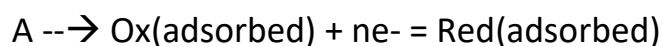


SURFACE $C_{rev}E$ MECHANISM IN SQUARE WAVE VOLTAMMETRY-MATHCAD SIMULATION PROTOCOL

Rubin Gulaboski, Valentin Mirceski, Milivoj Lovric

Abstract

Surface reaction of dissociation of an electrochemically inactive compound „A” leads to generation of reactive species “Ox” that can undergo electron transfer at the working electrode surface. If all participants of the electrode reaction are strongly adsorbed at the working electrode surface, we designate this mechanism as a SURFACECE Mechanism. The reaction scheme of this mechanism met in many physiological systems is as follows



SW voltammetric patterns are function of the electron transfer parameter related to the electrode reaction, but they also depend on the kinetic and thermodynamic parameters related to the chemical step. The entire interplay of all parameters leads to very specific voltammograms, whose features can reveal important kinetic and thermodynamic parameters relevant to the physiological systems of interest. Electrode transformation of many transient metal ions, drugs, lipophilic enzymes, fat-soluble vitamins and lipophilic metal-Ligand complexes follow this pathway. We provide the readers methods on how to access all thermodynamic and kinetic parameters relevant for this particular mechanism.

Es := 0.25 ΔE := 0.6 dE := 0.004 Esw := 0.05
n := 2 F := 96500 R := 8.314 T := 298.15

r := 1..1

f := 10 ε := kf + kb

f_r =
Pri Keq of 0.01 i kinet na elektrodna reakcija lambda od 1.778...da se napravi analiza pri cepenjetu od Kchem od 100 do 0.001--sliki

j := 1.. $\frac{\Delta E}{dE} \cdot 50$ α = 0.5

$$pot_j := Es + Esw - \left[\left(\text{ceil}\left(\frac{j}{25} \cdot \frac{1}{2}\right) \cdot dE + \text{if}\left(\frac{\text{ceil}\left(\frac{j}{25}\right)}{2} = \text{ceil}\left(\frac{j}{25} \cdot \frac{1}{2}\right), 1, -1\right) \cdot Esw + Esw \right) - dE \right]$$

ks_r := 10^{0.9r}
kf := 5 kb := 5
λ_r := $\frac{ks_r}{f}$ K_{chem} := 10⁻² K = 0.01

$\log\left(\frac{ks}{f_r}\right) =$

ε_{sw} := 10¹
z := $\frac{\epsilon}{f}$ λ_r = ε = 10

ε_r =

$\log\left(\frac{ks}{\epsilon_r} \cdot K\right) =$ z = 1

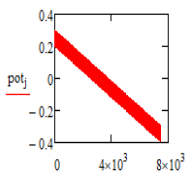
$\log(z_r) =$ λ_r =

k := 1.. $\frac{\Delta E}{dE} \cdot 50$

S_{kk} := e ^{$\frac{z}{50} \cdot (-k)$} - e ^{$\frac{z}{50} \cdot (-k+1)$}

ε = 10 z = 1

$\log(\lambda_1) = -0.1$



$\frac{0.099}{0.119} = 0.832$

$\frac{.239}{.252} = 0.948$

10^{1.2} = 15.849

Pri Konst na ramnoteza od Keq = 1, da se napravi analiza Kvizimax pri z-hem parametar od 0.2 do 1

Φ_{sw} := n · $\frac{F}{R \cdot T}$ · pot_j

ε_r =

$\log(\lambda_r) =$

log(z) = 0

THEORETICAL MODEL IN SQUARE-WAVE VOLTAMMETRY of a SURFACE CE (Chemical-Electrochemical) Mechanism

$$\Psi_{1,r} := \lambda_r \cdot e^{-\alpha \cdot \Phi_1} \cdot \frac{K}{1+K} \cdot \left[1 + \lambda_r \cdot e^{-\alpha \cdot \Phi_1} \cdot \frac{K}{(1+K) \cdot 50} - \frac{\lambda_r \cdot e^{-\alpha \cdot \Phi_1} \cdot S_1}{(K+1) \cdot z} \cdot (1) + \frac{\lambda_r \cdot e^{(1-\alpha) \cdot \Phi_1}}{50} \right]^{-1}$$

- kf and kb are rate constants of forward and backward chemical steps
- K is the equilibrium constant of chemical step
- ks is standard rate constant of electrode reaction
- λ is dimensionless kinetic parameter related to electron transfer step
- z is dimensionless chemical kinetic parameter
- S is integration factor
- F is Faraday constant
- R is universal gas constant
- Φ is dimensionless potential
- dE is potential step
- Es is starting potential
- f is SW frequency
- Ψ is dimensionless current
- α is electron transfer coefficient
- Esw is the square-wave amplitude

