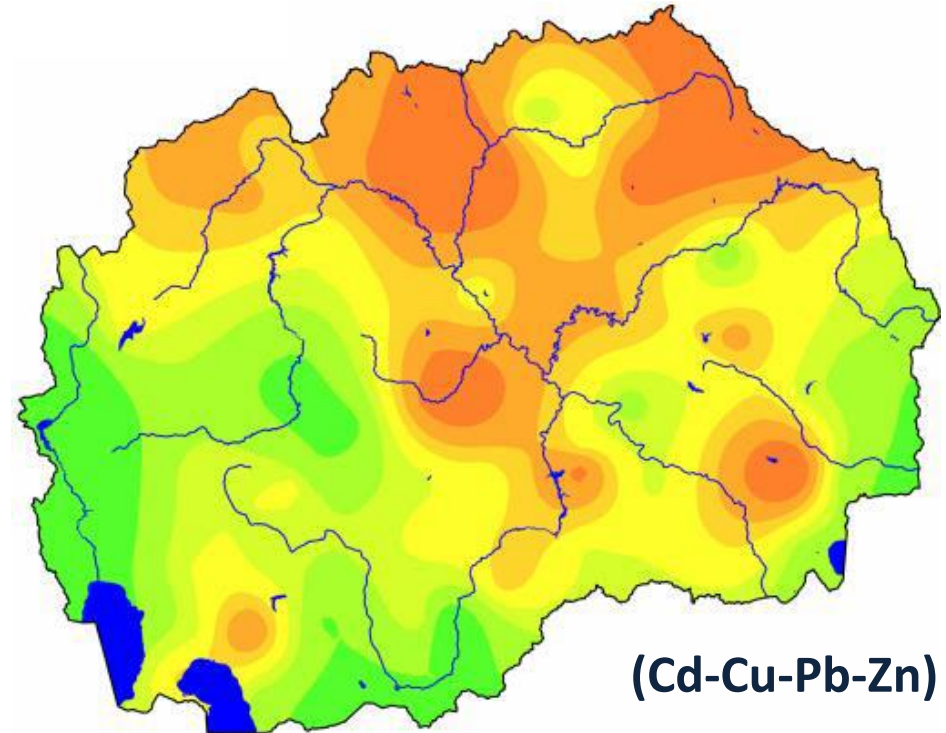
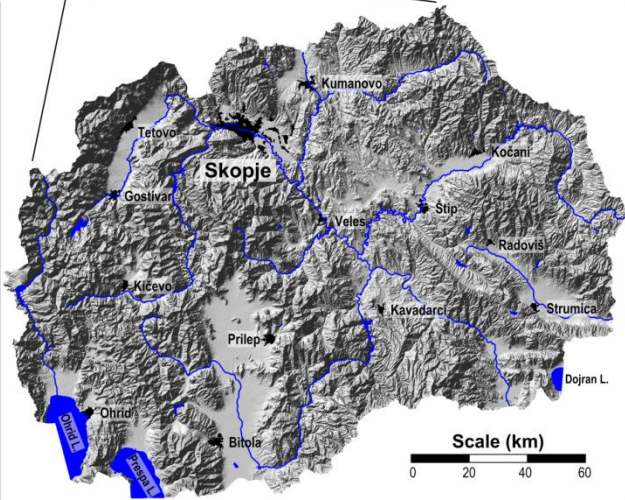


1. municipal wastewater-treatment plants
2. manufacturing industries,
3. mining,
4. transportation
5. agricultural cultivation & fertilization.....



**Biologically hazardous elements:**  
Bi, Be, Al, Cr, Ni, As, Nb, Ag, Cd, Sb, Ba, Hg, Pb,  
Zn

# MULTI-ELEMENT DISTRIBUTION IN SOIL REPUBLIC OF MACEDONIA

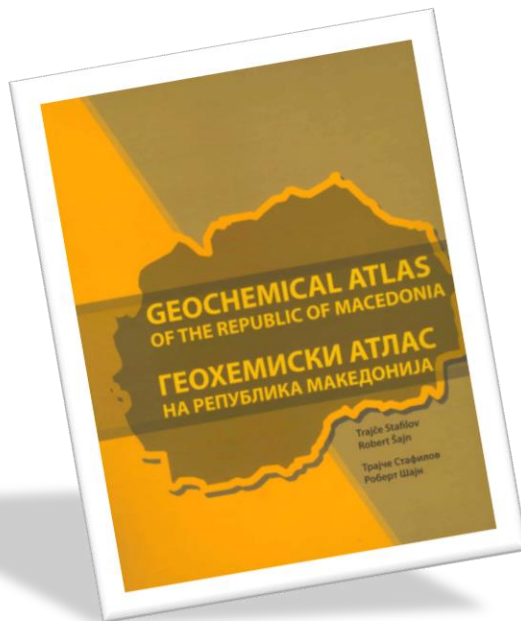


(Cd-Cu-Pb-Zn)

Factor 2 (percentiles of distribution)



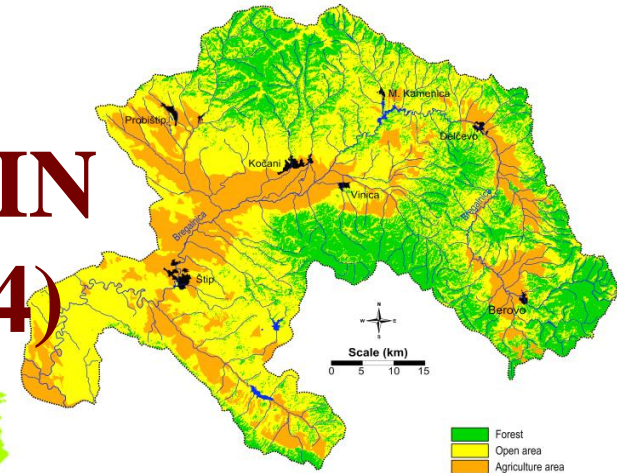
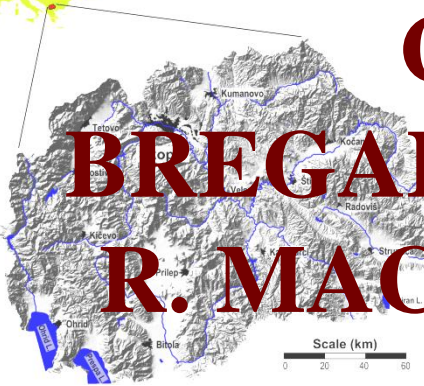
**Dominant geochemical association  
In areas with anthropogenic  
introducing of heavy metals**





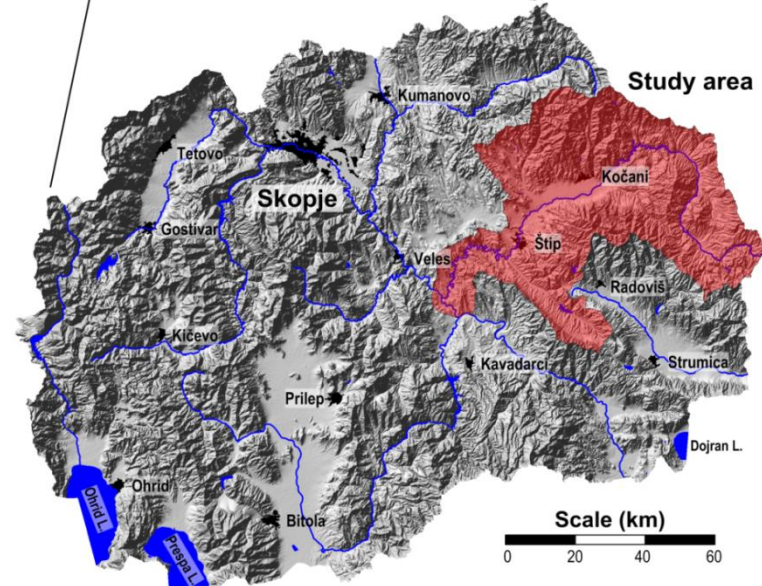
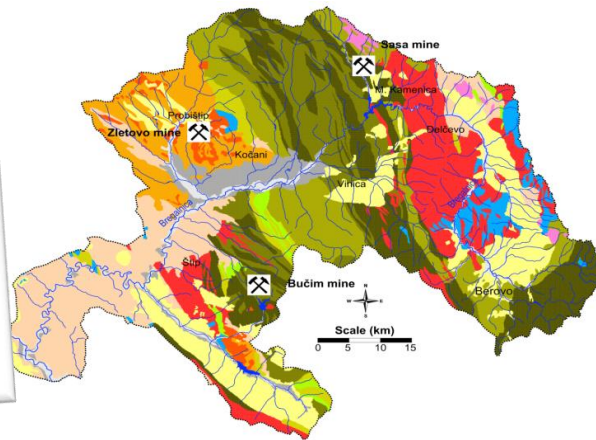
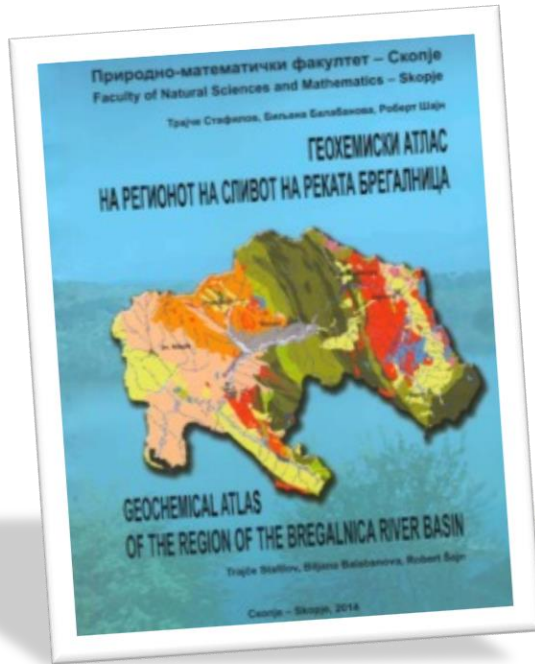
# CASE STUDY:

# BREGALNICA RIVER BASIN R. MACEDONIA (2012-2014)



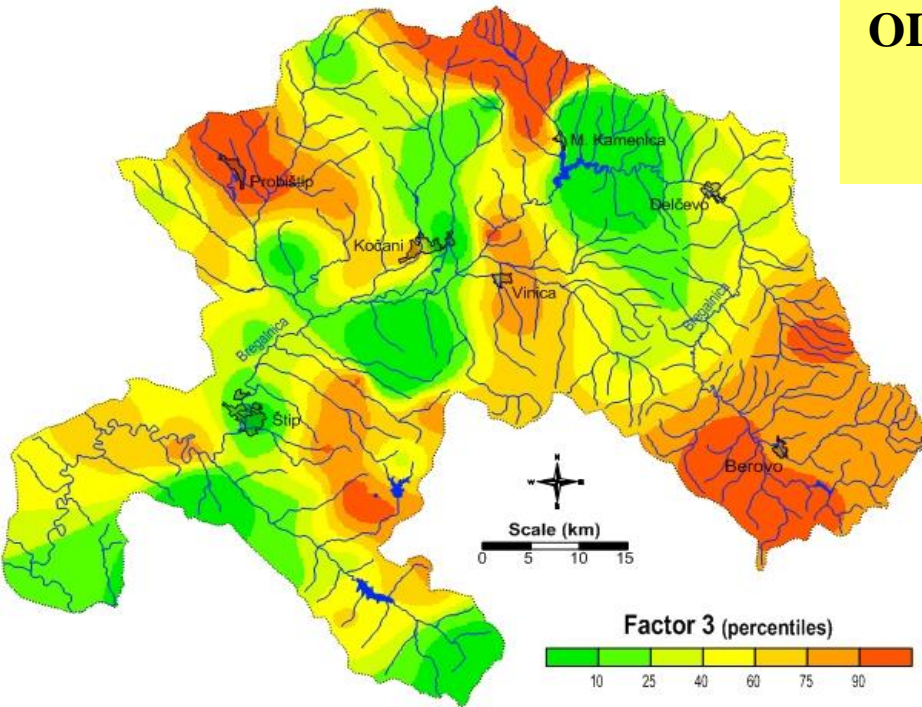
## Bregalnica River basin

N: 41°27'-42°09'  
E: 22°55'-23°01'



