# CHEMICAL AND MINERALOGICAL CHARACTERIZATION OF VOLCANIC GLASS (PERLITE) FROM REPUBLIC OF MACEDONIA

### Arianit A. Reka<sup>1\*</sup>, Blagoj Pavlovski<sup>2</sup>, Blazo Boev<sup>3</sup>, Ivan Boev<sup>3</sup>, Leonora Rexhepi<sup>1</sup>, Petre Makreski<sup>4</sup>

<sup>1</sup> – Faculty of Natural Sciences and Mathematics, University of Tetovo, Ilinden n.n., 1200 Tetovo, Republic of Macedonia, <sup>2</sup> – Faculty of Technology and Metallurgy, Ss. Cyril and Methodius University, Ruger Boskovic bb, 1000 Skopje, Republic of

Macedonia

<sup>3</sup> – Faculty of Natural and Technical Sciences, Goce Delčev University, Blvd. Krste Misirkov 10-A, 2000 Štip, Republic of Macedonia

<sup>4</sup> – Faculty Institute of Chemistry, Faculty of Natural Sciences and Mathematics, Ss. Cyril and Methodius University, Arhimedova 5, 1000 Skopje, Republic of Macedonia

(\*e-mail: <u>blazo.boev@ugd.edu.mk</u>)

**Abstract:** Characterization of the volcanic glass (perlite) from Bitola region (Republic of Macedonia) was performed by the chemical, XRPD, IR and SEM analyses. The chemical analysis shows that the volcanic glass (perlite) represents an acidic volcanic rock with high percentage of SiO<sub>2</sub> (72.45%), high percentage of alkali metal oxides (4.21% K<sub>2</sub>O, 3.56 % Na<sub>2</sub>O), and loss on ignition 3.54%. Results of the XRPD analyses indicate the presence of the amorphous phase with small presence of crystalline phases (feldspars, quartz, cristobalite and magnetite). SEM examinations point to the glassy structure with presence of pores with 50-100 μm in size.

Key words: perlite, volcanic rocks, X-ray, SEM

### INTRODUCTION

The name perlite, originates from the term perlstein (pearl stone) that was given by German petrologists in the nineteenth century, to certain rhyolitic, glassy rocks with numerous concentric cracks which on fragmentation yielded pieces vaguely resembling pearls (Evans, 1993) perlite was originally identified by its vitreous, pearly luster and characteristic curved (onionskin texture) perlitic fractures (Breese, 1984; Kogel, 2006; Koukouzas, 1998). Perlite is one of the natural volcanic aluminosilicate glasses (rhyolitic rock) which has formed by rapid cooling of viscous lava or magma (Breese, 1984; Kongkachuichay, 2006) -. Upon heating (760-1100 °C), perlite is known to expand up to 20 times its initial volume, while increasing the porosity and decreasing the density. The main components of perlite are the following oxides, SiO<sub>2</sub> (70-75%), Al<sub>2</sub>O<sub>3</sub> (12-18%) as well as K<sub>2</sub>O and Na<sub>2</sub>O, other oxides such as Fe<sub>2</sub>O<sub>3</sub>, CaO, MgO, TiO<sub>2</sub>, however in smaller quantities (Burriesci et al., 1985; Gifkins et al., 2015; Gürtürk et al., 2015; Jing et al., 2011; Oktay & Odabaş, 2017; Kabra et al., 2013; Kaufhold et al, 2014; Kolvari et al, 2015; Roulia et al., 2005; Varga et al., 2015; Varuzhanyan et al., 2006). The aim of this study is to perform the chemical and mineralogical characterization of volcanic glass (perlite) from the Republic of Macedonia.