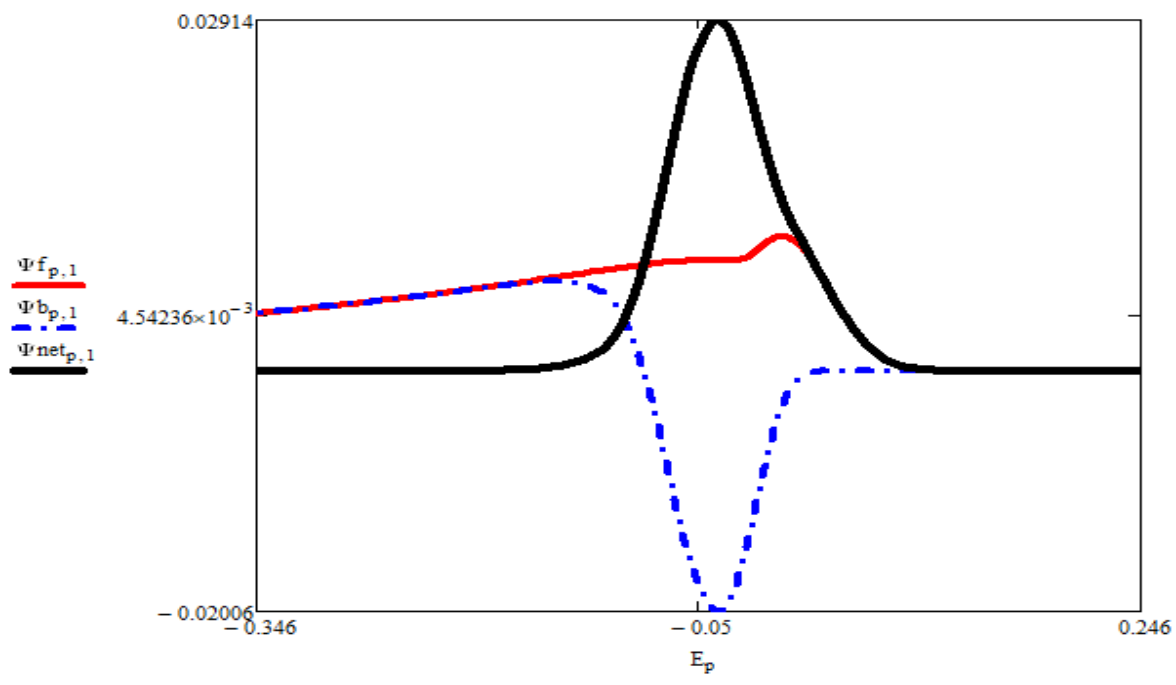
$$\Psi_{k,r} := \frac{\frac{\lambda_r \cdot e^{-\alpha \cdot \Phi_k} \cdot K}{1+K} \cdot \left(1 - \frac{1}{50} \cdot \sum_{j=1}^{k-1} \Psi_{j,r} \right) \cdot (-z)^{-1} \cdot \lambda_r \cdot \left(\frac{1}{1+K} \right) \cdot (-1) \cdot e^{-\alpha \cdot \Phi_k} \cdot \sum_{j=1}^{k-1} (\Psi_{j,r} \cdot S_{k-j+1}) - \frac{\lambda_r}{50} \cdot e^{\Phi_k(1-\alpha)} \cdot \sum_{j=1}^{k-1} \Psi_{j,r}}{\left(\frac{\lambda_r \cdot e^{-\alpha \cdot \Phi_k} \cdot K}{1+K} \cdot \frac{1}{50} \right) + 1 + (z)^{-1} \cdot \lambda_r \cdot (-1) \cdot \left(\frac{1}{1+K} \right) \cdot S_1 \cdot e^{-\alpha \cdot \Phi_k} + \frac{\lambda_r}{50} \cdot e^{\Phi_k(1-\alpha)}}$$

p := 1.. $\left(\frac{\Delta E}{dE}\right) - 1$

Ψ_f_{p,r} := Ψ_{(p+1)·50,r} Ψ_b_{p,r} := Ψ_{50·p+25} Ψ_{net}_{p,r} := Ψ_f_{p,r} - Ψ_b_{p,r}

E_p := Es - p · dE

$\exp\left(\frac{96480}{8.314 \cdot 30}\right) = 9.833 \times 10^{167}$



E_p =

0.246
0.242
0.238
0.234
0.23
0.226
0.222
0.218
0.214
0.21

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