- |                                      |   |
|--------------------------------------|---|
| Quaternary alluvium                  | Mesozoic and Paleozoic carbonate rocks  |
| Quaternary terrace                   | Pleozoic Shales                         |
| Neogene clastites                    | Rifeous Shales                          |
| Neogene pyroclastites                | Proterozoic Shales                      |
| Neogene vulcanites                   | Proterozoic Gneisse                     |
| Paleogene flysch                     | Proterozoic felsic plutonites (granite) |
| Paleogene volcanic sedimentary rocks | Proterozoic mafic plutonites (gabbro)   |

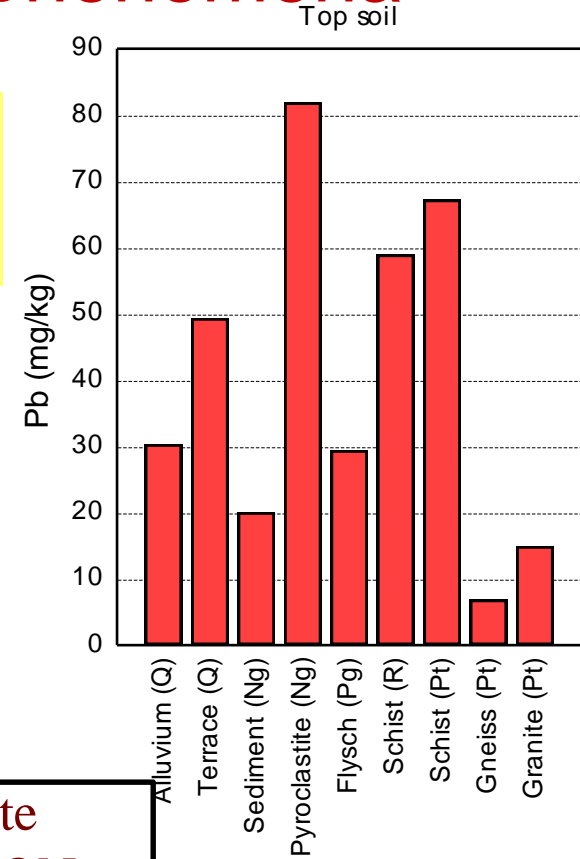
# Lithogenic vs. Anthropogenic phenomena



**(Ag-Bi-Cd-Cu-In-Mn-Pb-Sb-Te-W-Zn)**

**OLIGOCENE AND  
NEOGENE  
VOLCANISM**

Occurrence in  
area with  
dominance of



**Multivariate  
EXTRACTION**

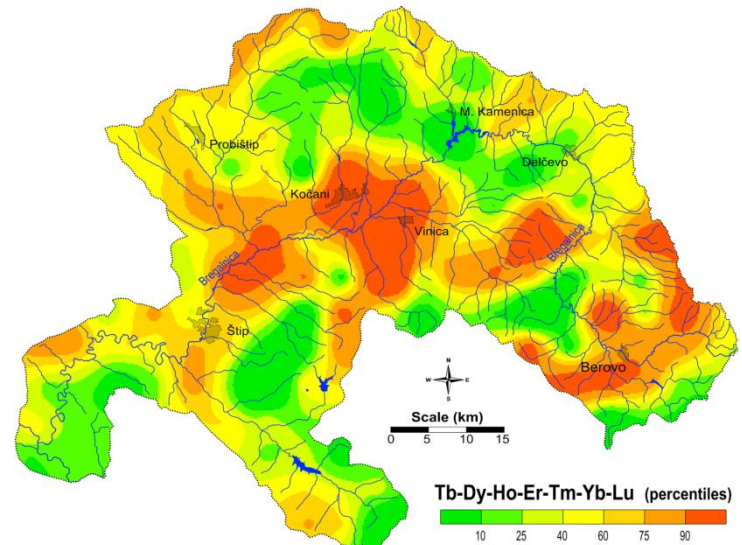
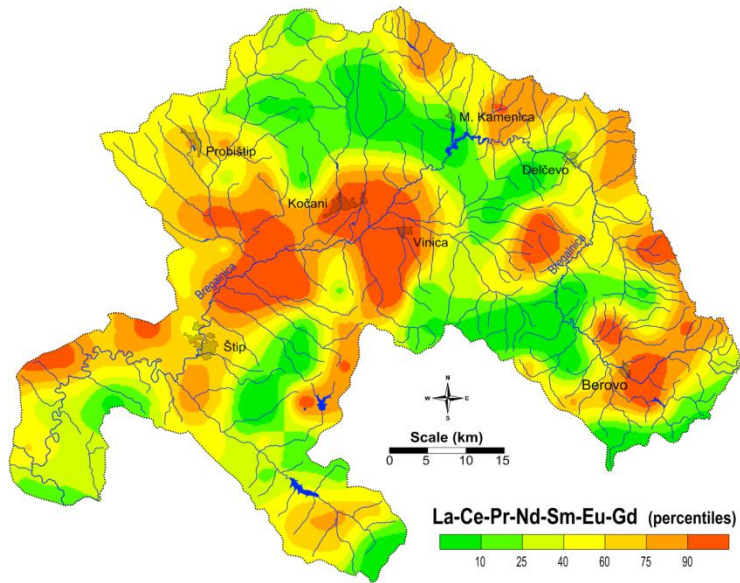
## ANTHROPOGENIC ANOMALIES!!!

**Total 69 elements: Ag, As, Al, Au, B, Ba, Be, Bi, Br, Ca, Cd, Ce, Co, Cr, Cs, Cu, Dy, Er, Eu, Fe, Ga, Gd, Ge, Hf, Hg, Ho, I, In, Ir, K, La, Li, Lu, Mg, Mn, Mo, Na, Nb, Nd, Ni, Os, P, Pb, Pd, Pr, Pt, Rb, Re, Rh, Ru, Sb, Sc, Se, Sm, Sn, Sr, Ta, Tb, Te, Ti, Th, Tl, Tm, V, W, Y, Yb, Zn and Zr**

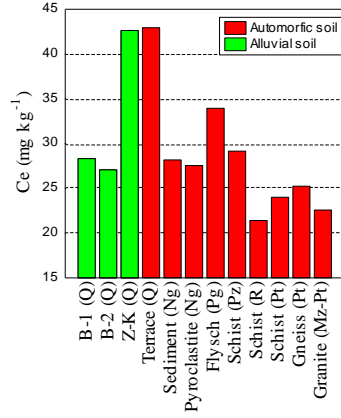
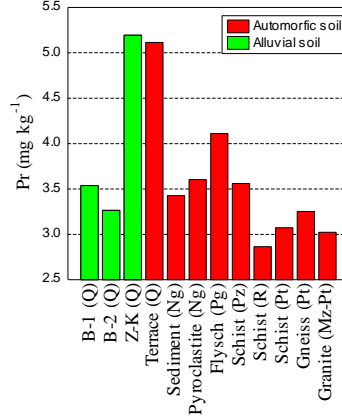
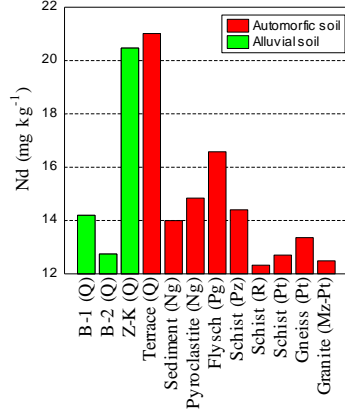
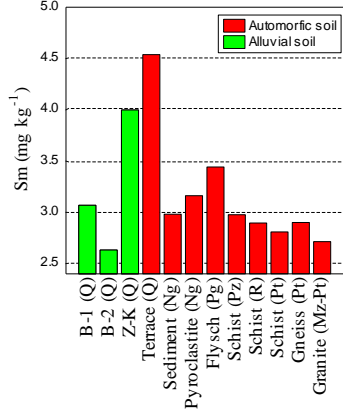
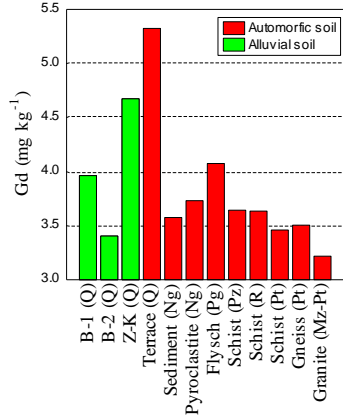
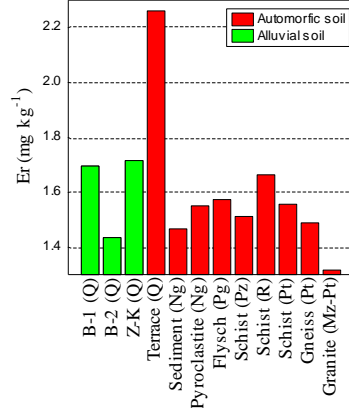
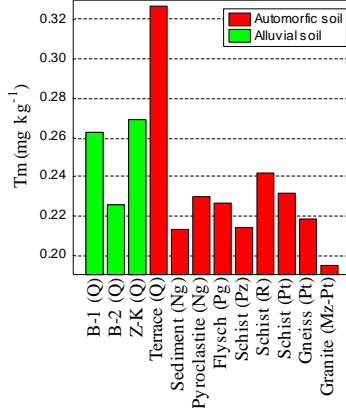
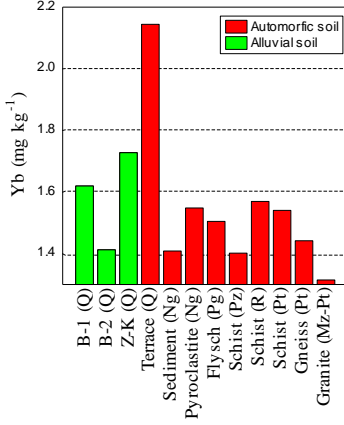
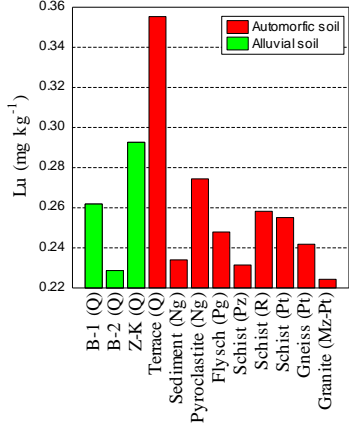
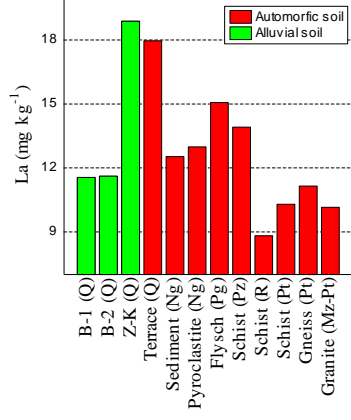
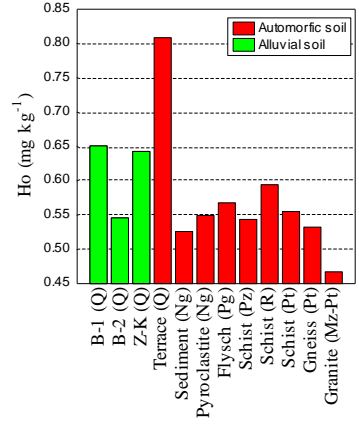
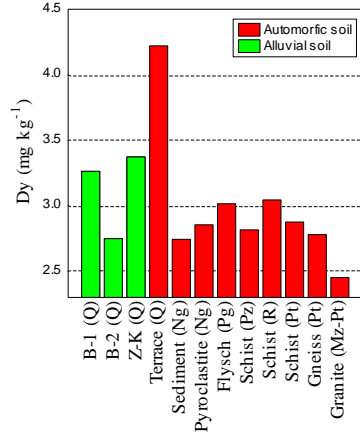
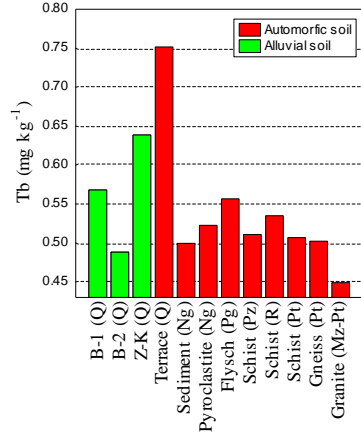
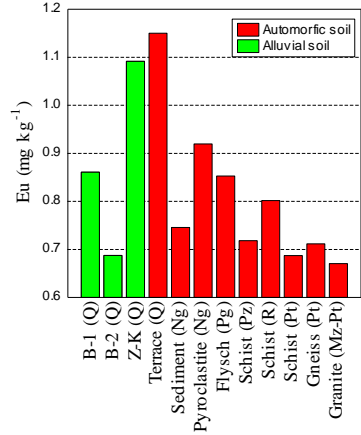
- Balabanova et al. (2017) Archives of Environmental Contamination and Toxicology.
- Balabanova et al. (2017) Journal of Environmental Science and Health, Part A.
- Balabanova et al. (2016) Environmental Science and Pollution Research.
- Balabanova et al. (2016) Macedonian Journal of Chemistry and Chemical Engineering.
- Balabanova et al. (2016) Journal of Environmental Science and Health, Part A.
- Balabanova et al. (2015) Journal of Environmental Health Science & Engineering.
- Balabanova et al. (2015) Macedonian Journal of Chemistry and Chemical Engineering.