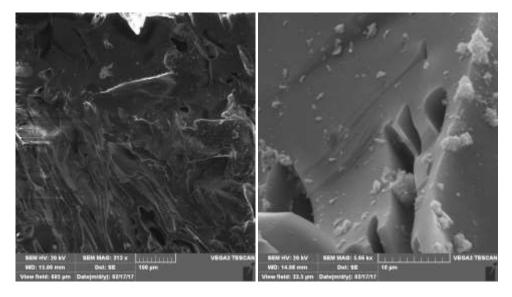
# METHODS

The raw material subject of this research is taken from Bitola region (Republic of Macedonia). The samples are characterized with a white to greyish colour, with fine grains, with clearly visible phenocrysts. The chemical composition of perlite is determined by the classical silicate analysis. The characterization from the mineralogical point of view was performed by X-ray powder diffraction (XRPD), thermal analysis (TGA/DTA), scanning electron microscopy (SEM) and infrared spectroscopy (IR).

# RESULTS

Perlite represents a white to light grey material, with clearly visible black particles of glass, with a specific gravity ranging from 2.23 - 2.40, and hardness of the samples from 4.20-5.30. Based on the results of the chemical composition, the raw material represents an acidic volcanic rock with high percentage of SiO<sub>2</sub> (72.48%), and high percentage of alkali metal oxides (K<sub>2</sub>O and Na<sub>2</sub>O) which are

present with 4.21% and 3.56% respectively. Results of the XRPD analysis of perlite depicts amorphous behaviour of the sample manifested by the appearance of one complex "bump" widely distributed between 15 and 30° (20) with the maxima peaking in the 22–26.7° (20) range, which is a result of the high percentage of aluminosilicate glass phase in which are small amounts of crystalline phases. The crystalline phases are mainly represented by the presence of feldspars (plagioclase and orthoclase). In smaller quantities compared to feldspars evident is the presence of SiO<sub>2</sub> polymorph modifications: quartz (20 - 26.674°) and cristobalite (20 - 21.946°), as well as the presence of magnetite. The IR spectrum of perlite exhibits absorption bands at 786 cm<sup>-1</sup> as result of the stretching vibrations of Si-O-Si band, whereas the band at 1078 cm<sup>-1</sup> is a result of the stretching vibrations of Si-O-Al. The band at 1644 cm<sup>-1</sup> is due to stretching and bending vibrations from the absorbed water, while the band at 3650 cm<sup>-1</sup> is due to the combination of OH stretching. On the SEM microphotograph is clearly visible the fluidal character in the glassy mass (Fig. 1a) and the compactness glass mass of perlite, with presence of some pores, as well as tiny crystals (Fig.1b).



*Figure 1.* Scanning electron microscopy of natural perlite a) The fluidal character in the glassy mass b) The compactness of the glassy mass of perlite, and the presence of tiny crystals

# DISCUSSION

- The chemical analysis of the volcanic rock shows that the raw material represents acidic volcanic rock, with the high percentage of SiO<sub>2</sub> (72.48%), while the presence of Al<sub>2</sub>O<sub>3</sub> is 13.52%. The presence of K<sub>2</sub>O and Na<sub>2</sub>O is 4.21% and 3.56% respectively.

- Results from the XRPD examinations suggest high percentage of amorphous phase, where the presence of the crystalline phases is minimal. The crystalline phase is mainly composed of feldspars, SiO<sub>2</sub> polymorphs (quartz and cristobalite), and the minor amount of magnetite and biotite.

- The results from the light microscopy confirm the results obtained from the XRD analysis; the volcanic rock is composed of isotropic amorphous glassy mass, in which present fine-grained particles - microlites.

- The results from the scanning electron microscopy adhere to the results from the optical microscopy. SEM pictures show the fluidal character in the glassy mass and the compactness glass mass of the perlite, with presence of some pores, as well as tiny crystals.

- The thermal analysis of perlite (DTA/TGA) show that the loss of mass during the thermal treatment is 3.80%, and this loss in mass corresponds to the average % of water found in perlite.

- The IR spectra provide valuable information regarding the presence of water and OH groups in perlite. The band at 1644 cm<sup>-1</sup> is due to stretching and bending vibrations from the absorbed water, whereas the band at 3650 cm<sup>-1</sup> is due to the combination of OH stretching.