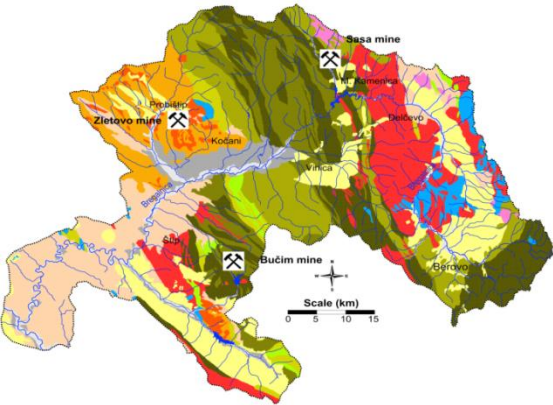


# Lithogenic distribution of REEs



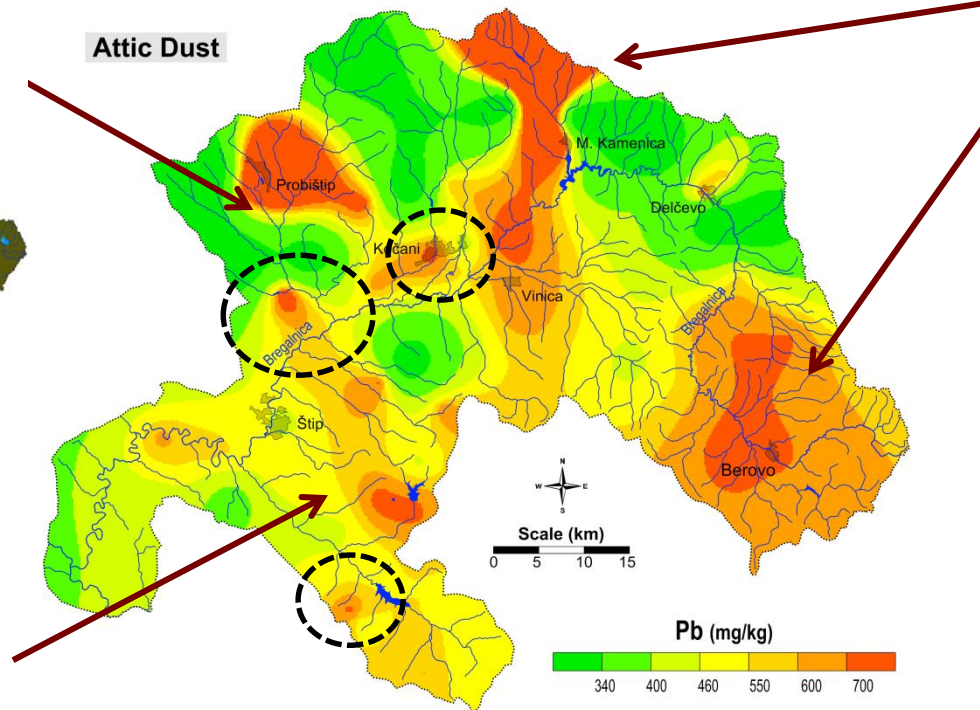
Ce	1.00															
Dy	0.77	1.00														
Er	0.70	0.99	1.00													
Eu	0.84	0.92	0.89	1.00												
Gd	0.87	0.97	0.94	0.96	1.00											
Ho	0.72	1.00	1.00	0.90	0.95	1.00										
La	0.96	0.72	0.65	0.79	0.84	0.67	1.00									
Lu	0.70	0.94	0.96	0.87	0.90	0.95	0.65	1.00								
Nd	0.96	0.88	0.83	0.93	0.96	0.85	0.93	0.83	1.00							
Pr	0.97	0.84	0.79	0.90	0.93	0.81	0.95	0.79	1.00	1.00						
Sm	0.91	0.94	0.89	0.96	0.99	0.91	0.88	0.88	0.99	0.97	1.00					
Tb	0.82	0.99	0.97	0.95	0.99	0.98	0.78	0.93	0.93	0.89	0.97	1.00				
Tm	0.69	0.98	1.00	0.87	0.92	0.99	0.64	0.97	0.82	0.78	0.88	0.96	1.00			
Yb	0.69	0.97	0.99	0.87	0.92	0.98	0.65	0.99	0.82	0.79	0.89	0.95	1.00	1.00		
	Ce	Dy	Er	Eu	Gd	Ho	La	Lu	Nd	Pr	Sm	Tb	Tm	Yb		



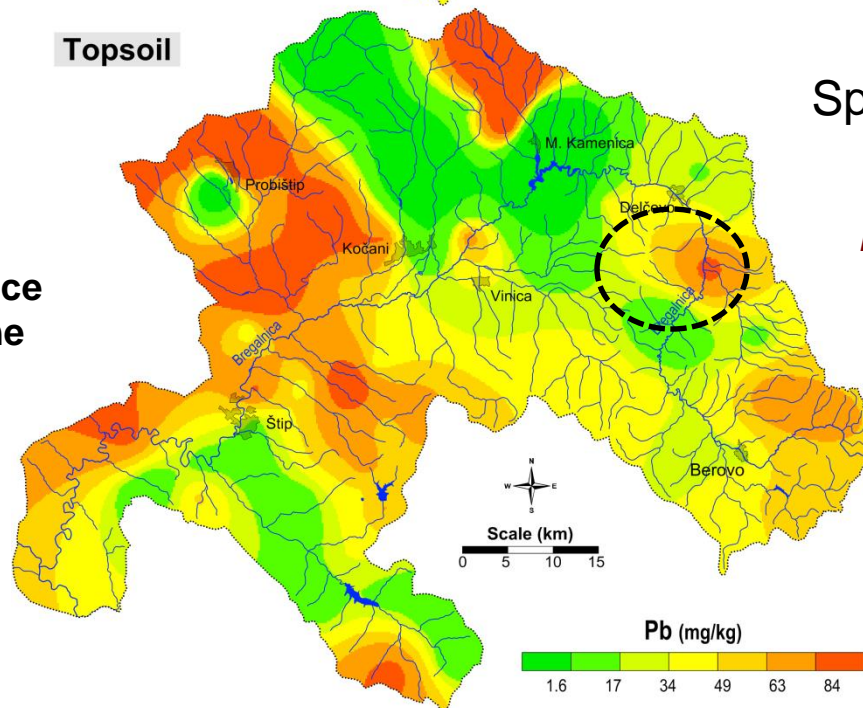


- Quaternary alluvium
- Quaternary terraces
- Neogene clastics
- Neogene pyroclastics
- Neogene volcanics
- Paleogene flysch
- Paleogene volcanic sedimentary rocks
- Mesozoic and Paleozoic carbonate rocks
- Paleozoic Shales
- Riftous Shales
- Proterozoic Shales
- Proterozoic Gneiss
- Proterozoic felsic plutonites (granite)
- Proterozoic mafic plutonites (gabbro)

### Attic Dust



### Topsoil

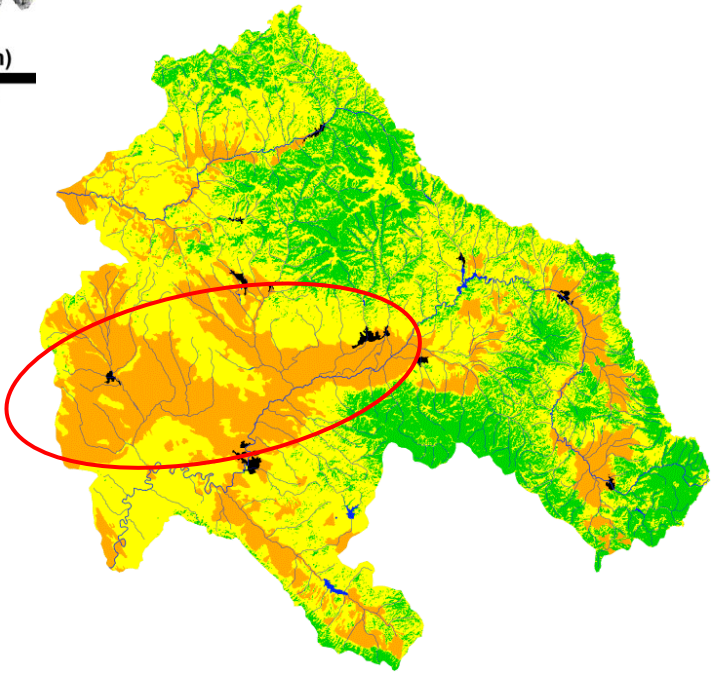
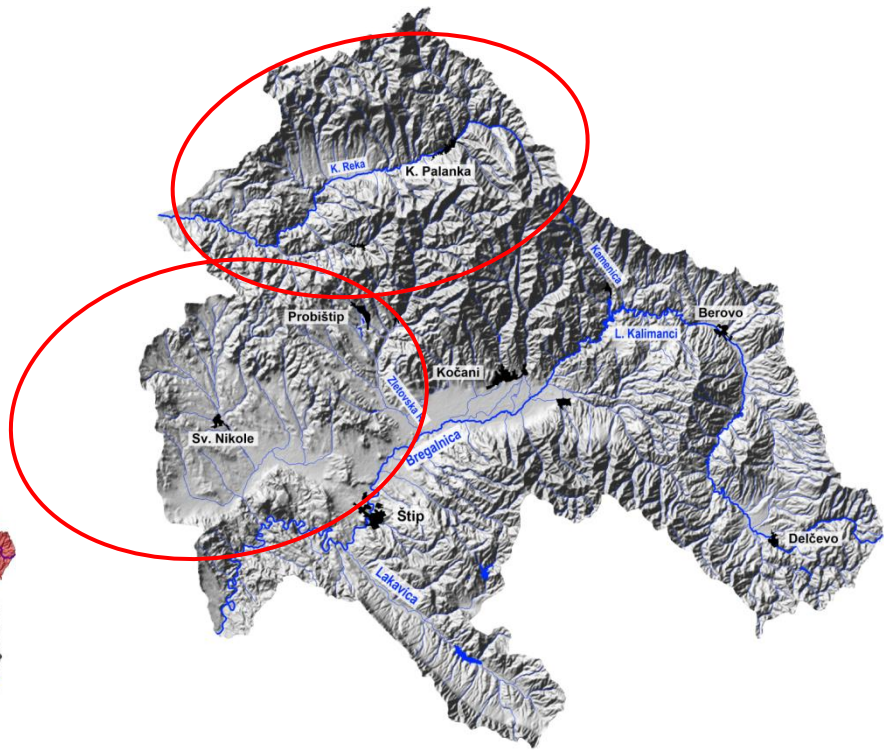
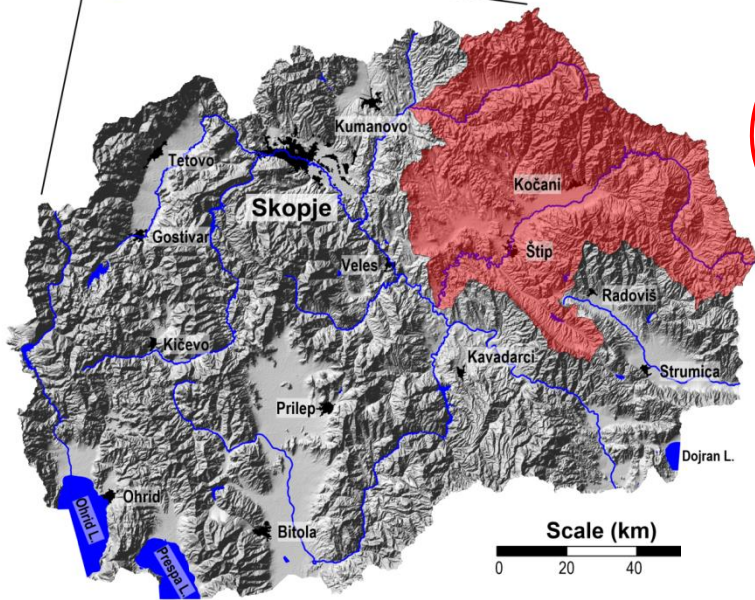


Spotting effect!!!!

*Kriging anomaly?*

Areas with dominantly presence of the Oligocene and Neogene volcanism

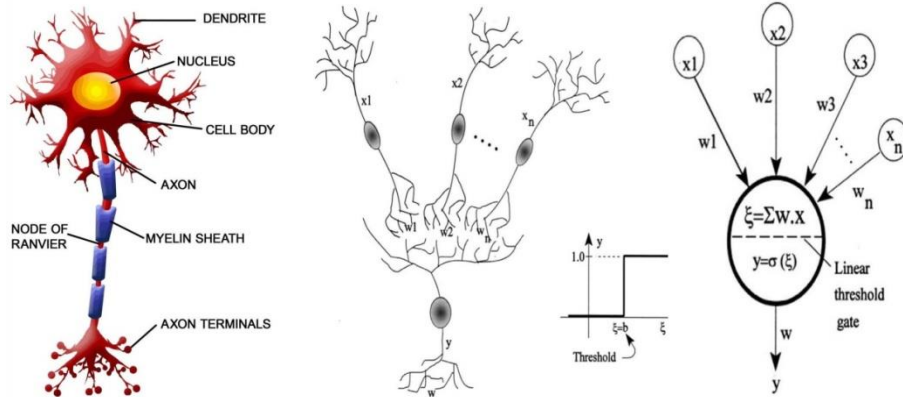




# Artificial neural networks

**Artificial Neural Network** - computer simulation of human neurons

SYSTEM THAT CAN HANDLE A LARGE NUMBER OF INPUT AND OUTPUT PARAMETERS

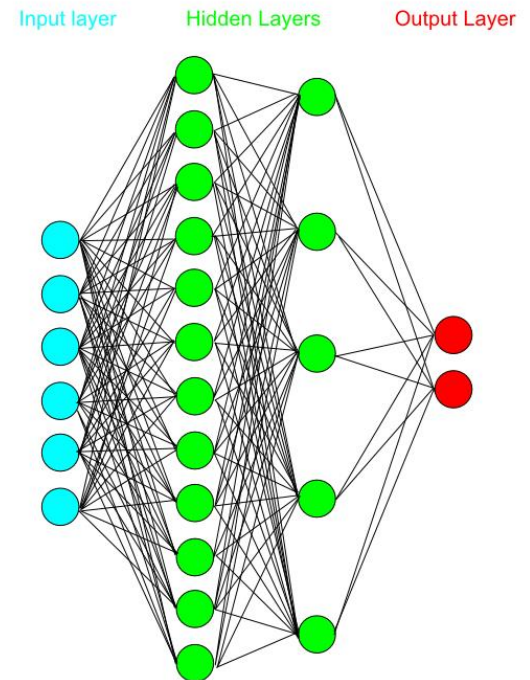


Biological neuron and mathematical model of McCulloch and Pitts neuron

## REASONS FOR APPLICATION

- They can model extremely complex systems, which cannot be modeled by methods based on linear algebra
- No problems with the dimensionality
- Due to well developed learning algorithms they are easy to use

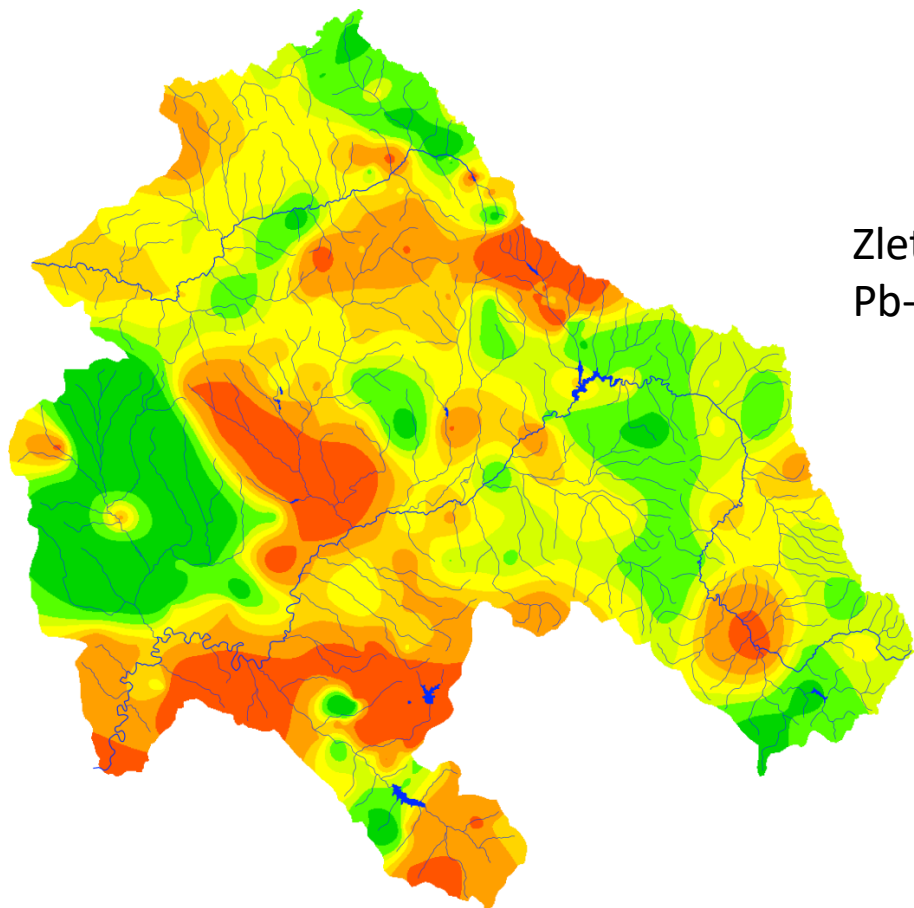
Multilayer perceptron architecture



# Multilayer perceptron

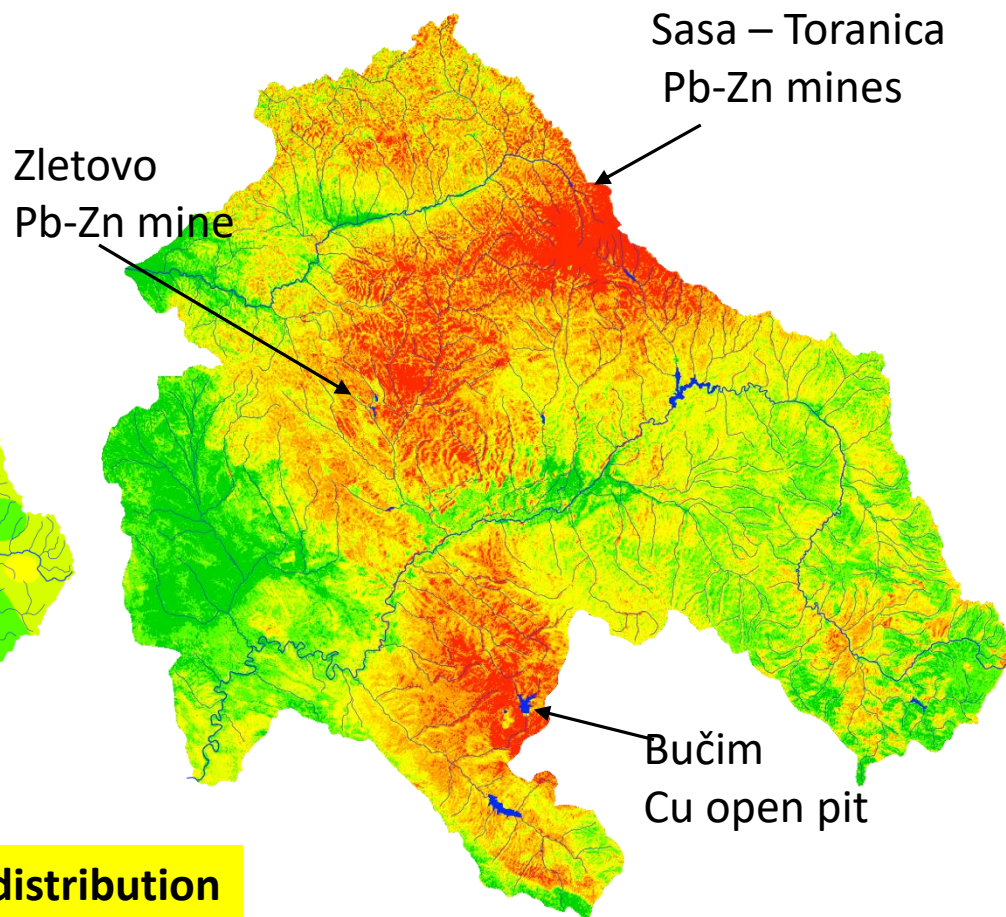


# Universal Kriging



Copper distribution

# (ANN-MLP)



(Ag-Bi-Cd-Cu-In-Mn-Pb-Sb-Te-W-Zn)



# ***PERSPECTIVES.....***

## **MULTI-DISCIPLINARY APPROACH**



**TRACE AND RARE-EARTH ELEMENT ANALYSIS**

**MULTI-ISOTOPE RATIO**

**METAL BINDING SCAN ANALYSIS**

### **CHEMOMETRIC TOOLS:**

Principal Component Analysis (PCA)

Cluster Analysis (CA)

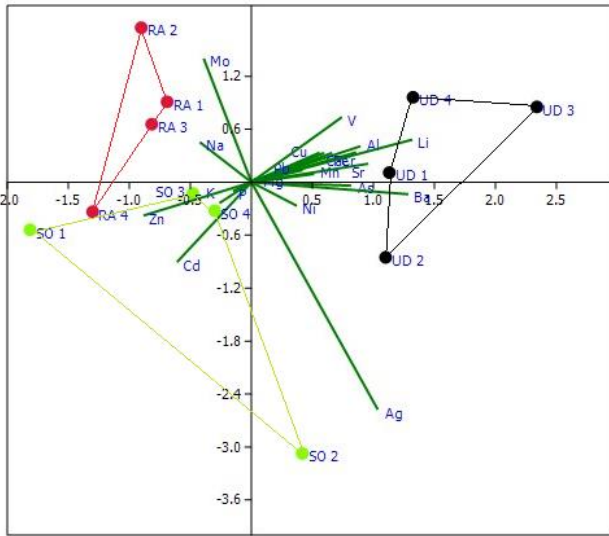
Linear Discriminant Analysis (LDA)

Canonical Discriminant Analysis (CDA)

Artificial Neural Networks (ANN)

# MULTI-ELEMENT CHARACTERIZATION OF PLANT FOOD

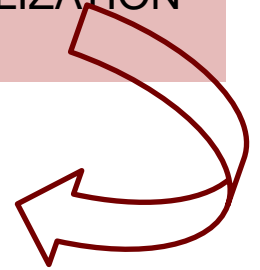
Spinach (*S. oleracea* - SO)  
 Sorrel (*R. acetosa* - RA)  
 Common nettle (*U. dioica* - UD)



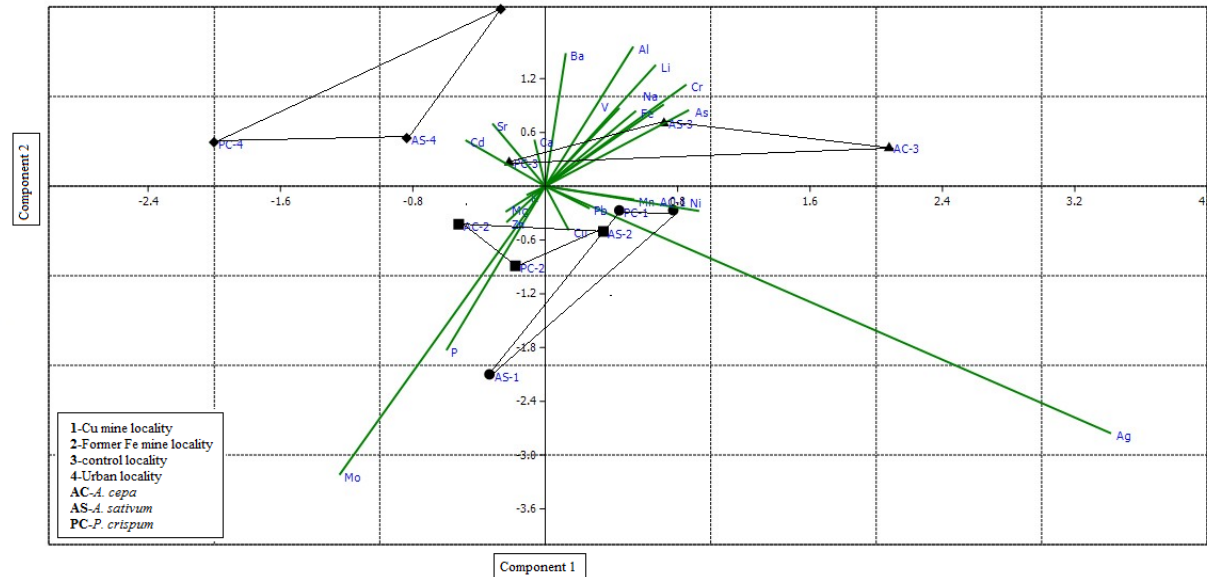
Factor 1

- 1-Cu mine locality
- 2-Former Fe mine locality
- 3-control locality
- 4-Urban locality