#### CONCLUSIONS

Based on the abovementioned results, it can be concluded that the volcanic rock (perlite) due to the low presence of crystalline phase, can be used to obtain high quality expanded perlite. As result of the high percentage of alkali metal oxides, the high percentage of SiO<sub>2</sub>, the volcanic glass (perlite) can be used as raw material for the production of container glass. The high percentage of the glassy phase, the high percentage of SiO<sub>2</sub> and alkali metal oxides makes this raw material an attractive for production of acid resistant glazing materials. Based on its chemical and mineralogical content, the volcanic glass can be used as ceramic flux for lowering the sintering temperature.

#### REFERENCES

Breese, R.O.Y., 1987. Rhyolite domes and flows at No Aqua Peaks, Proceeding of the 35th Annual Field Conference, Socorro, New Mexico Geological Society, 373–374 pp.

Burriesci N., Arcoraci C., Antonucci PL., Polizzotti G., 1985. Physico-chemical characterization of perlite of various origins, Mater. Lett. 3, 103–110.

Evans, A.M., 1993. Perlite, Ore Geology and Industrial Minerals, 3rd Edition, Oxford, Boston, Blackwell Science. 295–296 pp.

Gifkins C., Herrmann W., Large R., 2015. Altered Volcanic Rocks. Center for Ore Deposit Research, National Library of Australia, 100–101 pp.

Gürtürk M., Oztop H. F., Hepbasli A., 2015. Comparison of exergoeconomic analysis of two different perlite expansion furnaces, Energy 80, 589–598.

Jing Q., Fang L., Liu H., Liu P., 2011, Preparation of surface-vitrified micron sphere using perlite from Xinyang, China, Appl. Clay Sci. 53, 745–748.

Oktay B.M., Odabaş E., 2017. Determining Mechanical and Physical Properties of Phospho-Gypsum and Perlite-Admixtured Plaster Using an Artificial Neural Network and Regression Models, Pol. J. Environ. Stud. 26, 2425– 2430.

Kabra S., Katara S., Rani A., 2013, Characterization and Study of Turkish Perlite, International Journal of Innovative Research in Science, Engineering and Technology, 2, 4319–4326.

Kaufhold S., Reese, A., Schwiebacher, W., Dohrmann, R., Grathoff, G.H., Warr, L.N., Halisch, M., Müller, C., Schwarz-Schampera, U., Ufer., K., 2014. Porosity and distribution of water in perlite from the island of Milos, Greece. SpringerPlus 3, 598 p.

Kogel J. E., Trivedi N. C., Baker J. M., Kurkowski S. T., 2006. Industrial Minerals & Rocks, 7th Edition, Society for Mining, Metallurgy and Exploration, Inc., 865 pp.

Kolvari E., Koukabi N., Hosseini M. M., 2015. Perlite: A cheap natural support for immobilization of sulfonic acid as a heterogeneous solid acid catalyst for the heterocyclic multicomponent reaction, J. Mol. Catal. A-chem 397, 68–75.

Kongkachuichay P., Lohsoontorn, P., 2016. Phase Diagram of Zeolite Synthesized from Perlite and Rice Husk Ash, ScienceAsia 32, 13–16.

Koukouzas N., 1998. Volcanic glass (perlite) of Kimolos island, Greece: mineral chemistry and structure, Proceedings of the 8th International, Patras, Bulletin of the Geological Society of Greece, XXXII/3, 313–332 pp.

Roulia M., Chassapis K., Kapoutsis J. A., Kamitsos E. I., Savvidis T., 2006, Influence of thermal treatment on the water release and the glassy structure of perlite, J. Mater. Sci. 41, 5870–5881.

Varga P., Lexa J., Uhlík P., Rajnoha M., 2015. Characterization of perlites from Jastrabá and Lehôtka pod Brehmi deposits, Geology, Geophysics & Environment, 41, 146–146.

Varuzhanyan Av. A., Varuzhanyan Ar. A., Varuzhanyan H. A., 2006. A Mechanism of Perlite Expansion, Inorg. Mater. 42, 1039–1045.