- None of the species was specified as a hyper-accumulator
- species show potential for PHYTOSTABILIZATION of **Cd, Cu, Pb** and **Zn**



Garlic (*A. sativum* - AS)  
 Onion (*A. cepa* - AC)  
 Parsley (*P. hortense* - PC)



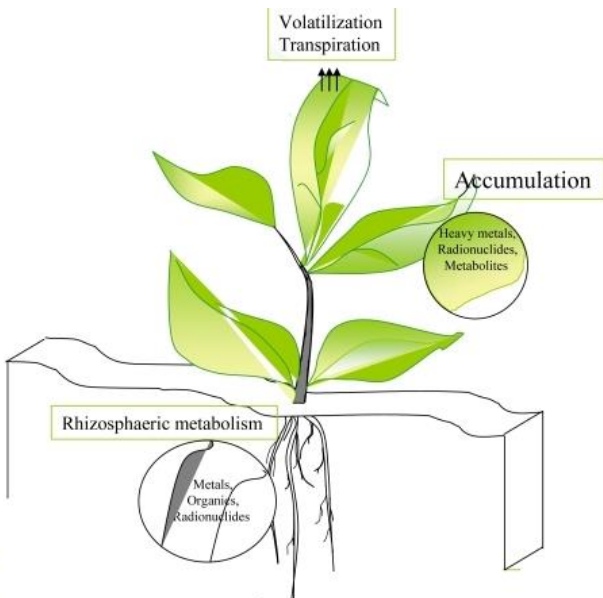
- 1-Cu mine locality
- 2-Former Fe mine locality
- 3-control locality
- 4-Urban locality
- AC-*A. cepa*
- AS-*A. sativum*
- PC-*P. crispum*

Component 1

# Isotopic measurements

## Good to know.....

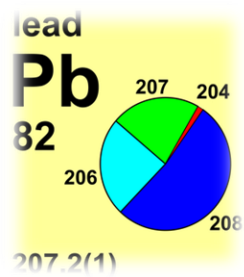
- Small Pb abundance variations occurs in nature
- The isotopic composition of lead in the environment is dependent on the local pollutant source



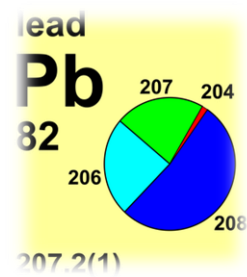
*subsequent isotope ratio studies might provide unique means of differentiating between different plant source of origin*

- **Small Sr abundance variations**
- **Large atomic mass  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios change little as they pass from weathered rocks through soils to the food chain**





# Lead isotopes



Four stable isotopes of Pb with the following approximate abundances:

- $^{208}\text{Pb}$  (52.4%)
- $^{207}\text{Pb}$  (24.1%)
- $^{206}\text{Pb}$  (22.1%)
- $^{204}\text{Pb}$  (1.4%)

$^{208}\text{Pb}$ ,  $^{207}\text{Pb}$  and  $^{206}\text{Pb}$  are formed by the radioactive decay of:

- $^{232}\text{Th}$  (half-life = 14 billion years),
- $^{235}\text{U}$  (half-life = 0.7 billion years),
- $^{238}\text{U}$  (half-life = 4.5 billion years), respectively.



# Strontium isotopes



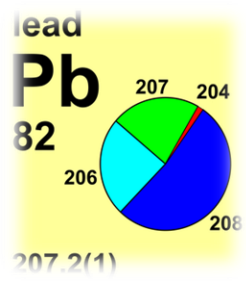
Strontium is a trace element that is found in most igneous, metamorphic and sedimentary rock

**Four stable isotopes of Sr with the following approximate abundances:**

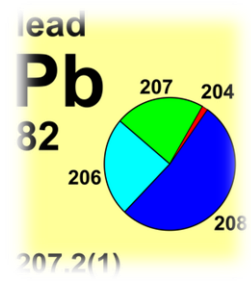
- **$^{88}\text{Sr}$  (82.5%)**
- **$^{87}\text{Sr}$  (7.04%)**
- **$^{86}\text{Sr}$  (9.87%)**
- **$^{84}\text{Sr}$  (0.56%)**

**$^{87}\text{Sr}$  is radiogenic formed by the radioactive decay of:**

- $^{87}\text{Rb}$  (half-life =  $4.88 \times 10^{10}$  years)

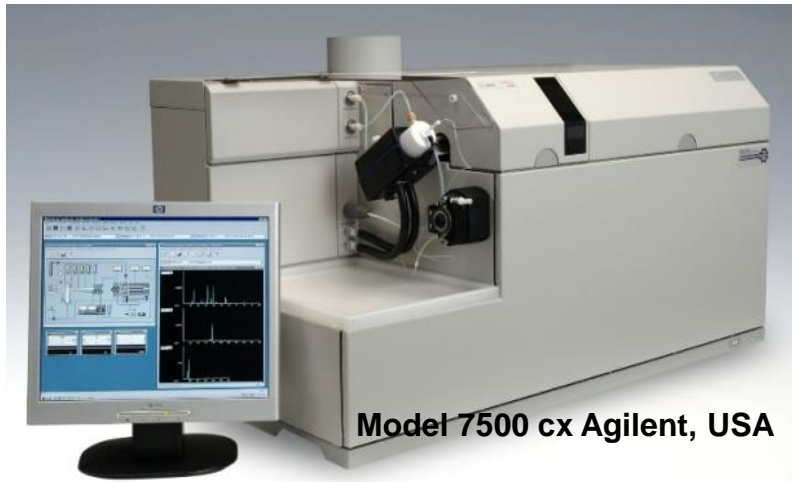
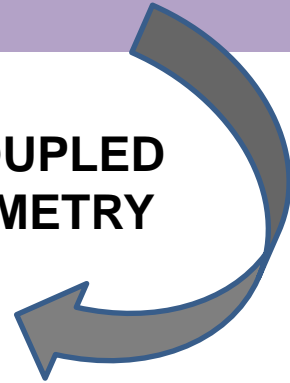


# Isotopes measurements



## DETERMINATION OF Pb ISOTOPES

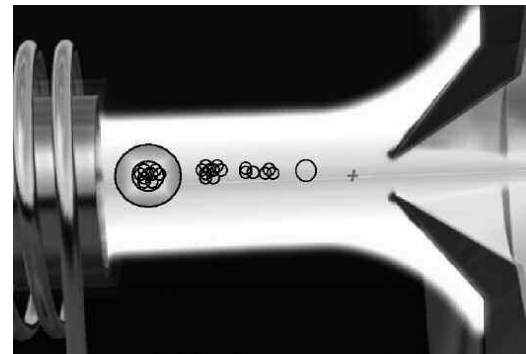
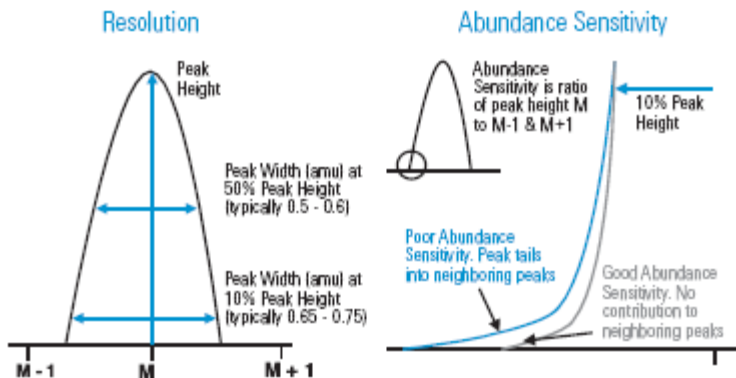
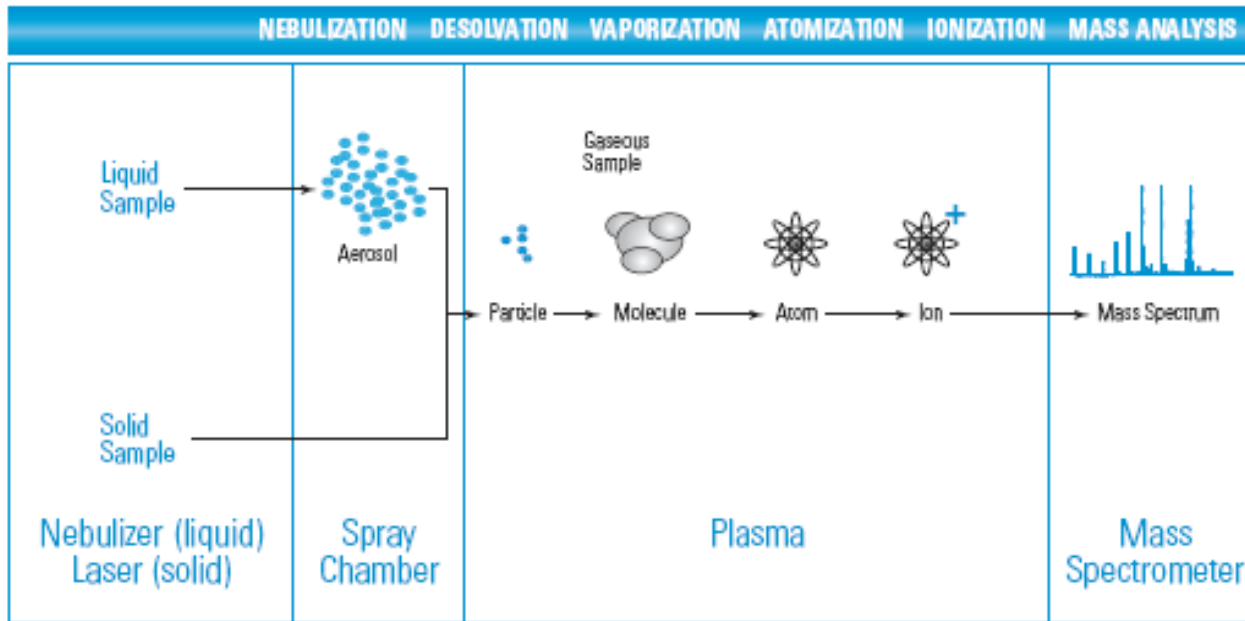
QUADROPOLE INDUCTIVELY COUPLED  
PLASMA WITH MASS SPECTROMETRY  
(Q-ICP-MS)

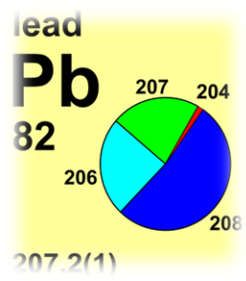


**UNILAB**

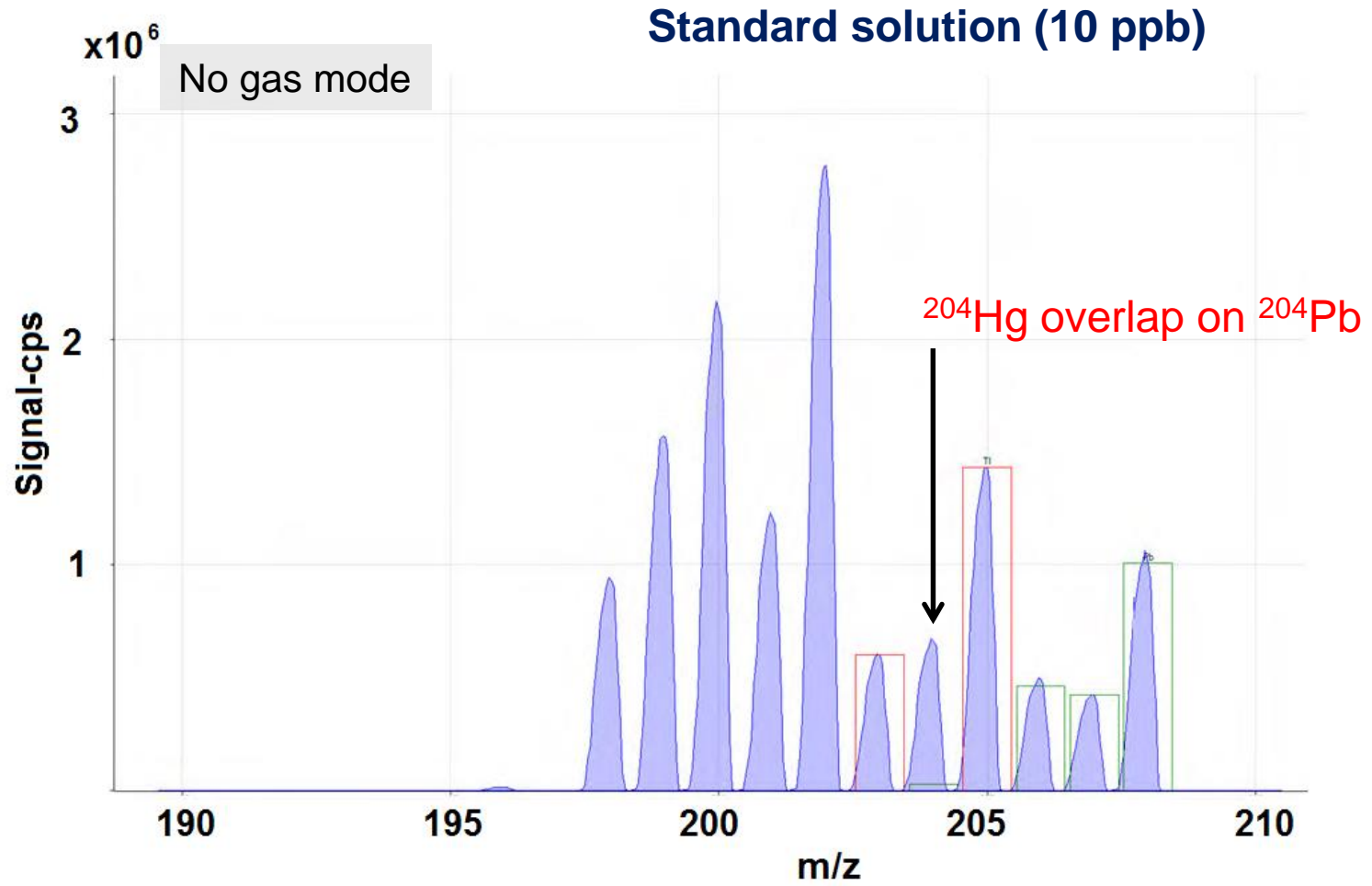
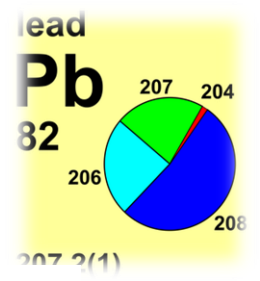


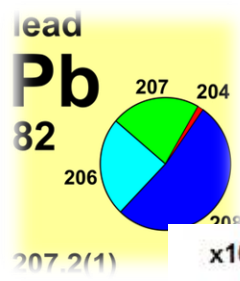
# ICP-MS measurements



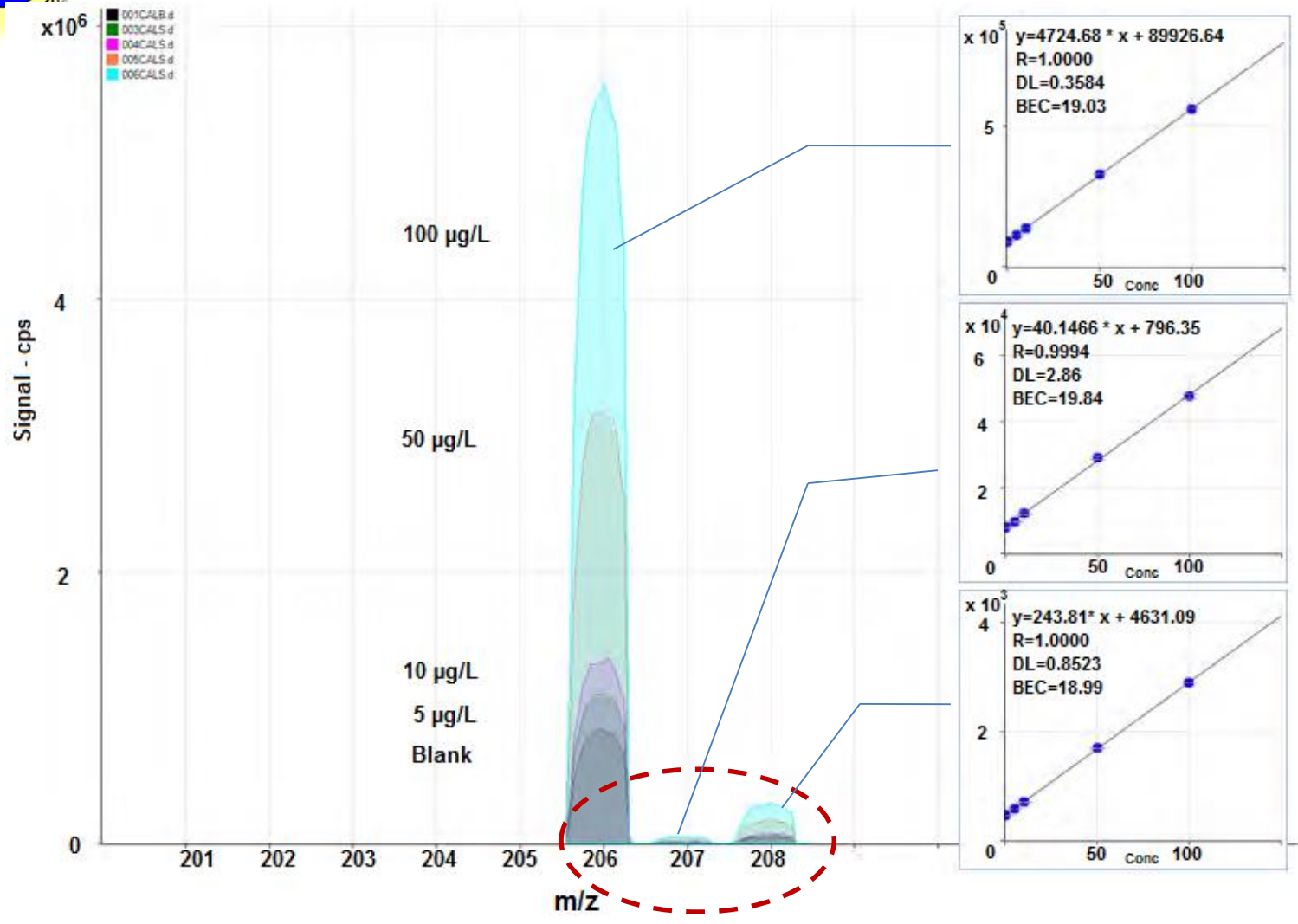
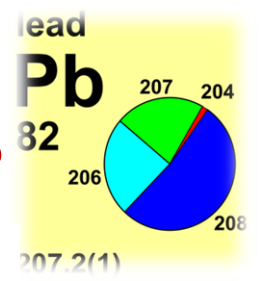


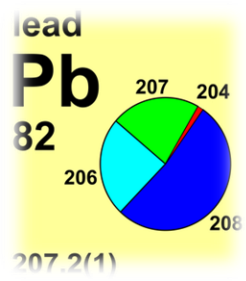
# Lead isotopes measurements



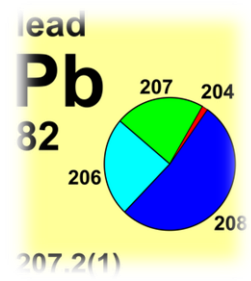


# Lead isotopes measurements



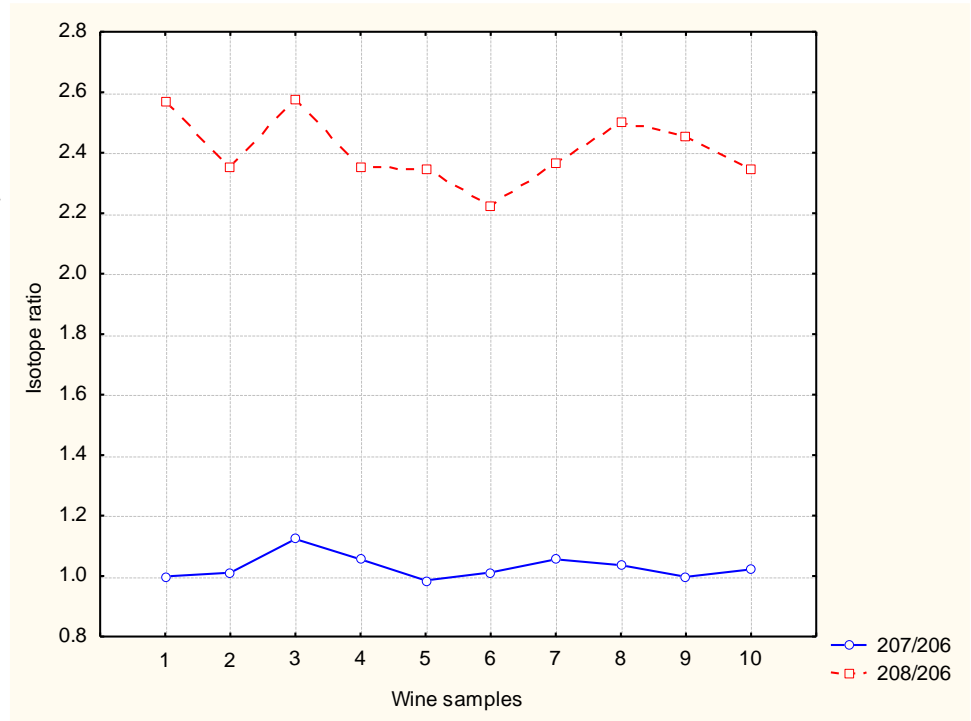


# Lead isotopes measurements for wine samples



Vranec wines - samples from the same geographical region  
Tikveš - R. Macedonia

Wine code	$^{207}\text{Pb}/^{206}\text{Pb}$	$^{208}\text{Pb}/^{206}\text{Pb}$	Pb ( $\mu\text{g/L}$ )*
1	0.998	2.569	6.35
2	1.011	2.356	8.79
3	1.122	2.578	3.74
4	1.055	2.356	3.81
5	0.985	2.345	9.75
6	1.012	2.221	11.2
7	1.055	2.365	10.2
8	1.036	2.998	16.3
9	0.997	2.457	9.58
10	1.022	2.345	10.4





# MULTI-ELEMENT CHARACTERIZATION



**VRANEC WINES** fermented with different yeast strains, autochthonous and commercial

7Li, 9Be, 11B, 23Na, 24 Mg, 27Al, 28Si, 31P, 34S, 43Ca, 48Ti, 51 V, 53Cr, 55Mn, 56Fe/57Fe, 59Co, 60Ni, 63Cu, 66Zn, 69Ga, 72Ge, 75As, 77Se, 85Rb, 88Sr, 95Mo, 107Ag, 114Cd, 115In, 120Sn, 121Sb, 125Te, 137Ba, 205Tl, 206Pb/207Pb/208Pb, and 209 Bi.

Autochthonous yeast,  
Vinalco

Four commercial yeasts:  
Clos, RC212, D254, BDX,  
from Lallemand

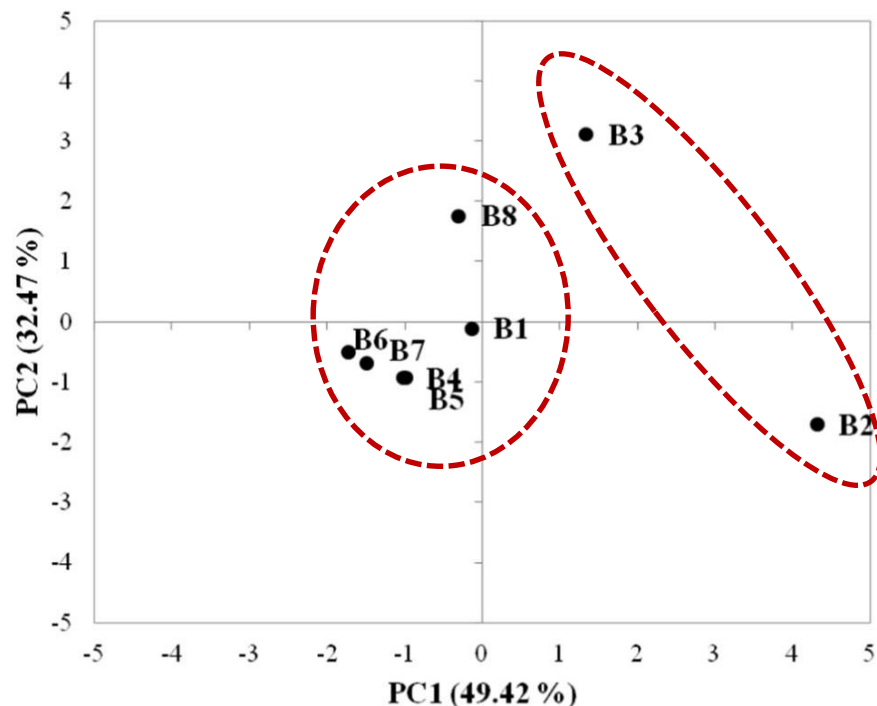
Total content		N	Diff.	Std. Dv. Diff	t	df	p
Vinalco yeast	613 mg/L	38	0.49	20.5	1.491	37	0.1444
Lallemand yeast	423 mg/L						

**NO SIGNIFICANT DIFFERENCES !!!**

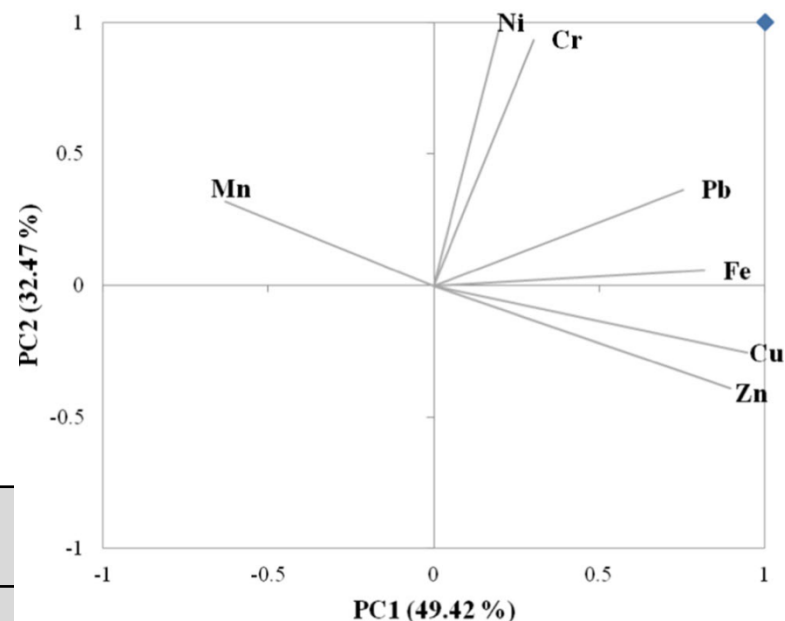


# MULTI-ELEMENT CHARACTERIZATION

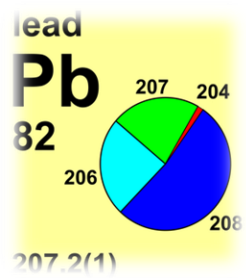
## “Rakija” (grape brandy)



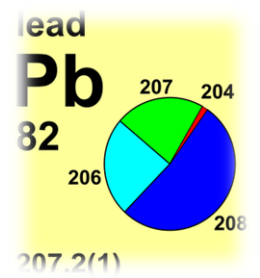
Brandy	Production	Alcohol (% v/v)
B1	Domestic distillation, aged in <b>inox</b> tank	48
B2	Domestic distillation, aged <b>inox</b> tank	56
B3	Domestic distillation, aged <b>inox</b> tank	47
B4	Industrial distillation, aged in <b>barrel</b>	62
B5	Industrial distillation, aged in <b>barrel</b>	62
B6	Industrial distillation, aged with <b>light oak chips</b>	56
B7	Industrial distillation, aged with <b>medium oak chips</b>	56
B8	Industrial distillation, aged with <b>dark oak chips</b>	56



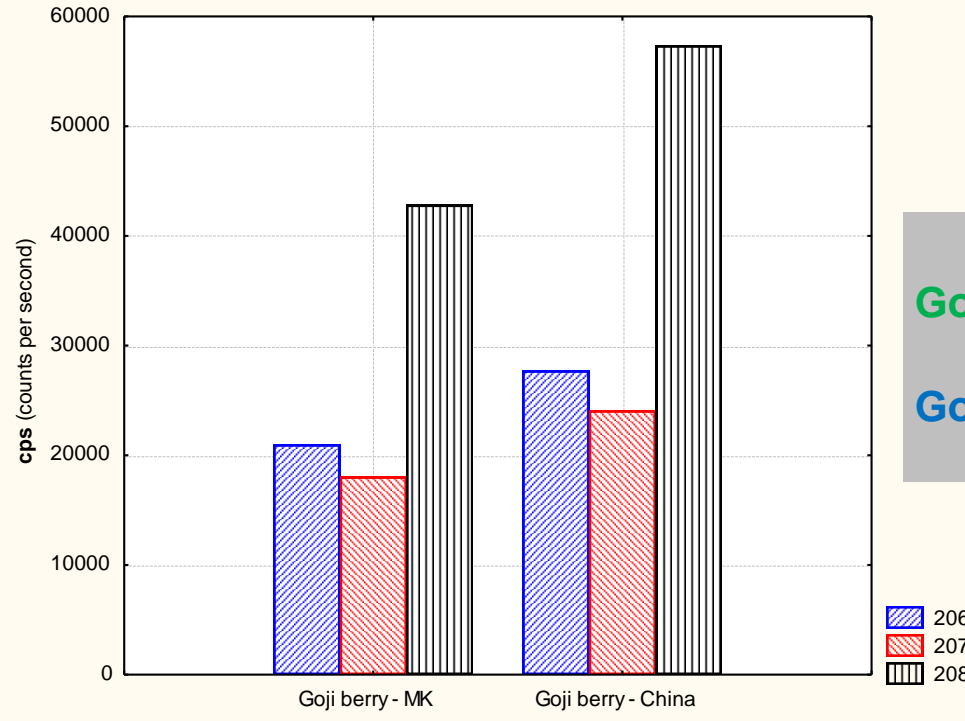
(µg/L)	Mn	Fe	Cu	Zn				
	ICP-OES				Cd	Pb	Cr	Ni
	GF-AAS							
<b>B1</b>	29	30	6120	23	< 0.3	8	< 0.7	2
<b>B2</b>	< 4	841	71200	3160	< 0.3	9	2	< 0.8
<b>B3</b>	49	241	17300	175	< 0.3	12	21	7
<b>B4</b>	38	62	2010	33	< 0.3	< 1.2	< 0.7	< 0.8
<b>B5</b>	36	74	2250	27	< 0.3	< 1.2	< 0.7	< 0.8
<b>B6</b>	118	37	3450	18	< 0.3	< 1.2	< 0.7	< 0.8
<b>B7</b>	87	11	4360	< 5	< 0.3	< 1.2	< 0.7	< 0.8
<b>B8</b>	110	594	5400	48	< 0.3	< 1.2	12	5



# Lead isotopes measurements in Goji berries



Total*	Varieties from R. China (in mg/kg)				Varieties from R. Macedonia (in mg/kg)				t	p
	min	max	Mean	SD	min	max	Mean	SD		
<b>Pb</b>	0.015	0.096	0.054	0.028	0.008	0.028	0.022	0.01	<b>3.08</b>	<b>0.03</b>

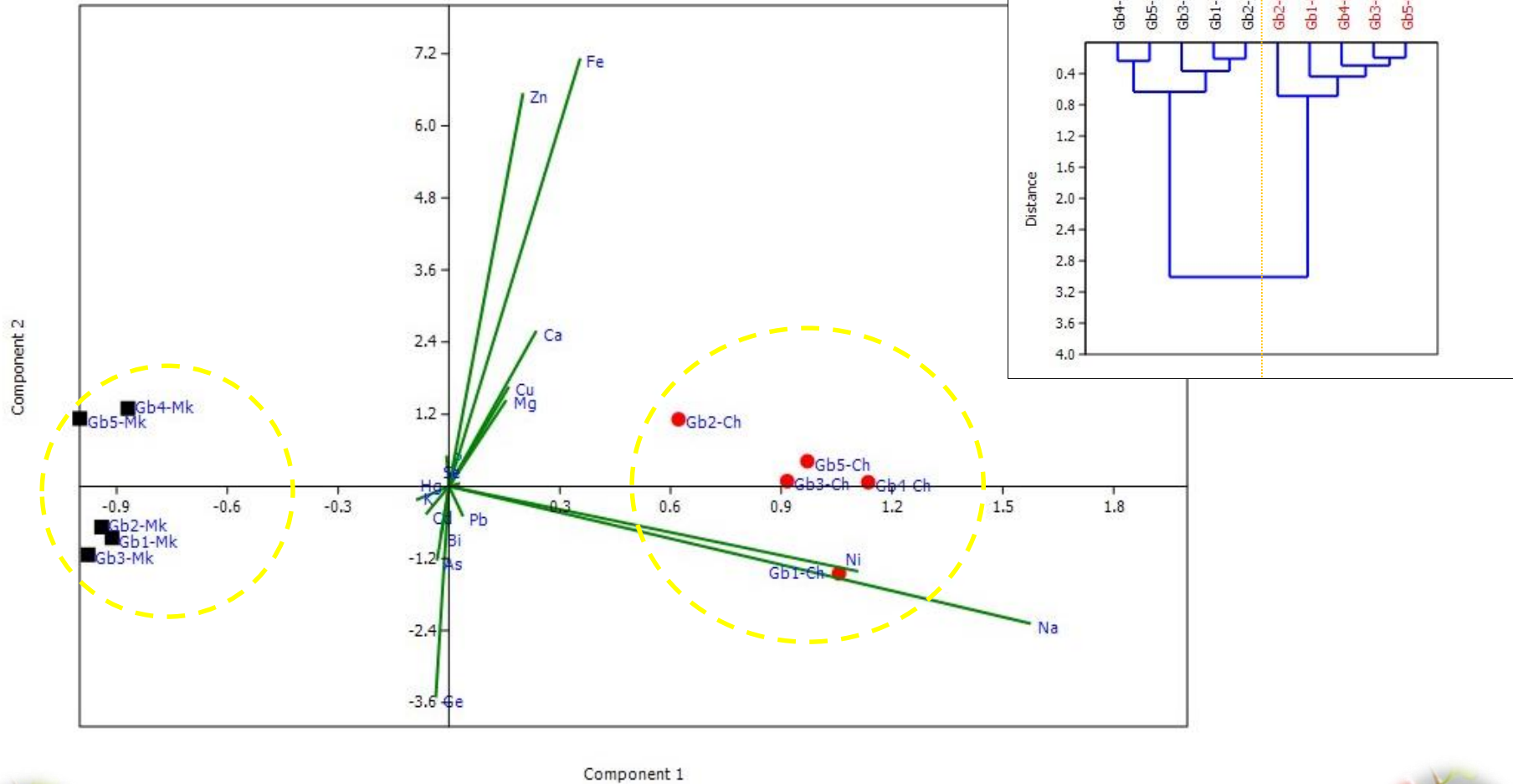


Sample	N	<sup>207</sup> Pb/ <sup>206</sup> Pb	<sup>208</sup> Pb/ <sup>206</sup> Pb
Goji berries - MK	10	0.859 ± 0.15	2.041 ± 0.09
Goji berries - Ch	10	0.868 ± 0.10	2.070 ± 0.08

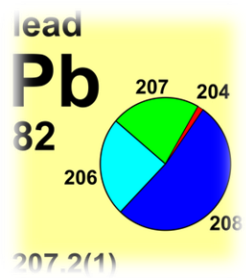
**NO SIGNIFICANT DIFFERENCES !!!**



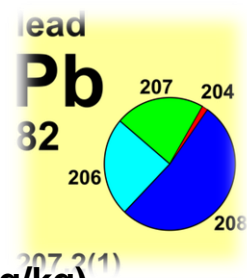
# Multi-element measurements in Goji berries



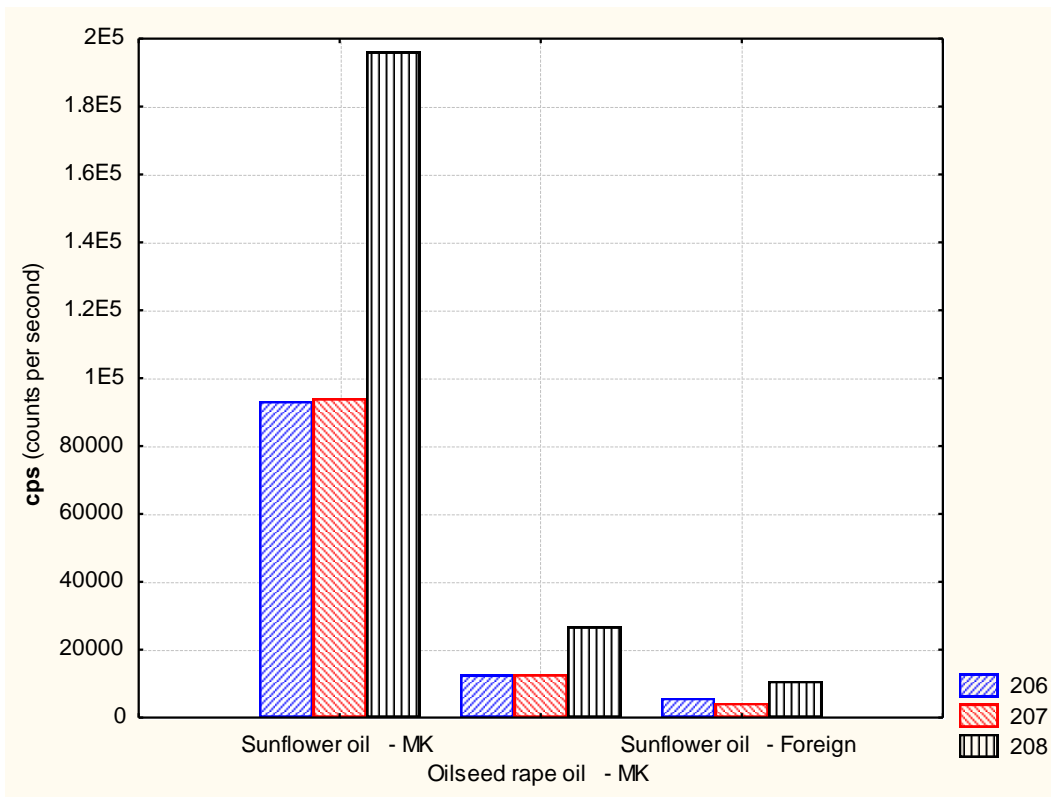




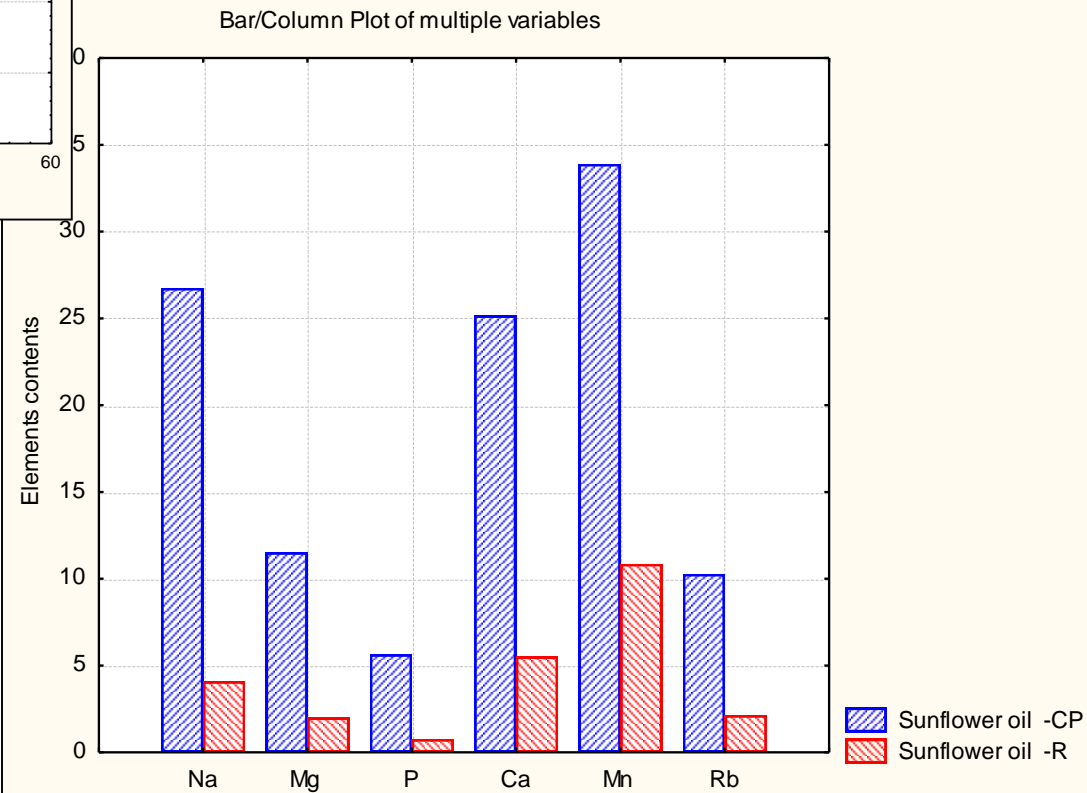
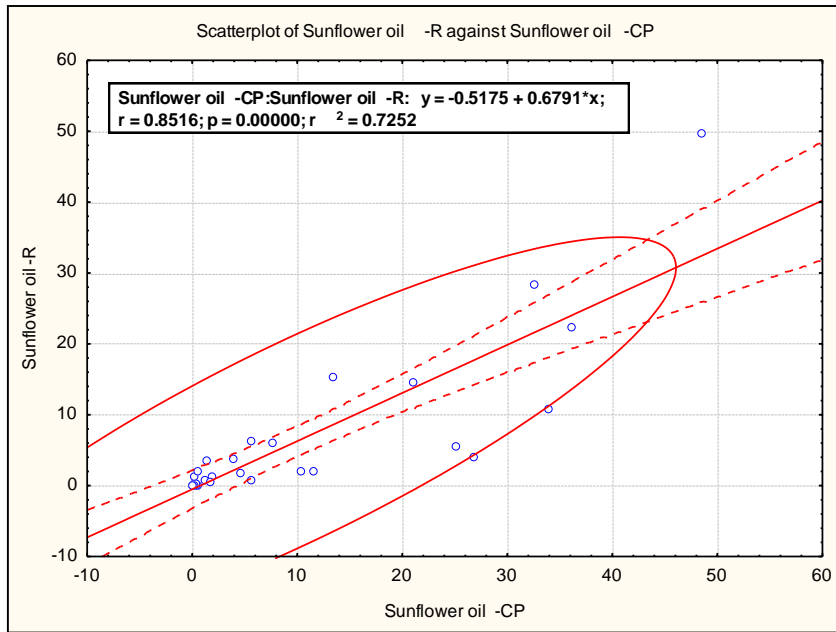
# Lead isotopes measurements for edible oil samples



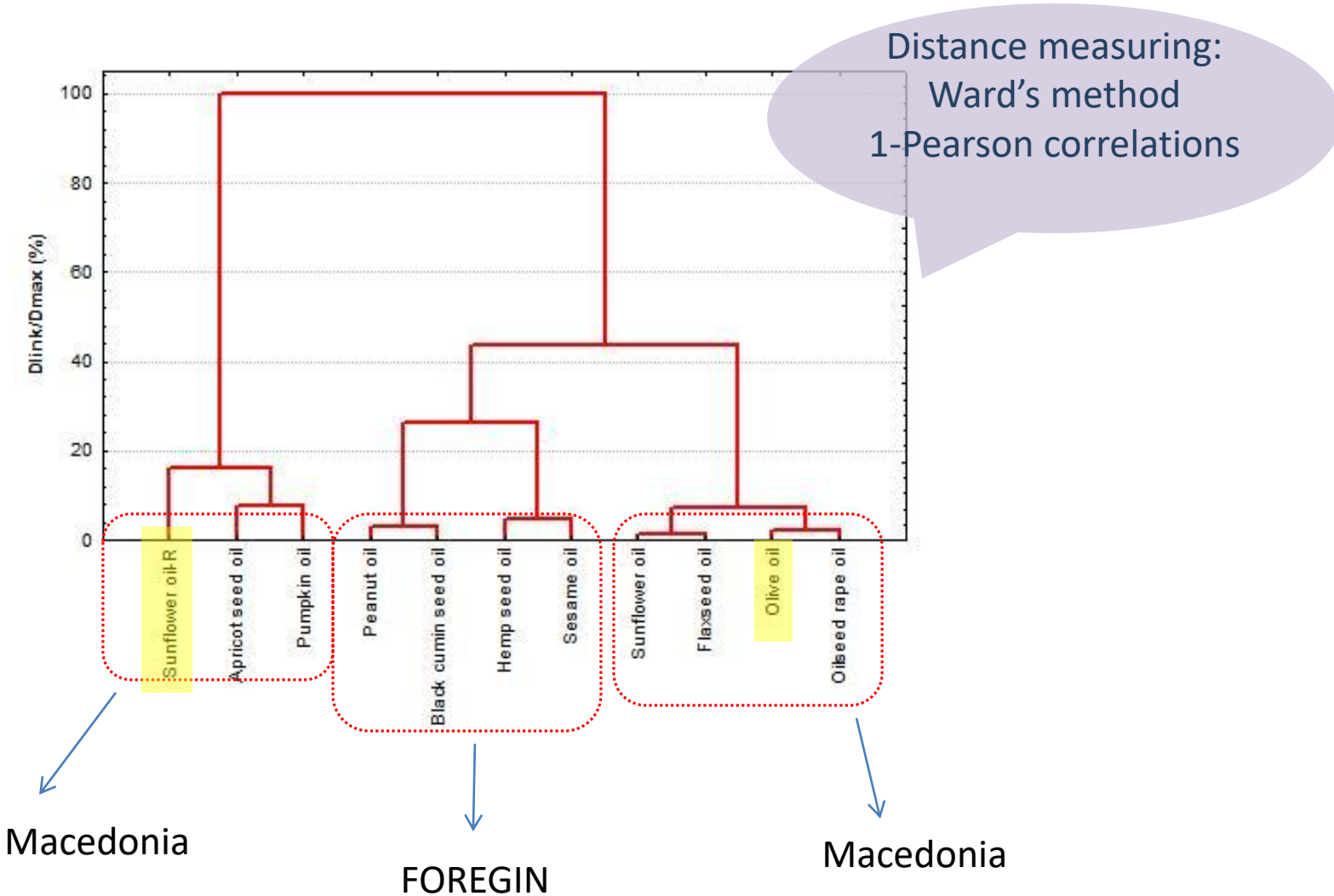
Sample	Origin	N	$^{207}Pb/^{206}Pb$	$^{208}Pb/^{206}Pb$	Pb-total ( $\mu g/kg$ ) Range*
Sunflower oil	R. Macedonia	22	$1.078 \pm 0.30$	$2.769 \pm 0.11$	18.3-29.6
Sunflower oil	<b>Foreign</b>	10	<b><math>0.859 \pm 0.18</math></b>	<b><math>2.044 \pm 0.28</math></b>	6.25-15.4
Oilseed rape oil	R. Macedonia	6	$1.09 \pm 0.25$	$2.596 \pm 0.08$	11.5 -33.6



# Multi-element measurements in edible oils



# Multi-element measurements in edible oils



# CONCLUSIONS

- **Q-ICP-MS** – sensitive method for simultaneous  $^{206}\text{Pb}$ ,  $^{207}\text{Pb}$ ,  $^{208}\text{Pb}$  and  $^{88}\text{Sr}$ , measurements using single tune mode
- $^{207}\text{Pb}/^{206}\text{Pb}$  and  $^{208}\text{Pb}/^{207}\text{Pb}$  ratios can be very useful for improving isotopic characterization of ENVIRONMENTAL ISOTOPE STUDIES.
- **Isotopic data often do not provide a *SIMPLE TRACER* to identify and distinguish source emissions.**
- **Multi-element characterization has much more expression in identification of geographical origin of the plant food**





***Thank you for your attention!***

